Recent Progress in the Synthetic Assembly of 2-Cyclopentenones

David J. Aitken,*a Hendrik Eijsberg,^{a,b} Angelo Frongia,^b Jean Ollivier,^a Pier Paolo Piras^b

^a Laboratoire de Synthèse Organique & Méthodologie, ICMMO (CNRS UMR 8182), Université Paris Sud, 15 rue Georges Clemenceau, 91045 Orsay cedex, France Fax +33(1)69156278; E-mail: david.aitken@u-psud.fr

^b Dipartimento di Scienze Chimiche e Geologiche, Università degli studi di Cagliari, Complesso Universitario di Monserrato, S.S. 554, Bivio per Sestu, 09042 Monserrato, Cagliari, Italy

Received: 09.07.2013; Accepted after revision: 21.08.2013

Abstract: An overview of the most important synthetic strategies currently available for the preparation of cyclopent-2-enones is presented and illustrated with recent applications.

- 1 Introduction
- 2 Multicomponent Ring Assembly
- 3 Cyclizations
- 4 Transformations of Existing Cyclic Systems
- 5 Miscellaneous Methods
- 6 Conclusions

Key words: cyclopentenones, cyclization, carbocycles, ring closure, rearrangement, annulation

1 Introduction

Scope of this Review

2-Cyclopentenones are a frequently encountered class of cyclic enone. They feature in many areas of organic chemistry and serve as benchmark substrates for numerous chemical transformations, and natural product structures containing a 2-cyclopentenone molecular feature are ubiquitous.

Some of the synthetic routes to the title compounds have been reviewed periodically¹ (see also specific sections), but a global appraisal has not appeared for some time.² This review covers the literature over the last decade or so, and it endeavors to provide an overview of the most commonly used synthetic approaches for assembling the eponymous core feature from unrelated precursors. It is structured according to the strategy by which the cyclic enone feature is created, rather than according to any particular type of derivative or substitution pattern. New approaches as well as new results using established ones are considered equally.

The numerous methods available for the generation of 2cyclopentenones by α,β -elimination reactions of appropriately functionalized cyclopentanone precursors fall outside of the scope of this review, as do the vast array of oxidations of cyclopentenes, 2-cyclopentenols and cyclopentanones. These limitations notwithstanding, there are a

SYNTHESIS 2014, 46, 0001–0024 Advanced online publication: 21.11.2013 DOI: 10.1055/s-0033-1340414; Art ID: SS-2013-E0467-R © Georg Thieme Verlag Stuttgart · New York considerable number of ways in which the target ring system can be created from acyclic precursors, in either intermolecular or intramolecular mode. Most of the possible disconnection strategies have been examined, and it is important to recognize that for any given target 2-cyclopentenone, there may be several convenient approaches available. The main approaches for ring construction are summarized graphically in Figure 1.



Figure 1 The main ring-construction strategies for 2-cyclopentenone synthesis, showing the atom connectivities made during ring assembly (left and center) and cyclization approaches (right). These and other strategies are covered in this review.

Useful procedures based on the transformation of existing cyclic structures also exist, as do some miscellaneous methods; these will be treated towards the end of this review.

The 'ideal' choice of synthetic route depends both on the specific features in the target skeleton, such as the presence of sensitive functional groups or stereogenic centers, and on contextual constraints, such as the employment (or the preclusion) of metals, heat, particular solvents, and so on. Control of the relative configuration of 4,5-disubstituted 2-cyclopentenones can be achieved either by using a precursor in which those chiral centers are already established, or by using an approach in which these centers are created in a diastereoselective fashion, such as the Nazarov or related cyclizations. The preparation of nonracemic 4- and/or 5-substituted 2-cyclopentenones frequently relies on the use of nonracemic chiral substrates. The use of chiral auxiliaries during the ring-creation process has allowed some asymmetric syntheses to be performed, but catalytic enantioselective syntheses are rare at present; indeed this might constitute an area of particular attention for future developments.

Applications of the synthetic approaches reviewed herein have been selected for illustrative purposes as best as possible, but the quantity of work conducted in the area and the structural diversity of the molecular targets make it unfeasible to present an exhaustive list of synthetic applications here.

Biographical Sketches











David J. Aitken was born in 1963 and studied chemistry at the University of Strathclyde (Glasgow), obtaining his PhD in 1986, under the supervision of Prof. H. C. S. Wood and Prof. C. J. Suckling. After a twoyear post-doctoral appointment with Prof. H.-P. Husson at the ICSN (Gif-

Hendrik Eijsberg studied at the Ecole Nationale Superieure de Chimie de Paris (France) and carried out undergraduate work on the asymmetric conjugate addition of organoboron compounds catalyzed by rhodium–diene complexes in the group of Prof. J.-P. Genet and Dr. S. Darses. In

Angelo Frongia was born in Cagliari (Italy) in 1973. He graduated and received his PhD degree in Organic Chemistry from the University of Cagliari under the supervision of Prof. P. P. Piras. Following collabora-

Jean Ollivier obtained his PhD degree in 1982 at the University Paris-Sud (Orsay) under the direction of Dr. J. Salaün, then joined the Centre National de la Recherche Scientifique (CNRS). In 1986 he obtained a Doctorat-ès-Sci-

Pier Paolo Piras received his Laurea in Chemistry at the University of Cagliari (Italy) in 1971 and began his academic career at the same university as Assistant Professor in 1972. He was a postdoctoral fellow (1980– 1981) with Professor C. J. M. Stirling at the University of Bangor (North Wales). In sur-Yvette) he was appointed CNRS researcher at Descartes University (Paris). In 1998 he became Professor of Organic Chemistry at the University of Clermont-Ferrand, and in 2006 he transferred to his current position as Professor at the University Paris Sud (Orsay). His research interests, con-

