Increased Oral Bioavailability of Piperine from an Optimized *Piper nigrum* Nanosuspension

Authors

Fatiqa Zafar¹, Nazish Jahan¹, Khalil-Ur-Rahman², Haq Nawaz Bhatti¹

Affiliations

- 1 Department of Chemistry, University of Agriculture, Faisalabad, Pakistan
- 2 Department of Biochemistry, University of Agriculture, Faisalabad, Pakistan

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Correspondence

Fatiqa Zafar, PhD Scholar

Natural Products Lab, Department of Chemistry, Faculty of Sciences, University of Agriculture

Jail Road, Faisalabad (38000), Punjab, Pakistan

Phone: +92 419 2001 61, Fax: +92 419 2007 64

fatiqazafar@gmail.com

Supporting information available online at <http://www.thieme-connect.de/products>

ABSTRACT

The aim of the present study was to enhance the pharmaceutical potential and oral bioavailability of piperine, which is the bioactive constituent of *Piper nigrum*, using the nanosuspension approach. Nanoprecipitation, which is a simple and reproducible process, was used for nanosuspension formulation. To prepare a pharmaceutical-grade nanosuspension with the required particle size, important formulation parameters (amount of plant extract, concentration of stabilizer, and anti-solvent-to-solvent ratio) were optimized using the central composite design of response surface methodology. The optimized nanosuspension was characterized using scanning electron microscopy, atomic force microscopy, Fourier transform infrared spectroscopy, and *in vitro* dissolution testing as well as by measuring the zeta potential. *In vivo* pharmacokinetic studies were conducted to determine the bioavailability of the prepared nanosuspension. Results of the optimization study indicated that 0.13% plant extract, 0.25% stabilizer, and an antisolvent-to-solvent ratio of 10.0 were the best parameters to obtain a homogeneous nanosuspension with the required particle size. The optimized nanosuspension demonstrated a mean particle size, polydispersity index, and zeta potential of 172.5 nm, 0.241, and –16.6 mV, respectively. The results of the characterization studies illustrated that the nanosuspension was in the nanometer size range and had good surface morphology. The optimized nanosuspension showed a better dissolution rate and a 3.65-fold higher oral bioavailability for the *P. nigrum* nanosuspension than its coarse suspension. The present outcomes clearly demonstrated that to obtain an effective therapeutic potential, nanof ormulation of medicinal plants is a better alternative than conventional dosage forms.

Introduction

Piper nigrum L. (Piperaceae), which is generally known as black pepper, is extensively used in household spices and has paramount importance in the field of alternative medicine [1]. Piperine is the major alkaloid of *P. nigrum* and possesses diverse pharmacological activities [2] such as antihypertensive [3], antioxidant [4], hypolipidemic [5], antiasthmatic [6], hepatoprotective [7], and antimicrobial [8] properties. It is not only a bioavailability enhancer but also a potent reactive oxygen species quencher and it

can inhibit lipid peroxidation [9–11]. Being a bioavailability enhancer, piperine serves as a drug receptor and potentiates the drug via conformational interactions by making target cells more receptive [2].

Although piperine acts as a bioavailability enhancer, its water solubility is negligible [12]. Its poor water solubility results in its limited bioavailability in biological systems and necessitates the use of a higher dose of drugs to obtain the anticipated pharmacological response. Therefore, it is imperative to improve the solubility of piperine to obtain its maximum therapeutic efficacy in the

ABBREVIATIONS

3D	three-dimensional
AFM	atomic force microscopy
ANOVA	analysis of variance
AS/S ratio	antisolvent-to-solvent ratio
AUC	area under the curve
CCD	central composite design
CV	coefficient of variation
DLS	dynamic light scattering
FTIR	Fourier transform infrared spectroscopy
HPMC	hydroxypropyl methylcellulose
PDI	polydispersity index
RSM	response surface methodology
SEM	scanning electron microscopy

biological system. Among the various approaches, nanosuspension formulation is the method of choice that enhances the delivery of sparingly water-soluble drugs [13, 14].

