SYNLETT Spotlight 110

p-Toluenesulfonylmethyl Isocyanide (TosMIC)

Compiled by V. V. Ramana Reddy

This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

V. V. Ramana Reddy was born in Tenali, Andhra Pradesh, India. He obtained his MSc in 1999 from Pondicherry University, Pondicherry, India. He is currently working as a senior research fellow towards his PhD under the supervision of Dr. P. Radha Krishna at the Indian Institute of Chemical Technology, in Hyderabad, India. His research interests include the total synthesis of bioactive natural products and *C*-nucleosides.

D-206, Discovery Laboratory, Organic Chemistry Division-III,
Indian Institute of Chemical Technology, Tarnaka,
Hyderabad-500 007, India
E-mail: ramanareddyvv@yahoo.co.in

Introduction

p-Toluenesulfonylmethyl isocyanide (TosMIC) is one of the most versatile and widely applicable reagents in organic synthesis. The methylene group of TosMIC is highly activated ($pK_a = 14$) by the two electron-withdrawing substituents. Deprotonation of TosMIC has been achieved with an array of bases ranging from K_2CO_3 in MeOH to *n*-BuLi in THF. TosMIC is considered as a formaldehyde equivalent with reversible polarity. The reagent is a stable solid (mp 116–117 °C) that is commercially available, or

can be prepared from *p*-toluenesulfonic acid¹ in a two-step process. Many heterocycles, such as oxazole, pyrrole, imidazole, thiazole and 1,3,4-triazole, can be synthesized from TosMIC.

Figure 1

Abstracts

(A) Most ketones are converted in one operation into cyanides with TosMIC in the presence of potassium *tert*-butoxide in non-protic solvents (DME, DMSO).² The reductive cyanation of some aldehyde³ was carried out at low temperature and needs addition of methanol.

(B) TosMIC on reaction with aldehyde in methanol at room temperature leads to oxazolene, where as oxazoles were formed at reflux temperature.⁴ The addition of TosMIC to the aldehyde is followed by cyclization and subsequent elimination of the tosyl group to afford oxazole. Dhar et al. reported a modified oxazole synthesis using DBU in DME at 80 °C.^{4b}

(C) Base-induced addition of TosMIC to aldimines in protic medium occurs with concomitant cyclization followed by elimination of *p*-toluenesulfonic acid to result in imidazoles.⁵

(D) Pyrroles⁶ are obtained by base-induced addition of TosMIC to Michael acceptors. The ring closure between the isocyano and enolate carbons is followed by aromatization.

(E) TosMIC, on reaction with diazonium salts, results in 1,2,4-triazoles.⁷ The TosMIC anion attacks the electrophilic β -nitrogen of the diazonium ion, then ring closure occurs.

(F) Mono- and dialkylated TosMIC⁸ were formed from corresponding alkyl halides under phase transfer conditions. Hydrolysis of dialkylated TosMIC leads to symmetrical and unsymmetrical ketones.⁹ The reduction of mono and dialkylated TosMIC with Li in liquid NH₃ afforded the corresponding hydrocarbons.¹⁰

(G) Reaction of TosMIC with carbondisulfides under phase transfer conditions provides the tetrabutylammonium salt of thiazole, which can be converted to thiazole.¹¹

(H) Recently, the synthesis of C-nucleosides by the TosMIC approach from sugar-derived aldehydes and other Michael acceptors was reported.¹²

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