2008, he joined the group of Prof. D. J. Aitken at the University of Paris-Sud (Orsay) where he carried out his PhD studies on the photochemistry of cyclopentenones and alkenes beyond the [2+2] stage. He collaborated during his PhD with the group of Prof. P. P. Piras from the University of Ca-

tive post-doctoral research in the group of Prof. D. J. Aitken in the ICMMO, University Paris Sud (Orsay), he joined the academic staff at the University of Cagliari in 2010. He is currently Assistant Professor of Organic

ences degree in Organic Chemistry and then spent a year (1987) as a postdoctoral fellow at the Dyson Perrins Laboratory (Oxford) with Dr. S. G. Davies. He then returned to Orsay and is presently Chargé de Recherche in the ICMMO,

1990–1991 he spent a sabbatical year at the Laboratoire des Carbocycles, University of Paris-Sud (Orsay) working on cyclopropane derivatives with Dr. J. Salaün. Returning to Cagliari, he was appointed Associate Professor in 1985, then full Professor of Organic Chemistry in 2001. In ducted in the ICMMO research institute, include the synthesis of functionalized small-ring compounds, particularly unnatural amino acids, as building blocks for foldamers and peptidomimetics, and synthetic organic photochemistry.

gliari (Italy) on an organocatalysis project. He finished his PhD in 2012 and joined the group of Professor I. Marek in the Technion Institute (Israel). His research interests now include zirconium-mediated reactions and carbometalation of small rings.

Chemistry at the Faculty of Sciences. His research interests include asymmetric synthesis and development of new synthetic methods based on transformation of strained organic compounds.

University Paris-Sud. His research interests include the chemistry and applications of small-ring compounds implicating stereoand enantioselective reactions, ring expansions and, more recently, organocatalysis.

2009 he was a Visiting Professor in the University of Paris-Sud (Orsay). His primary research interests focus on the synthesis and reactivity of strained carbocycles, the synthesis of natural products, and asymmetric organocatalytic reactions.

3

Overview of 2-Cyclopentenone Reactivity

While the purpose of this review is to relate the main synthetic approaches for the preparation of 2-cyclopentenones, it is useful to present here a brief summary of the chemical reactivity of this molecular core, for two reasons. Firstly, the diversity of chemical transformations that can be carried out thereupon goes some way to explaining the popularity of the system, and secondly, these reactivity features should be kept in mind when planning the synthesis of any particular 2-cyclopentenone derivative. The core structure is highly reactive, with methods available for the modification of every position (Figure 2).



Figure 2 The multiple reactivity profile of the 2-cyclopentenone core structure

At the 1-position, the carbonyl group can react in a typical manner with nucleophiles that give regioselective 1,2-additions to conjugated ketones, such as Luche reduction or addition of organometallic reagents.³

The 2-position can be functionalized in a number of ways. The conjugate addition of a weak nucleophile to the 3-position creates an enolate, which reacts at C2 with carbonyl compounds⁴ or imines⁵ (in Baylis–Hillman-type reactions) or with arylating reagents.⁶ 2-Halocyclopentenones can be prepared similarly⁷ or by other straightforward means⁸ and then engaged in carbon–carbon bond-forming reactions such as palladium-catalyzed couplings⁹ or radical induced additions.¹⁰ 2-Cyclopentenone-2-boronic acids can also be prepared for cross-coupling reactions.¹¹

The 3-position can easily be functionalized via conjugate addition of a variety of nucleophiles, both organic¹² and heteroatomic,¹³ and these reactions are often amenable to high degrees of enantiomeric control. Heck-type reactions can also be carried out on this position.¹⁴ The tandem sequence of nucleophilic attack at C3 followed by electrophilic capture of the intermediate enolate at C2 is an elegant route to double functionalization.¹⁵

The 4-position can be brominated with *N*-bromosuccinimide,¹⁶ which opens the way to further functionalization at this position.¹⁷ Vinylogous deprotonation–alkylation procedures generally require a heteroatom substituent at the 3-position, as well as conditions which provide the thermodynamic enolate.¹⁸

The 5-position can react as a typical α carbon to a carbonyl function. The generation of a kinetic enolate allows regioselective electrophilic alkylation at C5.¹⁹ Aldolizations have been described,²⁰ with recent developments allowing for enantiomeric control.²¹

2-Cyclopentenones are frequent partners in cycloaddition reactions, including photochemical [2+2]-cyclo-additions²² and Diels–Alder reactions.²³ Stereoselective

 \mathbbm{C} Georg Thieme Verlag Stuttgart \cdot New York

cyclopropanations,²⁴ aziridinations,²⁵ and epoxidations²⁶ are also feasible reactions.

2 Multicomponent Ring Assembly

2-Cyclopentenones can be prepared by assembling two or more components through the formation of at least two new σ bonds, generally in a sequential fashion. This is a versatile and often efficient approach for the construction of the target core and it allows for considerable structural diversity. Metal catalysts are often employed.

(2+2+1) Ring Assembly

The historic example of this type of approach is the Pauson-Khand reaction, in which a cyclopentenone is formed from an alkene and an alkyne in the presence of $[Co_2(CO)_8]$. The generally accepted mechanism (Scheme 1) shows how provision can be made for substituents in all positions. The intermolecular version can be qualified as a (2+2+1) ring assembly. Regioselectivity is an important issue, and is dependent on steric factors: usually R¹ is larger than R². Strained or reactive alkenes are privileged; congested alkenes react less well. The intramolecular version - formally a (4+1) assembly, but treated here nonetheless - overcomes a good number of the selectivity issues, and provides a versatile entry to polycyclic skeletons. The reaction has been widely studied and considerable synthetic use has been made thereof. Progress has been reviewed regularly,27 with particular attention paid to intramolecular²⁸ and catalytic²⁹ versions, and a comprehensive monograph has appeared very recently.30 Amongst the numerous developments of the Pauson-Khand reaction, it is worth noting that complexes of metals other than cobalt may serve as catalysts,³¹ while formates or aldehydes can be used as safer CO sources.³² The presence of tertiary amine N-oxides is thought to have an accelerating effect, helping in the oxidative removal of one CO from the alkyne-cobalt complex.