Nanoformulated herbal drugs possess efficient biopharmaceutical properties and desirable target characteristics [15]. These nano-sized phytotherapeutic agents offer better pharmaceutical benefits over traditional herbal preparations, including enhanced solubility, improved bioavailability, reduced medicinal doses, and better residence time of drugs in biological systems [16]. The enhanced bioavailability due to the smaller size and greater surface area [17] consequently reduces the treatment dose [18]. Nanosuspensions can also improve the pharmacokinetics of pharmaceuticals [19].

To obtain an effective and pharmaceutically stable nanosuspension with the required particle size and desired morphology, it is imperative to optimize formulation parameters. The optimization of each parameter is time-consuming and expensive and does not determine the effect of individual parameters on various responses [20]. Moreover, large numbers of trials are required in factorial design, which eventually does not provide the required information (interactive effect), making the situation more complex [21]. To overcome these problems, optimization studies can be effectively conducted using RSM [22].

The present research was conducted to enhance the oral bioavailability of *P. nigrum* plant extract by formulating its nanosuspension. Different process parameters were optimized using RSM to obtain a homogenous nanosuspension with a minimum particle size. The optimized nanosuspension was characterized by spectroscopic techniques and evaluated by *in vitro* dissolution testing and *in vivo* pharmacokinetic studies.

Results and Discussion

Nanosuspensions were prepared by the antisolvent precipitation method using an HPMC stabilizer. The formulation parameters considered were the amount of plant extract (A), concentration of stabilizer (B), and AS/S ratio (C). These parameters were optimized using the CCD of RSM to obtain a homogenous nanosuspension with a minimum particle size (z-average; nm) and stan-

dard PDI value. The statistical experimental design suggested that the quadratic model was the most suitable model, with the smallest p values and largest Fisher values (F values), to explain the relationship between independent variables (amount of plant extract, concentration of stabilizer, and AS/S ratio) and response variables [particle size (R1) and PDI (R2)]. After selecting the model, regression equations (Equations 1 and 2) for the response variables were established. The positive sign before the coefficients suggests a synergistic effect of independent variables on particle size and PDI reduction, whereas the negative sign represents an antagonistic effect.

P. nigrum (Size; nm)

$$(R1) = + 365.42 + 37.40A - 1.97B + 31.97C - 74.77AB - 99.50AC + 9.05BC + 0.92A^2 + 66.74B^2 + 1.70C^2$$

(Equation 1)

P. nigrum (PDI)

$$(R2) = + 0.42 - 0.012A - 0.047B + 0.060C - 0.16AB + 0.049AC - 0.071BC - 4.633E-003A^2 + 0.15B^2 - 0.080C^2$$

(Equation 2)

ANOVA was used to evaluate the linear, quadratic, and interactive effects of independent variables on R1 and R2. The probability ($p = 0.008$ and $p < 0.0001$) and F (9.42 and 48.82) values for R1 and R2, respectively, reflected the significance of the quadratic model for the optimal production of nanosuspensions with the desired properties (► **Tables 1 and 2**). P values were used as a tool to check the significance of the model terms. A p value of < 0.0500 indicated that the model terms were significant. The smaller the p value, the more significant the corresponding coefficient was. The variables A and C and the interactions AB, AC, and B^2 had a significant effect on the nanosizing of *P. nigrum*, and the parameters B, C, AB, AC, BC, B^2 , and C^2 had a significant effect on reducing PDI; the remaining parameters showed no effect or an inverse relationship on particle size and PDI reduction (► **Tables 1 and 2**).

The nonsignificant lack of fit F value of R1 (0.47) and R2 (1.47) demonstrated good predictability of the model. The quality of fit for the quadratic model was further evaluated using the coefficient of determination (R^2). R^2 for particle size (0.8945) and PDI (0.9777) indicated 89.45 and 97.77% variability in both responses. The model was stronger and predicted a better response as R^2 was closer to 1.000. In previous studies, a regression model with $R^2 > 0.9000$ has been considered to have a very high correlation [23]. The CV was calculated to be 12.41 and 9.26% for R1 and R2, respectively, and was found to be satisfactory (► **Tables 1 and 2**).

