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Spotlight

Ethyl Diazoacetate

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Ethyl diazoacetate (EDA; $C_4H_6N_2O_2$) is a yellow liquid with a pungent odor used mainly in organic synthesis for cyclopropanation of unsaturated compounds, but also for cyclopropanation, cycloaddition reactions, synthesis of triazoles and pyrazolines, and insertion reactions (Scheme 1).¹⁻³ The use of EDA is also studied in continuous-flow methods with homogeneous and heterogeneous catalysis with transition metals, for example, in selective olefination of aldehydes using copper(II) complexes.^{3,4} It can be prepared through reaction between sodium nitrite and glycine ethyl ester hydrochloride in the presence of diluted sulfuric acid, ethanol, and sodium acetate. Commercially, it is available in dichloromethane or toluene solutions since it is flammable, shock sensitive, and toxic.^{1,2} Nevertheless, EDA is an important synthon for organic synthesis (Table 1).



Gabriel A. S. Aquino is a pharmacist and, currently, a PhD student at the Federal University of Rio de Janeiro (UFRJ) under the supervision of Prof. Sabrina B. Ferreira and Prof. Floriano P. Silva Jr. His work involves the synthesis and evaluation of novel nitrogenated heterocycles to test against SARS-CoV-2.

Floriano P. Silva Jr. received his PhD from the Federal University of Rio de Janeiro in 2005. His postdoctoral experiences were held at Oswaldo Cruz Institute (IOC) in 2006 and at the University of Kansas in 2008. Currently, he is the head of the Laboratory of Experimental and Computational Biochemistry of Drugs in IOC and his research efforts focus on structure and chemistry of proteins, molecular modeling, drug design, and structural molecular biology.

Sabrina B. Ferreira received her PhD from the Federal University of Rio de Janeiro in 2008 under the supervision of Prof. Carlos R. Kaiser and Prof. Vitor F. Ferreira. After her postdoctoral experience held at Fluminense Federal University (UFF) with Prof. Vitor F. Ferreira, she became a professor at UFRJ in 2010 where she is the head of the Laboratory of Organic Synthesis and Biological Prospecting. Her research efforts focus on organic synthesis, acting on the following subjects: heterocycles, carbohydrates, and natural products, and search for biologically active compounds.

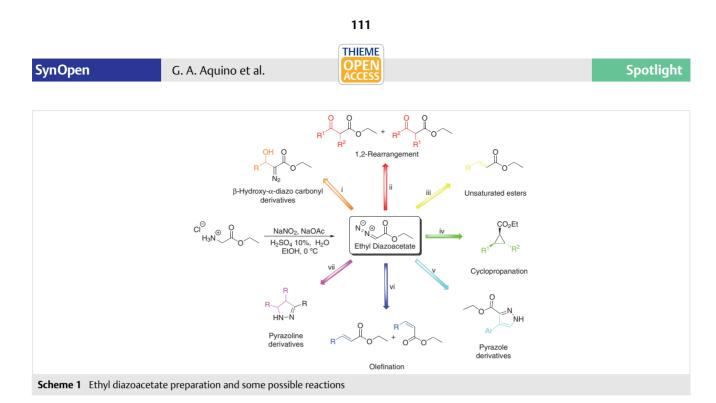
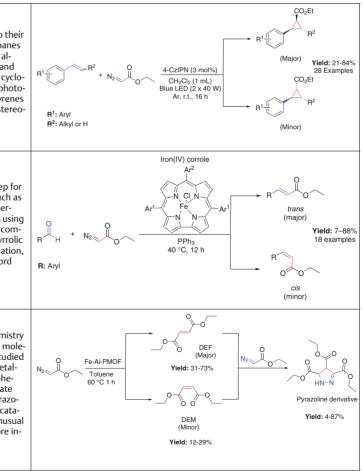