A few recent applications of the Pauson–Khand reaction are presented here (Scheme 2). Compound 1 was subjected to a one-pot reduction–lactonization–Pauson–Khand reaction to give highly functionalized hydropentalenes 2 with high stereoselectivity.³³ Conditions were found for the Pauson–Khand reaction of unactivated alkene 3 with alkyne 4 to provide the 2-cyclopentenone 5 on a multigram scale, as the first step in a total synthesis of the marine alkaloids (±)-axinellamines A and B.³⁴ In the preparation of structural analogues of the anti-cancer sesquiterpene thapsigargin, an intramolecular rhodiummediated Pauson–Khand reaction was carried out on allene–alkyne 6 to close a seven-membered ring and give product 7 in good yield.³⁵ In a recent evaluation of synthetic routes to polycyclic targets, chiral dienediynes 8 were found to undergo highly chemoselective Pauson– Khand reaction in benzaldehyde, which served as the CO source, using [Rh(cod)Cl]₂ as the catalyst in the presence of racemic BINAP. High *cis*-diastereoselectivity (up to >20:1) was observed in the products 9, particularly when bulky substituents were borne adjacent to the chiral center.³⁶





A π -allylic precursor can be used instead of the alkyne component. For example, reaction of allyl methyl carbonate with norbornene gave a good yield of the cyclopente-none adduct **10** with exclusive *exo*-selectivity (Scheme 3).³⁷



Scheme 3

Reductive (2+2+1) cyclocarbonylations of internal alkynes require more drastic conditions than those habitually employed in the Pauson–Khand reaction, but they have been achieved using a rhodium catalyst in the presence of urea under high carbon monoxide pressure (Scheme 4). A high diastereomeric excess was observed in the products **11**, with *cis*-isomers arising from dialkylalkynes and *trans*-isomers from diarylalkynes. In the latter cases, yields were lower due to the formation of a lactone byproduct.³⁸





(3+2) Ring Assembly

In an effort to overcome some of the limitations of intermolecular Pauson–Khand reactions, a number of (3+2)ring assemblies have been considered. Construction of the five-membered-ring target has been achieved using most of the conceivable disconnections.

Nickel-catalyzed cycloaddition of α , β -unsaturated phenyl esters **12** with internal alkynes provided trisubstituted 2-cyclopentenones **13** (Scheme 5). The regioselectivity varied from poor to excellent, depending on the alkyne used, while terminal alkynes were inefficient substrates. A mechanism was proposed, implicating a η^3 -oxaallyl phenoxynickel intermediate.³⁹



Scheme 5

Nickel-mediated cyclization of alkenyl Fischer carbenes 14 with internal alkynes⁴⁰ provided a wide selection of adducts 15 in a highly regioselective manner (Scheme 6). A recent variation employed a chromium alkynylcarbene 16 and an alkenyl organolithium 17, both of which were prepared from simple precursors. Again, a variety of substitution patterns were accommodated in the products 18, and the method can be adapted for enantioselective cyclizations.⁴¹ Mechanistic models for these transformations were proposed by the authors.



Scheme 6

Vicinal donor-acceptor disubstituted cyclopropanes are convenient precursors for 1,3-dipoles. The Lewis acid mediated reaction of cyclopropanes **19** with silvl ynol ethers gave the [3+2]-cycloaddition adducts which spontaneously eliminated ethanol to give the cyclopentadienes **20**, which required deprotection with hydrofluoric acid to give the corresponding 2-cyclopentenones **21** (Scheme 7).⁴²



Scheme 7

Zinc chloride promoted a [3+2]-cycloaddition between isoprenyl chloride and methylthio phenylthio ethyne (Scheme 8). The reaction lacked regioselectivity, but the two adducts **22** and **23** were separated and transformed easily into the corresponding phenylthio 2-cyclopentenones **24** and **25** in good yields. The chemistry of the thioether function was exploited in order to access further derivatives.⁴³



Scheme 8

Interesting results have been observed using allenes as either the two-carbon or three-carbon component in (3+2) assemblies (Scheme 9). Chiral α -ethers of allenyl carboxamides **26** reacted with alkenyllithium reagents to give adducts which, upon addition of acid, gave transient protonated vinyl alkenyl ketones that cyclized to chiral 2cyclopentenones **27** in a conrotatory 4π -electron process, in Nazarov fashion (*vide infra*).⁴⁴ It was suggested that this process resulted in axial-to-tetrahedral chirality transfer.⁴⁵ In a complementary fashion, reaction of a chiral lithiated allene **28** with an α -methylcinnamide followed by acidic treatment gave the enantiomerically enriched 2cyclopentenone derivative **29**.⁴⁶





An organocatalytic iminium ion/N-heterocyclic carbene tandem reaction sequence has been used to combine α,β unsaturated aldehydes and β -keto phenyltetrazolesulfones **30** to give 2,4-disubstituted 2-cyclopentenones **32** in a highly enantioselective manner (Scheme 10). Besides the elegance of the sequential organocatalyzed asymmetric Michael addition–benzoin condensation, the judicious inclusion of a phenyltetrazolesulfone (SO₂PT) moiety facilitated a Smiles rearrangement to liberate the target 2cyclopentenones **32**; a rational mechanism for this was proposed.⁴⁷



Scheme 10

Tandem condensation-Wittig cyclization reactions between 2-oxo- or 2,4-dioxo-alkylidinephosphoranes and glyoxals or diacylolefins represent another (3+2) type of assembly leading to cyclopentenones in an efficient manner, although cyclohexenone formation may be a competing process. Some aspects of this methodology were reviewed recently.⁴⁸ In the reaction of phosphoranes 33 with a series of maleic diesters, the 2-cyclopentenone products 34 were obtained in moderate yields but excellent diastereoselectivities (Scheme 11).49 One study was carried out using a chiral sulfoxide derivative of 2-oxopropylidine phosphorane 35, prepared in situ and treated with a series of (E)-enediones.⁵⁰ In the presence of a key copper additive, the Michael addition and subsequent intramolecular Wittig reaction proceeded in a highly regioand stereoselective fashion to give the corresponding 3methyl-5-sulfoxy-2-cyclopentenones 36. Conditions were also established for the facile removal of the sulfoxide adjuvant, and the resulting 3,4-disubstituted 2-cyclopentenones 37 were obtained with very high enantiomeric excess (Scheme 11).