The influence of all independent variables on the various responses of the nanosuspension formulation of *P. nigrum* was evaluated using 3D response surface plots. In each plot, the combined effect of two variables was simultaneously examined, whereas a third factor was kept at its central point. Response surface plots showing the interactive effect of all independent variables on the particle size and PDI of *P. nigrum* nanosuspensions are presented in ► **Figs. 1 and 2**. These plots show that all three formulation parameters have a significant impact on particle size and PDI reduction of *P. nigrum* nanosuspensions. However, the impact of the

► **Table 1** ANOVA for response surface quadratic model for the particle size of *P. nigrum* nanosuspensions.

Source of variance	Sum of square	df	Mean square	F value	P value Prob > F	Remarks
Model	2.23E+05	9	24 740.20	9.42	0.0008	Significant
A-Plant extract	19 100.72	1	19 100.72	7.28	0.0224	
B-Stabilizer concentration	53.04	1	53.04	0.02	0.8898	
C-Antisolvent/solvent ratio	13 956.52	1	13 956.52	5.32	0.0438	
AB	44 727.41	1	44 727.41	17.04	0.0021	
AC	79 209.96	1	79 209.96	30.17	0.0003	
BC	654.86	1	654.86	0.25	0.6283	
A ²	12.24	1	12.24	4.66E-03	0.9469	
B ²	64 199.45	1	64 199.45	24.45	0.0006	
C ²	41.53	1	41.53	0.016	0.9024	
Residual	26 252.41	10	2625.24			
Lack of fit	8403.54	5	1680.71	0.47	0.7860	Nonsignificant
Pure error	17 848.87	5	3569.77			
Cor total	2.49E+05	19				
R ²	0.8945			Adj R ²	0.7996	
Pred R ²	0.6395			CV (%)	12.41	
Adeq precision	11.685					

R² = Coefficient of determination, Pred R² = predicted R², Adj R² = adjusted R², Adeq precision = adequate precision, CV = coefficient of variation

► **Table 2** ANOVA for response surface quadratic model for PDI of *P. nigrum* nanosuspensions.

Source of variance	Sum of square	df	Mean square	F value	P value Prob > F	Remarks
Model	0.82	9	0.091	48.82	<0.0001	Significant
A-Plant extract	1.93E-03	1	1.93E-03	1.04	0.3323	
B-Stabilizer concentration	0.03	1	0.030	16.34	0.0024	
C-Antisolvent/solvent ratio	0.05	1	0.050	26.80	0.0004	
AB	0.21	1	0.210	115.17	<0.0001	
AC	0.019	1	0.019	10.28	0.0094	
BC	0.04	1	0.040	21.76	0.0009	
A ²	3.09E-04	1	3.09E-04	0.17	0.692	
B ²	0.33	1	0.330	176.53	<0.0001	
C ²	0.092	1	0.092	49.52	<0.0001	
Residual	0.019	10	1.86E-03			
Lack of fit	0.011	5	2.21E-03	1.47	0.3416	Nonsignificant
Pure error	7.53E-03	5	1.51E-03			
Cor total	0.84	19				
R ²	0.9777			Adj R ²	0.9577	
Pred R ²	0.8762			CV (%)	9.26	
Adeq precision	27.348					

R² = Coefficient of determination, Pred R² = Predicted R², Adj R² = Adjusted R², Adeq precision = Adequate precision, CV = Coefficient of variation

amount of plant extract is more prominent. Minimum particle size and PDI are observed when the amount of plant extract is less. The classical crystallization theory explains the impact of drug

concentration (plant extract in the present case) on particle size. According to this theory, during nanoformulation, the precipitation of nanoparticles involves a series of steps including nuclea-

► **Fig. 1** 3D response surface graphs for the particle size of *P. nigrum* nanosuspensions. (A) Amount of plant and concentration of stabilizer, (B) concentration of stabilizer and AS/S ratio, and (C) amount of plant and AS/S ratio.

