

SYNLETT Spotlight 457

6-Diphenylphosphinopyridin-2-(1H)-one (6-DPPon)

Compiled by Vahid Khakyzadeh



This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

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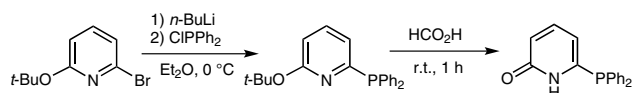
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Dedicated with best wishes to Prof. Dr. Bernhard Breit at Albert-Ludwigs-Universität Freiburg

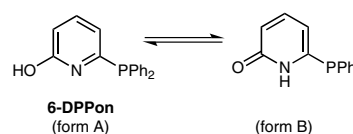
Introduction

Among heterocyclic structural units, pyridines are the most prevalent and have attracted the attention of chemists.¹ One important aspect of pyridine chemistry is designing new ligands based on the pyridine core.² Inspired by DNA base pairing, 6-DPPon (white solid, mp: 187 °C) was introduced by Bernhard Breit (Albert-Ludwigs-Universität Freiburg) as a monodentate ligand.³ This compound can not only be easily prepared (Scheme 1) but also has a brilliant property: the ability for self-assembly.⁴



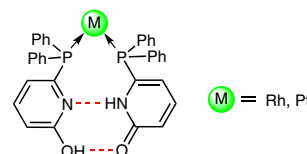
Scheme 1 Preparation of 6-DPPon

6-DPPon has two tautomeric forms, a 2-pyridone and a 2-hydroxypyridine tautomer (Scheme 2).



Scheme 2 Tautomeric forms of 6-DPPon

Interaction between form A and form B through hydrogen bonding can in situ generate a bidentate donor ligand in the coordination sphere of a metal (rhodium and platinum) center (Scheme 3). The present subject can open new gates to the design of self-assembled ligands and can be considered in related chemistries.

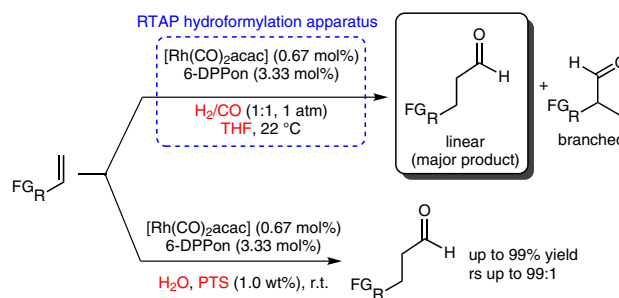


Scheme 3 Self-assembly of the 6-DPPon ligand

Abstracts

(A) Room Temperature Ambient Pressure (RTAP) Hydroformylation of Terminal Alkenes

Breit and co-workers have developed a hydroformylation of terminal alkenes under mild conditions: at room temperature and under ambient pressure. A bidentate donor ligand, generated in situ by self-assembly of 6-DPPon, reacted with rhodium and created a new catalyst with unique properties. Various ligands were tested in comparison to 6-DPPon and the best results (high yield and little isomerization) were obtained with 6-DPPon. High selectivity, low catalyst loading, a high level of generality, and excellent reactivity were some promising aspects of this protocol.⁵ It was found that terminal alkenes can be hydroformylated in aqueous media by slightly changing the reaction conditions. Breit and co-workers investigated various surfactants, and the results indicated that polyoxyethanyl- α -tocopheryl sebacate (PTS) was the best choice. Interestingly, 6-DPPon was the best ligand for this reaction. It is worth noting that in addition to the advantages described above, with the new protocol the structure of the self-assembly catalyst is stable in water as a protic solvent; an important point for self-assembled structures.⁶



SYNLETT 2014, 25, 0300–0301

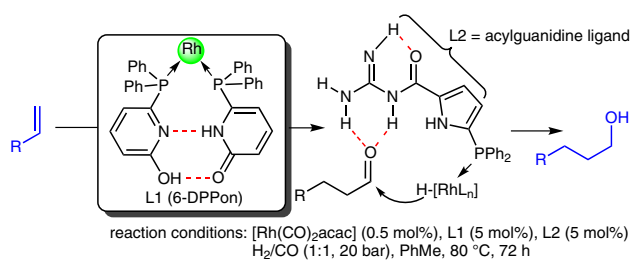
Advanced online publication: 02.12.2013

DOI: 10.1055/s-0033-1340359; Art ID: ST-2013-V0464-V

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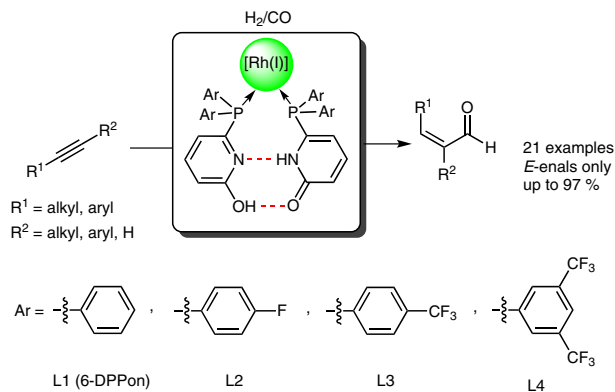
(B) Tandem Rhodium-Catalyzed Hydroformylation–Hydrogenation of Alkenes

In 2012, a unique tandem reaction was designed. In this reaction, a one-pot conversion of alkenes into linear alcohols is achieved using two different transformations (hydroformylation of alkenes and aldehyde hydrogenation). The first step (hydroformylation) was mediated by a rhodium complex which was generated by coordination of two 6-DPPon's, and a second step was carried out with an acylguanidine ligand. High regioselectivity and simultaneous a highly chemoselective reduction were two highlights of this work.⁷



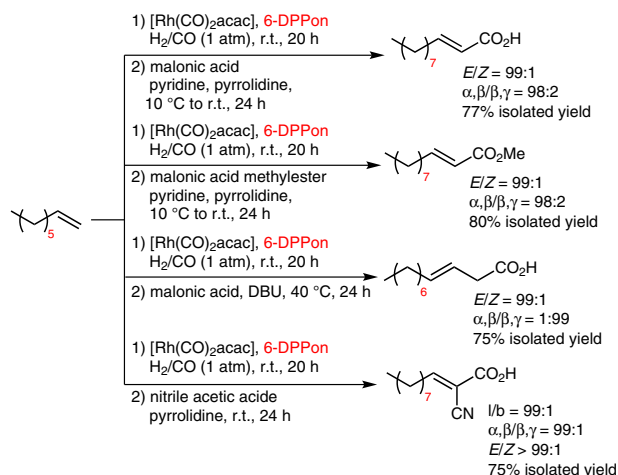
(C) Hydroformylation of Alkynes

Alkynes were hydroformylated stereo- and chemoselectively using 6-DPPon as a self-assembling ligand. In this study, several derivatives of the mentioned ligand were designed and investigated. This was the first time that dialkyl- as well as diaryl-substituted alkynes furnished *E*-enals with excellent chemo- and stereoselectivity.⁸



(D) One-Pot C3-Homologation of Terminal Alkenes

A new method to furnish carbonyl and carboxylic compounds was established by Breit and co-workers. In this method, by a combination of regioselective RTAP hydroformylation with 6-DPPon and a rhodium catalyst followed by decarboxylative Knoevenagel reaction (organocatalysis), various interesting compounds were produced.⁹ In all of the reactions, the presence of 6-DPPon was crucial.



References

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