Intermolecular condensations involving classical carbanion chemistry appear to offer a simple route to the 2-cyclopentenone core, but side reactions often limit the synthetic utility. Nevertheless, successful applications of such chemistry do appear. A number of recent papers have described crossed-aldol condensations between benzil derivatives and selected ketones to give highly functionalized 4-hydroxy-2-cyclopentenones, usually as diastereoisomeric mixtures.⁵¹

A comprehensive study of the cyclization of 1,2-diketones with 1,3-dicarbonyl dianions **38**, or with the correspond-

ing silyl enol ethers **39** under acidic conditions, revealed this to be a convenient and quite general approach for the preparation of a series of 2-acyl-4-hydroxy-2-cyclopentenones, **40** or **41**, respectively (Scheme 12). The basemediated reactions required a silica gel treatment to induce the cyclization, while the more direct silyl enol ether approach gave slightly lower yields.⁵²



The reaction of 3-substituted allenoates **42** with symmetrical diaryl 1,2-diketones in the presence of a phosphine gave highly substituted 2-cyclopentenones **43** in excellent yields (Scheme 13). The reaction appeared to be highly diastereoselective although data were not given. The reaction also proceeded with unsymmetrical diones, although without regioselectivity. A zwitterionic adduct formed from the allene and the phosphine was proposed as the key reactive intermediate, which first attacked one carbonyl with elimination of water then attacked the second carbonyl with water assistance to induce cyclization.⁵³





(4+1) Ring Assembly

An alternative approach to the 2-cyclopentenone core is a (4+1) assembly, which has the clear advantage of removing regioselectivity issues. Carbon monoxide is the obvious 'one-carbon' component, providing C1 of the target structure, but a few other reagents have been used successfully to provide C3 or C5 when combined with appropriate 'four-carbon' partners.

Carbonylative cyclization of 1,3-butadiene derivatives in the presence of carbon monoxide has been known for some time, and is still considered a pertinent strategy. A

7

This document was downloaded for personal use only. Unauthorized distribution is strictly prohibited.

series of bicyclic enones **45** was prepared in excellent yields from the appropriate halogeno-dienes **44** using a palladium catalyst under an atmosphere of carbon monoxide (Scheme 14).⁵⁴



Scheme 14

An interesting recent development is the carbonylation of a 1-lithiobutadiene **46** followed by spontaneous cyclization to give a cyclopentadienyl enolate **47** (Scheme 15). It was shown that this intermediate could be trapped by acylation at the γ -position, providing the corresponding 2cyclopentenone **48** in a one-pot process. However, the system was sensitive to steric factors and α - or O-acylation products were obtained when the 4,5-positions were substituted.⁵⁵



Scheme 15

A titanium-mediated (4+1) assembly of 1,3-butadienes and nitriles has been described, in which the nitrile acts as the 'one-carbon' component.⁵⁶ Treatment of 2-silyloxybutadiene **49** with titanium isopropoxide and a Grignard reagent gave a titanacyclopentene intermediate which reacted with a nitrile to give a silyloxycyclopentenylamine **50**; spontaneous hydrolysis during work-up led to the corresponding 2-cyclopentenone **51** directly (Scheme 16).



Scheme 16

Vinyl ketenes are useful intermediates in synthesis and they can be stabilized to some extent as trialkylsilyl derivatives. A selection of such vinyl ketenes **52** reacted with nucleophilic carbenes, generated in situ thermally, to provide highly substituted 2-cyclopentenones **53** in good yields (Scheme 17).⁵⁷ In another study, the reaction of vinyl ketene **54** with selected α -benzotriazolyl (Bt) organolithium reagents **55** gave the 2-cyclopentenones **56** in fair to good yields, although the addition of a Lewis acid was sometimes necessary to facilitate departure of the Bt group (Scheme 17). The products in this case were obtained with good *trans*-stereoselectivity, particularly when the 5-position was monosubstituted.⁵⁸





A selection of stable silvl vinyl ketenes bearing tricarbonylchromium(0) arene substituents 57 were prepared from Fischer carbene complexes and alkynes, and reacted with diazomethane, or a derivative thereof, to give the (4+1)annulation products 58 in excellent yields and in a completely stereoselective fashion (Scheme 18).59 Removal of the chromium moiety was subsequently achieved using cerium(IV) ammonium nitrate. This process was applied in an elegant intramolecular mode, to provide an efficient synthesis of the rocaglamide skeleton, whereby the Fischer carbene alkyne 59 was transformed in a three-step process into adduct 60 in good yield and complete stereoselectivity (Scheme 18).⁶⁰ Related studies showed that the Köbrich reagent (CH₂I₂ with BuLi) could be used instead of a diazoalkane, while the use of tert-butyl isocyanate provided the 2-cyclopentenone core with an exocyclic (Z)-imine moiety at the 5-position.⁶¹