Table 1 Recent Applications of Ethyl Diazoacetate

(A) Cyclopropanes are a unique synthon within organic chemistry due to their properties such as high ring strain and geometry, which gives cyclopropanes an unusual structure with relatively shorter C–C and C–H bonds than in alkanes, and enhanced π -character of C–C bonds than normal.⁵ Langletz and collaborators have recently reported the stereoconvergent synthesis of cyclopropanes using EDA and different styrenes in the presence of different photo-catalysts under irradiation with blue LEDs. Using, *E*-, *Z*- or *E*-, and *Z*- β -styrenes mixture, they observed the formation of the same product, a single diastereomer with high yields.⁶

(B) Olefination, i. e., obtention of C=C double bonds, is an important step for organic synthesis and it is chiefly achieved by using ylide compounds such as phosphorus reagents in the Wittig reaction. Zou and co-workers have performed different reactions to first investigate an olefination of aldehydes using EDA in the presence of triphenylphosphine (PPh₃) with iron(IV)–corrole complexes as catalysts. These complexes are analogues of porphyrin tetrapyrrolic macrocycle used mainly as catalysts for oxidation, epoxidation, hydroxylation, and insertion reactions. Their results showed an efficient method to afford *trans*-olefination products.⁷

(C) Pyrazoline derivatives are important synthons used in medicinal chemistry due to their biological properties and also in the synthesis of fluorescent molecules due to their chromophoric properties. Abeykoon and co-workers studied the synthesis of pyrazolines from EDA using a metal porphyrin based metalorganic framework (Fe-Al-PMOF) and iron tetrakis(4-methoxycarbonylphenyl)porphyrin (FeP). Their results indicate the formation of diethyl maleate (DEM) and diethyl fumarate (DEF) with subsequent formation of the pyrazoline ring by the reaction of DEM and DEF with more EDA. The more the catalyst was washed, the more product was formed the next run, which is unusual for this type of reaction, and it can be explained by the activation of more inner catalytic sites in each wash.⁸

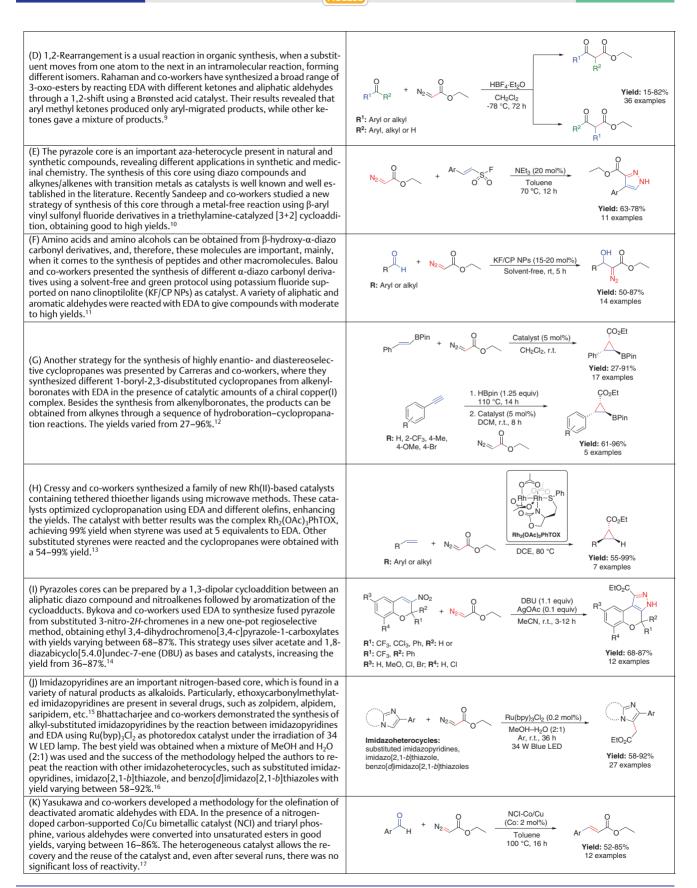


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Conflict of Interest

The authors declare no conflict of interest.

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