► **Fig. 2** 3D response surface graphs for PDI of *P. nigrum* nanosuspensions. (A) Amount of plant and concentration of stabilizer, (B) concentration of stabilizer and AS/S ratio, and (C) amount of plant and AS/S ratio.

tion, molecular growth, and growth by coagulation and condensation, followed by agglomeration. Furthermore, the rate of each step governs the final particle size and particle size distribution. Supersaturation is the crucial driving force of this process and determines not only the nucleation rate but also the diffusion-controlled growth rate. The nucleation and growth of particles occur simultaneously, and both compete for supersaturation [24]. At a higher drug concentration, due to higher supersaturation, the rapid diffusion-controlled growth and agglomeration rates were more dominant than the nucleation rate that gave rise to larger particles [25, 26].

A gradual increase in particle size was also observed by increasing the concentration of HPMC. This may be due to the fact that an excessive amount of polymer increases particle size by increasing the size of the outer polymer surface and inhibits diffusion between the solvent and antisolvent during precipitation. Moreover, an increase in osmotic pressure by increasing polymer concentration results in an enhanced attraction among colloidal particles, thus leading to a larger particle size [27]. Furthermore, an increase in particle size and PDI by increasing the concentration of the stabilizer may result from Ostwald ripening [28]. In another

study, the particle size of resveratrol nanosuspension was increased by increasing the amount of the drug and concentration of the stabilizer [26].

Desirability and overlay plots created using Design Expert Software (version 7.1, Stat-Ease, Inc.) to attain the optimum values of independent variables (A, B, and C) for the formulation of *P. nigrum* nanosuspension with a minimum particle size and PDI are shown in ► **Fig. 3**. These plots further confirm the validity of the selected quadratic model of CCD of RSM in optimization studies. Particle size (172.5 nm), PDI (0.241), and zeta potential (−16.6 mV) confirmed the stability of the formulated nanosuspension (► **Fig. 4A, B**).

AFM of *P. nigrum* nanosuspension was conducted for illustrating the particle distribution. Thus, the overall particle distribution was observed to be uniform; however, the nanoparticles were observed to be submerged in the polymer layer at some places. The observed mean particle height of the *P. nigrum* nanosuspension was 55.78 nm (► **Fig. 5**). The particle size observed by AFM was smaller (50.78 nm) than that (172.9 nm) obtained by DLS. Fritzen-Garcia et al. [29] stated that the particle size of a nanosuspension measured by AFM was smaller than that measured by

► **Fig. 3** (A) Desirability and (B) overlay plot for particle size and PDI values of *P. nigrum* nanosuspensions.

► **Fig. 4** (A) Zeta size, PDI, and (B) zeta potential value of optimized *P. nigrum* nanosuspension.

DLS. This is due to the presence of a solvent in measurements taken by DLS, which causes nanoparticle swelling [30].

SEM of the *P. nigrum* nanosuspension at two different resolutions demonstrated an average particle size of $< 1 \mu\text{m}$ (► **Fig. 6**), which indicated no aggregation of particles. HPMC provided better stability to the nanosuspension, even after lyophilization. The particles were discrete and uniform and had a spherical- and rod-like shape with a smooth surface, which is a quality characteristic of HPMC [31,32].

Compared to the coarse plant extract, FTIR analysis of *P. nigrum* revealed a slight shift or disappearance of some peaks in the spectrum of the nanosuspension (**Fig. 1S**, Supporting information). However, the peak characteristics of the functional group region ($3500\text{--}1900 \text{ cm}^{-1}$) showed smaller variations than peaks in the fingerprint region ($1600\text{--}450 \text{ cm}^{-1}$). In the nanosuspension, peaks in the region of $3200\text{--}3550 \text{ cm}^{-1}$, which were characteristics of an --OH bond, became more intense with a slight decrease