The reaction of methylenecyclopropanes **61** with Fischer carbene chromium complexes **62** provided 2-cyclopentenones **63** in which the ring had been formed from all four of the methylenecyclopropane carbon atoms plus one equivalent of carbon monoxide (Scheme 19). The proposed mechanism involved an initial [2+2]-cycloaddition followed by a rearrangement to give an intermediate alkylidinemetallacyclopentane which then underwent CO insertion, chromium elimination, and finally isomerization.⁶²





3 Cyclizations

In this review, 'cyclization' implies the formation of a new ring structure from an acyclic molecule (or from an acyclic fragment of a larger molecule) through the formation of one new σ bond. This definition includes cases where a new π system is generated (or an existing π system is shifted) as the σ bond is formed. A comprehensive review of ring-closure approaches to cyclopentane derivatives, including some useful precursors of 2-cyclopentenones, appeared recently.⁶³

Nazarov Cyclization

Arguably one of the most important methods for the preparation of 2-cyclopentenones is the acid-promoted cationic pericyclization of a divinyl ketone, first reported by Nazarov in 1944.⁶⁴ It is now established that the reaction is initiated by acid complexation of the ketone to give a pentadienyl cation which undergoes a 4π -electron cyclization in a conrotatory fashion to provide an oxyallyl cation intermediate. Elimination of a proton followed by reprotonation of the acid-bound enolate gives the 2-cyclopentenone product (Scheme 20).





The obvious synthetic potential was for some time offset by selectivity issues, involving the regiochemistry of the proton elimination step and the stereochemistry of the enolate protonation step, as well as the harsh acidic conditions which were often required. As work progressed, it emerged that steric and/or electronic effects (particularly polarized double bonds) could be harnessed to control the selectivity, and milder reaction conditions were discovered. Significant recent developments include the use of organocatalysts⁶⁵ and transition-metal-complex catalysts,⁶⁶ which open the way to enantioselective reactions and/or tandem transformation sequences. The Nazarov reaction has established itself in the modern synthetic chemist's toolbox, and progress has been documented regularly in comprehensive reviews, particularly in the last decade,⁶⁷ and include focuses on catalytic versions,⁶⁸ asymmetric versions,⁶⁹ and alternative routes to the intermediate pentadienyl cation in order to circumvent the highly reactive divinyl ketone substrates.⁷⁰ Processes in which the cyclopentenyl cation intermediate is intercepted by a nucleophile constitute a rich and developing area referred to as 'interrupted Nazarov reactions', but generally they deviate the reaction course away from 2-cyclopentenone formation.⁷¹

Only a few of the many recent elegant applications of the Nazarov cyclization are presented here (Scheme 21). In a synthesis of (\pm) -xanthocidin, very fast cyclization of the highly substituted divinyl ketone 64 was achieved using iron(III) chloride. The sterically challenged oxyallyl cation intermediate underwent exocyclic elimination leading to the 5-methylene-2-cyclopentenone product 65 in less than three minutes.⁷² The preparation of a 2-hydroxycyclopentenone core can be achieved using a vinyl diketone as the precursor. As part of the total synthesis of (+)-fusicoauritone, the acidic treatment of the macrobicycle 66 with a Lewis acid gave the requisite tricyclic product 67 with an all-syn stereochemistry.73 The Nazarov cyclization of the 3-acylated benzofuran 68 was the key step in a short formal synthesis of (\pm) -methyl rocaglate; while other Lewis acids only induced a retro-Friedel-Crafts reaction, acetyl bromide was able to induce cyclization to give 69 in very good yield.⁷⁴ A chiral Brønsted acid was used to catalyze the cyclization of 70 with excellent yield and torquoselectivity, to furnish 71 with 82% ee in a concise formal synthesis of (+)-roseophilin. In principle, water is required to trap the oxyallyl intermediate and presumably was furnished by the reagent-grade carbon tetrachloride used as the solvent.⁷⁵ Copper(II)-complex-mediated cyclizations may be accompanied by skeletal rearrangements, again at the oxyallyl cation stage, and such a process was exploited in a total synthesis of enokipodin B. A bulky bisoxazolidine copper(II) complex induced a sequential cyclization-double-[1,2]-Wagner-Meerwein shift transformation of divinyl ketone 72 (as an easily isomerized mixture, of which only the Z-isomer reacted) to give the 2-cyclopentenone 73 with impressive regioselectivity, although the enantioselectivity was poor.⁷⁶



Scheme 21

Treatment of chiral oxazolidinone-derived divinyl ketones **74** with methanesulfonic acid gave good yields of Nazarov cyclization products **75** as single regioisomers and with excellent diastereoselectivity (Scheme 22).⁷⁷ A unique role for the chiral auxiliary was proposed during the cyclization process, whereby two transition-state conformers are involved, each of which progresses with the same torquoselectivity directed by allylic strain.⁷⁸



Scheme 22

An impressive multi-step one-pot Wittig–Nazarov protocol was conceived for the construction of 4-alkylidene-2cyclopentenones starting from α -diazoketones **76** and acid chlorides **77** (Scheme 23). The transformation involved iron-catalyzed formation of the stabilized ylide **78** on the one hand and base-induced formation of a ketene on the other hand. These intermediates reacted together to form a vinyl allenyl ketene **79**, which then underwent trifluoro-acetic acid mediated Nazarov cyclization to give the requisite products **80** in good yields and with a high selectivity for the *Z*-isomer.⁷⁹



Scheme 23

Substrates other than divinyl ketones have been developed for the generation of pentadienyl cations, which further enhances the scope of conrotatory 4π -electron cyclizations. Reactions in this category are commonly referred to as Nazarov cyclizations, although there is some convergence with the Rautenstrauch rearrangement and perhaps also with pentadienal cyclizations (vide infra). Reduction of vinylalkylidine dioxolanones 81 provided the corresponding 5-hydroxy-2-cyclopentenones 82 as single diastereoisomers, presumably via conrotatory closure of an intermediate 1,2-oxidopentadienyl cation (Scheme 24).⁸⁰ The strategy was therefore applied in a key transannulation step in a total synthesis of (\pm) -cephalotaxine.⁸⁰ Selective oxidation of alkoxyallenes 83 using dimethyldioxirane provided an entry to oxypentadienyl zwitterions, which cyclized in a Nazarov fashion to give the bicyclic adducts 84, often as single cis-isomers (Scheme 24). A mechanism involving diastereoselective epoxidation directed by the difference in the steric bulk of the allene substituents followed by the usual concerted 4π electron cyclization was evoked to explain the diastereoselectivity.81



Scheme 24

Rautenstrauch Rearrangement

The palladium(II)-catalyzed isomerization of 1-ethynyl-2-alkenyl acetates to give 2,3-disubstituted 2-cyclopentenones, was first reported in 1984.⁸² The proposed mechanism involves consecutive metal additions to the π systems and migration of acetate, with a hydrolysis step to liberate the enone (Scheme 25).



Scheme 25

The palladium(II) version of the reaction is still employed today: in a stereoselective assembly of the ABCE ring system of the natural product azadirachin, the E ring was constructed in this way from the enyne ester **85**, giving the tetracyclic adduct **86** as a 1:1 mixture of diastereoisomers (Scheme 26).⁸³



Scheme 26

Most recent developments of this reaction have focused on the use of gold(I) catalysts. A plausible mechanism involves initial formation of an alkyne-gold complex which undergoes a [1,2]-shift of carboxylate to generate a goldcoordinated allylic cation, which then evolves by a Nazarov-type cyclization process. The resulting acyloxy cyclopentadiene is hydrolyzed to give the bicyclic product. The standard process was applied to a series of 1ethynyl-2-alkenyl pivaloates 87 to allow access to a wide range of 3,4- or 3,5-disubstituted 2-cyclopentenones 88 under mild conditions (Scheme 27).⁸⁴ When a nonracemic substrate 89 was used, conditions were found in which the 3,4-disubstituted products 90 could be obtained with excellent chirality transfer.84 In related work, 1-ethynyl-1allenylalkyl acetates 91 underwent cycloisomerization to acetoxyfulvenes 92, in the presence of a cationic bisoxazolidine-gold(III) complex, that then evolved to provide 4-methylene-2-cyclopentenones **93** upon methanolysis.⁸⁵

In further developments of this theme, conjugated enynyl derivatives have been cyclized, again with gold-mediated migration of an oxygen function (Scheme 28). Enynyl acetates **94** were treated with a gold(I) complex to give



Scheme 27

3,5-disubstituted or 3,4-fused bicyclic 2-cyclopentenones 95. It was suggested that tandem gold(I)-catalyzed [3,3]rearrangement of the substrate and activation of the allenic acetate led to the gold-coordinated allylic cation intermediate, which cyclized as indicated above.⁸⁶ 5-Silyloxypent-3-en-1-ynes 96 underwent cyclization when treated with a gold(I) catalyst, to give 2-cyclopentenones 97 in generally good yields. In this case it was proposed that gold complexation of the alkyne induced siloxycyclization followed by carbon–oxygen bond fragmentation to give an allylic carbocation, which subsequently cyclized at the gold-bound alkenyl site, leading to the substituent topology observed in the final products.⁸⁷





The hydrative rearrangement of 1,1-diethynylcarbinol acetates **98** to 5-acetoxy-2-alkyl-2-cyclopentenones **99** was achieved using a gold(I) catalyst (Scheme 29). The substrates first underwent a gold-mediated [3,3]-rearrangement to allenyne acetates which then reacted further by gold-induced oxacyclization then nucleophilic attack by water. Gold-promoted 5-*endo*-dig cyclization provided the cyclic products, accompanied by small amounts of the allenone by-product **100**.⁸⁸





Hydrative Carbocyclization

In a somewhat different fashion from the reactions described above, the hydrative carbocyclization of 1,5-diyn-3-ones **101** was achieved using a gold(I) catalyst to furnish 4-acyl-2-cyclopentenones **102**, although other acyclic products were obtained as well (Scheme 30). An important difference compared to the Rautenstrauch process is the absence of oxygen moiety migration, meaning that the ketone carbon of the substrate was retained as C1 in the cyclic product. One of the alkyne carbon atoms was incorporated as the acyl function in an exocyclic locus in the product structures. It was suggested that the goldmediated hydration of the electron-rich alkyne generated a gold enolate which then cyclized to a gold cyclopentenonyl intermediate.⁸⁹





Ring-Closing Metathesis

One of the most important developments in metathesis has been to provide a tool for the closure of organic ring systems. Ring-closing metathesis reactions have been reviewed regularly, notably with regard to their applications to natural product synthesis.⁹⁰ Five-membered-ring closure is generally easy, but the reaction is sensitive to electronic factors, with electron-poor alkenes being lessfavored substrates. As a result, most ring-closing metathesis studies have targeted 3-cyclopentenols, although these compounds are often readily oxidized to 2-cyclopentenones without difficulty. Here, we report only on ring-closing metathesis reactions that provide 2-cyclopentenones directly.