in wavenumber. This may be due to H-bonding [33]. The other peaks varied to a very small extent from 2936.77 cm^{-1} and 2855.42 cm^{-1} in the plant extract to 2917.63 cm^{-1} and 2850.21 cm^{-1} in the nanosuspension. These peaks (2936.77 cm^{-1} and 2855.42 cm^{-1}) illustrated the presence of symmetric and asymmetric --CH and --CH_2 stretching. The peak at 1632.15 cm^{-1} represented --CO--N stretching, and a sharp peak at 927.91 cm^{-1} was characteristic of --CO stretching. A more prominent peak at 1444.42 cm^{-1} was representative of --CH_2 bending. The peaks at wavenumbers 803.80 , 846.49 , and 821.20 cm^{-1} were for out of the plane --CH bending and two adjacent substituted hydrogen atoms of 1,2,4-trisubstituted phenyl, respectively. All these peaks confirmed the presence of piperine as the major constituent in the coarse extract of *P. nigrum*. The fingerprint region ($1600\text{--}450 \text{ cm}^{-1}$) of the spectrum of the nanosuspension showed greater resemblance with the spectrum of the stabilizer than with the spectrum of the coarse plant extract. This may be due to some physical in-

teractions between the plant extract, stabilizer, and antisolvent [34] during nanoformulation. The present outcomes, which were similar to those obtained in a previous study [35], indicated that the plant extract in pure form or in nanosuspension form has the same structural features in terms of functional groups.

Results of the dissolution profile of the *P. nigrum* coarse extract and nanosuspension are presented in ► Fig. 7. A greater concentration of piperine in the dissolution medium was observed for the nanosuspension (73.66%) after 120 min than in the coarse extract (14.37%). This indicated a 3.65-fold increase in the dissolution behavior of the nanosuspension. The dramatically enhanced dissolution rate of the nanosuspension may result from the increased effective surface area [36] and decreased particle size [31] in accordance with Noyes Whitney's equation [37]. Comparable results were found by Kakran et al. [38], who fabricated nanoparticles of silymarin, hesperetin, and glibenclamide by the evaporative precipitation of the nanosuspension and concluded that the dissolution rate increased by reducing the particle size and increasing the surface area available for dissolution.

In *in vivo* trials, the concentration-time graph in pharmacokinetic studies showed a higher concentration of piperine in the plasma samples of rats treated with the *P. nigrum* nanosuspension than in those of rats treated with the coarse suspension at all studied time intervals (► Fig. 8). The maximum concentration (C_{max}) of piperine was achieved after 1 h (T_{max}) of oral administration of the *P. nigrum* nanosuspension; this maximum concentration was 1.73-fold higher than that of the coarse suspension. After T_{max} , the concentration of piperine started decreasing, indicating drug clearance from the biological system. However, the drug clearance rate was slower for the nanosuspension than for the coarse suspension, demonstrating its sustained release and greater residence time in the biological system. Tian et al. [39] found a prolonged residence time of the nanosuspension. The greater mean AUC (AUC_{0-24h}) for the nanosuspension than for the coarse suspension (► Table 3) represents a 2.7-fold increase in the bioavailability of the *P. nigrum* nanosuspension compared to its coarse suspension. Significant improvement in the C_{max} and AUC of the nanosuspension indicated better *in vivo* exposure of the nanosuspension through particle size reduction [40], which can be explained by the improved saturation solubility of the nanoparticles, as they are absorbed without the initial time-consuming step [36]. Furthermore, nanosuspension preparation for oral administration results in effective therapeutic concentrations in the blood because solubility and absorption problems in the gastrointestinal tract can be managed by extensively reducing the particle size of the nanosuspension [41, 42].

To the best of our knowledge, the formulation and pharmacokinetic evaluation of the nanosuspension of the *P. nigrum* plant extract have been conducted for the first time. In summary, nanosuspensions possess smaller particle sizes, higher surface areas, faster dissolution rates, less drug doses, lesser side effects, and enhanced bioavailability [43]. The results showed a distinctive comparison of the pharmacokinetics and dissolution properties between the nanosuspension and the coarse suspension of *P. nigrum*.

► Fig. 5 AFM image of *P. nigrum* nanosuspension.

► Fig. 6 SEM images of *P. nigrum* nanosuspension at (A) 1000 × magnification (10 μm bar size) (B) 10000 × magnification (1 μm bar size).