Despite the involvement of a deactivated alkene, the Grubbs I catalyst induced the cyclization of diene **103** to

 \tilde{C} Georg Thieme Verlag Stuttgart \cdot New York

provide 2-cyclopentenone **104** in 67% yield (Scheme 31).⁹¹ The use of the Grubbs II catalyst to transform the substrates **105** was more efficient, and provided the series of 2- and/or 4-substituted 2-cyclopentenones **106** in excellent yields.⁹² As part of the total synthesis of (–)-heptemerone B, the ring-closing metathesis reaction of the highly substituted substrate **107** was conducted using Grubbs II without incident to give the key intermediate **108**, again in excellent yield.⁹³ Comparable ring-closing metathesis processes were used to obtain single enantiomers of Bocprotected 2- and 5-amino-2-cyclopentenones.⁹⁴





Tandem ring-opening and ring-closing procedures have been described within the context of elegant syntheses of complex natural product skeletons (Scheme 32). For the construction of the tricyclic core of tricycloclavulone, readily available compound **109** was treated with the Grubbs I catalyst to provide the requisite bicyclo[3.3.0] ring system of **110** in good yield.⁹⁵ A tandem ring-opening and double-ring-closing metathesis protocol was applied in a total synthesis of (+)-cyanthiwigin U. Starting from the dialdehyde **111**, double vinyl Grignard addition followed by oxidation gave the alkene-bis(enone) **112** which was treated immediately with the Grubbs II catalyst to provide the desired tricyclic product **113** in 43% yield for three steps.⁹⁶





The metathesis reactivity of 2-alkylidene-1,3-dicarbonyls is low; even so, when substrate **114** was subjected to the

Grubbs II catalyst, the highly substituted cyclic keto ester **115** was obtained in an acceptable 65% yield (Scheme 33). With the less bulky substrate **116**, the requisite ringclosing metathesis product **117** was obtained in higher yield using a lower catalyst loading.⁹⁷





Related to the ring-closing metathesis reaction is the socalled ring-closing enyne metathesis, which can provide a useful access to vinylcyclopentenes.⁹⁸ However, the reaction is not efficient with the appropriate precursors for the formation of 2-cyclopentenones (Scheme 34). Cyclization of the alkene-ynone **118** in the presence of Grubbs II catalyst gave the 2-vinyl-2-cyclopentenone **119** in 32% yield,⁹¹ while the alternative alkyl-enone mode prevalent in **120** (albeit with a deactivated enone) evolved only in the presence of titanium(IV) isopropoxide and even then gave a meager 4% yield of the 3-vinyl-2-cyclopentenone product **121**, accompanied by other cross-metathesis products.⁹⁷





Other Transition-Metal-Mediated Cyclizations

Intramolecular oxidative Heck coupling has been reported using vinyl 2-bromovinyl carbinols **122** as substrates: the 5-endo-trig cyclization gave the 2-cyclopentenones **123** directly in decent yields (Scheme 35).⁹⁹ With 2-methylallyl 2-bromovinyl carbinol substrates **124**, the cyclization mode switches to 5-exo-trig which furnishes the corresponding 4,4-dimethyl-2-cyclopentenones **125** in good yields.¹⁰⁰ Propargyl 2-bromovinyl carbinols **126** have also been transformed into 4-methyl-2-cyclopentenones **127**, although in this case the yields were modest.¹⁰¹

The Liebeskind–Srogl coupling reaction was applied in intramolecular mode to a highly functionalized 2-vinyl-stannane thioester **128** (Scheme 36). After optimization,











4-Alkynals **130** were converted into 2-cyclopentenones **131** by way of a rhodium(I)-catalyzed intramolecular hydroacylation process (Scheme 37). Substituents in any position were compatible with the process, although no 4,5-disubstituted case was examined; acetone was required as the solvent in order to obtain good yields.¹⁰³ With a coordinating methoxy group in the 4-position of the substrates **132**, the employment of a chiral phosphine ligand and a noncoordinating solvent, the hydroacylation provided an excellent kinetic resolution, giving the enantiomerically enriched 4-methoxy-2-cyclopentenones **133**. The other enantiomers of the substrates **132** were either recovered intact, or were converted into the isomeric 2-alkylidene-cyclobutanones **134**, depending on the phosphine ligand employed.¹⁰⁴



Scheme 37

Pentadienal Cyclizations

The δ -carbon of a doubly conjugated carbonyl moiety is not particularly nucleophilic, and few attempts to effect cyclization of such compounds had been described until recently. A study of the reactivity of simple 2.4-dienals 135 in the presence of a Lewis acid demonstrated the feasibility of the approach but also revealed some limitations: the 2-cyclopentenones 136 were obtained in only modest yields (Scheme 38).¹⁰⁵ The cyclic aliphatic 2,4-dienal 137 provided the corresponding bicyclic enone 138 in poorer yield, and the cyclization failed entirely when the γ -methyl group was absent, or when attempted with a triple-conjugated substrate. Several mechanistic possibilities were suggested, implicating the formation of a cyclopentadiene epoxide either by a concerted process or via a Nazarovlike mechanism involving the conrotatory 4π -electron cyclization of an oxypentadienyl cation; isomerization of the epoxide in the acidic medium would account for the formation of the final products (Scheme 38).¹⁰⁵



Scheme 38

An improvement was devised, on the premise that the δ carbon nucleophilicity would be enhanced by making it a part of a vinylogous allyl silane system. In the event, when the silylated precursors **139** were treated with a Lewis acid they provided, after isomerization, the spiro derivatives **140** as single diastereoisomers in good yields (Scheme 39). The excellent diastereoselectivity was explained by a preferred carbonyl coordination by the Lewis acid from the less-hindered face of the six-membered ring.¹⁰⁶ In a study of the various cyclization modes possible for the cyclic dienal **141**, it was found that platinum(II) chloride in the presence of *p*-toluenesulfonic acid, the latter being used to induce isomerization, drove the reaction to exclusive formation of the fused 2-cyclopentenone **142**.¹⁰⁷

Base-Induced Annulations

Base-mediated condensations of carbonyl compounds have been a mainstay of organic synthesis for over a century. They are still popular methodologies, notably for the construction of cyclic structures, including the title family of compounds. A few recent applications are presented here.