Materials and Methods

Plant collection and extract preparation

P. nigrum L. (fruit) was purchased from the local market of Faisalabad and identified by a plant taxonomist (Dr. Mansoor Hameed) at the Department of Botany, University of Agriculture, Faisalabad (UAF). A voucher specimen (228–2–2016) was deposited at the herbarium of the Department of Botany, UAF. The plant material was grounded to a fine powder after washing and drying. Fat/oil contents were removed with *n*-hexane (1 : 10 w/v) using a Soxhlet extractor. To obtain crude piperine, defatted plant material (30 g) was extracted with ethanol (300 mL) for approximately 6–8 h and the filtered extract was concentrated in a rotary evaporator (Buchi) and stored in a refrigerator for further use.

Formulation and optimization of nanosuspension

Nanoprecipitation (bottom-up approach) was used for the formulation of the nanosuspensions [44], with some modifications. The plant extract was completely dissolved in ethanol, and the organic phase was slowly injected (1 mL/min) with a syringe connected to a thin Teflon tube into an aqueous phase containing the stabilizer

► **Fig. 7** *In vitro* dissolution profile of *P. nigrum* nanosuspension and coarse extract. Results are expressed as the mean $\pm$ SD ($n = 3$). C. Ext = coarse extract, Nano = nanosuspension.

(HPMC) with continuous stirring at 6000 rpm for 6 h at room temperature. The requisite formulation parameters, i.e., amount of plant extract, concentration of stabilizer, and AS/S ratio, were optimized using the CCD of RSM. A standard stratagem of preliminary trials was used to determine the best possible conditions for the formulation of nanosuspensions. The CCD of RSM propounded 20 different conditions to perform the experiment for the optimal production of nanosuspensions by varying the amount of plant extract from 0.13 to 0.5%, the concentration of the stabilizer from 0.25 to 2%, and the AS/S ratio from 10 to 20 (Table 1S, Supporting information). The average particle size (z-average; nm) and PDI of the formulated nanosuspensions were selected as the response variables. The optimized nanosuspension (nanosuspension with a minimum particle size and appropriate PDI) was lyophilized at -60°C for 72 h. The freeze-dried sample was ground to a fine powder and used for solid-state characterization.

Characterization of the nanosuspension

The mean particle size (z-average; nm), PDI, and zeta potential of the nanosuspensions were measured by DLS using Malvern Zetasizer (Nano ZS). AFM (Shimadzu) was used for 3D characterization of the optimized nanosuspension. Surface morphology was evaluated by SEM (JEOL). Drug excipient interactions were evaluated by means of FTIR spectroscopy (Perkin Elmer). FTIR spectra of the crude herbal extract, optimized nanosuspension, and stabilizer (HPMC) were recorded.

In vitro dissolution testing

In vitro dissolution testing was conducted by adopting a modified version of the method used by Gera et al. [37]. USP dissolution apparatus type II (pharma test de) was used for dissolution testing of the coarse herbal extract and nanosuspension. For this purpose, an encapsulated 500-mg sample (coarse plant extract and lyophilized nanosuspension) was placed in 900 mL of the dissolution medium (0.1 M phosphate buffer at pH 7.4) at a temperature of $37 \pm 0.5^{\circ}\text{C}$ with a stirring rate of 50 rpm. Aliquots (5 mL) were withdrawn from the dissolution medium at predetermined time

► **Fig. 8** Concentration of piperine ($\mu\text{g/mL}$) in plasma samples of rats after oral administration of *P. nigrum* coarse suspension and nanosuspension. Results are expressed as the mean $\pm$ SD ($n = 3$). C. sus = coarse suspension, Nano = nanosuspension.

► **Table 3** Pharmacokinetic parameters after oral administration of *P. nigrum* nanosuspension and coarse suspension to experimental rats.

Parameters	Nanosuspension	Coarse suspension
C_{max} ($\mu\text{g/mL}$)	2.86 ± 0.24	1.65 ± 0.37
T_{max} (h)	1.0 ± 0.00	1.0 ± 0.00
$\text{AUC}_{0-24\text{ h}}$ ($\mu\text{g h/mL}$)	8.85 ± 1.21	3.28 ± 0.86

Results are expressed as the mean $\pm$ SD ($n = 3$). C_{max} = maximum concentration, T_{max} = time to reach maximum concentration, AUC = area under the curve

intervals (0, 15, 30, 45, 60, 75, 90, and 120 min), and the same volume of the prewarmed (37°C) dissolution medium was immediately added to the dissolution vessel to maintain sink conditions. The concentration of the dissolved drugs (piperine equivalent) was spectrophotometrically determined at a wavelength of 342 nm (λ_{max} of piperine). The concentration of piperine was evaluated from the regression equation generated from the calibration curve of standard piperine. The results are presented as drug dissolved (%) for the coarse plant extract and nanosuspension. All experiments were conducted in triplicate, and results are presented as the mean $\pm$ SD ($n = 3$).

In vivo pharmacokinetic study

According to the International Ethical Guidelines and under the supervision of veterinary doctors of the Clinical Medicine and Surgery Department, UAF, the animal model was designed to conduct *in vivo* pharmacokinetic studies. The proposal was approved by the Synopsis Scrutiny Committee (No. Chem-354, dated 15-02-2016) and endorsed by the Graduate Study Research Board through letter no. GDS/15501-4 dated 09-03-2016. Prof. Dr. Ghulam Muhammad oversaw the use of rats in this research; these rats were utilized as per the principles of the 3R's. For pharmacokinetic studies, male Wistar albino rats ($250 \pm 20\text{ g}$) were kept in an animal house for 1 week to acclimatize and were fed

with a normal rat diet (rat chaw and grains). Two groups of three rats were made. Before the experiment, the rats were fasted overnight with free access to water. For comparison, the rats in the first group were orally administered the *P. nigrum* nanosuspension (50 mg/kg body weight), whereas those in the second group were orally administered the coarse suspension of *P. nigrum* at the same dose.

Blood samples (0.5 mL) were withdrawn by cardiac puncture in sodium heparinized tubes at predetermined time intervals of 0.5, 1, 2, 4, 6, 12, and 24 h. Plasma was separated after centrifugation at $170 \times g$ for 20 min and stored at -20°C for further analysis. Piperine was extracted from the plasma samples by adopting the method used by Roy et al. [45] and quantified ($\mu\text{g/mL}$) by HPLC (model LC-10A; Shimadzu) using the standard calibration curve of piperine. A Supelco analytical HS (C-18) column having a length of 15 cm, diameter of 4.6 mm, and thickness of $5\ \mu\text{m}$ was used. The mobile phase (methanol:water, 70:30) flow rate was adjusted to 1 mL/min. The temperature of the column oven was set to 30°C , and the pressure of the delivery pump was adjusted to 4413 kilopascal. A UV-visible detector (model SPD-10A; Shimadzu) was set at a wavelength of 340 nm. Acquisition software (class LC-10; Shimadzu) was used for analyzing the chromatograms.

Determination of pharmacokinetic parameters

Pharmacokinetic parameters such as peak plasma concentration (C_{max}) and time required to achieve peak plasma concentration (T_{max}) were determined directly from the concentration-time graph. The AUC ($\text{AUC}_{0-24\text{h}}$) was determined by the trapezoidal method [37] using Microsoft Excel, 2007. Results are presented as the mean $\pm$ SD ($n = 3$).

Statistical analysis

The CCD of RSM was used for optimizing the formulation parameters. The significance of the studied independent variables and their interactions were tested by ANOVA. Dissolution rates and values of various pharmacokinetic parameters are expressed as the mean $\pm$ SD ($n = 3$).

Supporting information

FTIR spectra of the *P. nigrum* coarse plant extract, *P. nigrum* nanosuspension, and stabilizer (HPMC), HPLC chromatograms of the *P. nigrum* coarse plant extract, *P. nigrum* nanosuspension (at C_{max}) and piperine standard, and experimental conditions used in the optimization study are provided as Supporting Information.

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Conflict of Interest

The authors declare no conflict of interest

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