Intramolecular aldolization of a 1.4-diketone or a 4-ketoaldehyde moiety followed by crotonization leads conveniently to the 2-cyclopentenone core, usually as the thermodynamically preferred structural isomer (Scheme 40). A series of 4,4-spiroannulated 2-cyclopentenones 144, inspired in part by the acorone natural product skeleton, were prepared from the cyclic keto aldehyde precursors 143 in uniformly good yields regardless of the existing ring size.¹⁰⁸ The final step in the synthesis of a model azatriquinane 146 was achieved easily by subjecting compound 145 to mild basic conditions.¹⁰⁹ Sodium hydride was used to cyclize the diketone 147 to provide bicyclic derivative 148 in good yield, as part of the first total synthesis of (+)-minwanenone.¹¹⁰ The Schöllkopf bislactim 149 was used to prepare the unusual α -amino acid 150, featuring the 5,5-disubstituted 2-cyclopentenone moiety, in enantiomerically enriched form.¹¹¹



Scheme 40

A thiol-mediated tandem Michael–aldol reaction of the dihydronaphthalene derivatives **151** has been described as a route to fused cyclopentenones **152** (Scheme 41). The proposed mechanism resembles an intramolecular Baylis–Hillman reaction: the thiol adds in a conjugate manner to the α , β -unsaturated ester chain of the bicyclic system,

then base-induced condensation of the α -carbanion on the aldehyde, followed by elimination of the nucleophile and prototropy, leads to the 2-cyclopentenone moiety.¹¹²





Other related classical enolate-type condensation procedures have been used fruitfully in intramolecular mode (Scheme 42). With the diketo ester substrate 153, the Knoevenagel reaction was employed to close the fivemembered ring in the 5,7,5-tricyclic system of 154 in a total synthesis of (\pm) -sordaricin.¹¹³ A Knoevenagel reaction was also carried out on the polyfunctional hydroxypyranones 155, which are masked 4-keto-2-enals. Despite a number of potential condensation paths being available, treatment of these compounds with piperidinium acetate gave good yields of the cyclopenta[b]pyran derivatives 156, which were subsequently used to prepare natural product analogues for cytotoxicity evaluation.¹¹⁴ The triester 157 was converted smoothly into the cyclopent-2enone-5-carboxylic ester derivative 158 by a Dieckmann cyclization during the initial steps of a short synthesis of (-)-kjellmanianone.¹¹⁵





A novel endocyclic keto-enamine annulation was discovered during a total synthesis of (\pm) -cephalotaxine (Scheme 43). Treatment of compound **159** with ferrous sulfate under aerobic conditions provided the pentacyclic

Synthesis 2014, 46, 1-24

adduct **160** in reasonable yield.¹¹⁶ The oxygen-dependent acid-mediated formation of a conjugated iminium ion was proposed to account for this reaction.





Using the highly functionalized cyclohexanone **161**, an intramolecular Horner–Wadsworth–Emmons reaction was performed to construct the five-membered ring of **162** in good yield during a total synthesis of the neurotrophic modulator (\pm) -jiadifenin (Scheme 44).¹¹⁷ The detailed study of a range of 2,5-diketophosphonates **163** revealed that the choice of base was critical for the success of the Horner–Wadsworth–Emmons reaction to provide **164**. Furthermore, the by-product **165** resulting from a competing intramolecular aldol reaction was sometimes observed, and non-racemic substrates suffered from stereochemical erosion in some cases.¹¹⁸





Other Cyclizations

Selected α -(2-chloro-2-propenyl)- α -aryl acetic esters **166** were cyclized under Friedel–Crafts acylation conditions to give the corresponding 3-chloro-5-aryl-2-cyclopentenones **167** (Scheme 45). This was the key step in the development of a short and general method for obtaining 3,5diaryl-2-cyclopentenones **168**.¹¹⁹





The rhodium(II)-mediated decomposition of diazo compounds **169** induced an intramolecular C–H insertion, to furnish the corresponding cyclopent-2-enone-5-carboxylic esters **170** as single diastereoisomers in excellent yield, although the reaction failed with an aromatic substituent in the δ -position (Scheme 46).¹²⁰ An intramolecular C–H insertion reaction was applied in the total synthesis of (+)przewalskin B, whereby the diazo compound **171** gave the spiro-2-cyclopentenone derivative **172** in good yield when treated with rhodium(II) acetate.¹²¹





4 Transformations of Existing Cyclic Systems

As stated in the introduction, this review will not cover the extensive work done on the transformation of existing cyclopentanoid rings. However, there are some approaches to the title compound family that involve the reorganization of other, non-cyclopentane, ring systems. Two distinct categories of reactions can be identified. In one case, some masked chemical reactivity of the cyclic starting material is revealed upon treatment with a specific reagent. Often this implies a 1,4-dicarbonyl intermediate, which undergoes aldolization or crotonization, so there is some overlap with reactions discussed in the section above. In the second case, energetic factors are at play in the rearrangement of the ring system, either due to inherent ring strain or photochemical activation.

Rearrangements of Furans and Pyrans

Furans are an excellent source of 1,4-dicarbonyl compounds, and the literature on the preparation of these species and their transformations into 2-cyclopentenones up to the mid-1990s were reviewed extensively.¹²² Furans may be conveniently transformed into pyranones via the Achmatowicz reaction. Alternatively, sugars provide a facile entry to diverse pyran systems.

2-Furylcarbinols undergo isomerization to 4-hydroxy-2cyclopentenones upon treatment with either Brønsted or Lewis acids, a process sometimes referred to as the Piancatelli rearrangement. The accepted mechanism involves a series of cationic intermediates, and a final 4π - electron ring closure,¹²³ which ensures a *trans* configuration for the substituent at the 5-position (Scheme 47). In some cases, the acid medium (or more conveniently, basic treatment) induces further rearrangement to the more thermodynamically stable 2-substituted 4-hydroxy-2-cyclopentenones. Applications of such procedures continue to appear as part of multi-step syntheses.¹²⁴





In a large-scale (>0.1 mol) synthesis of 2-normisoprostol, a 2-furyl carbinol with an esterified alkyl chain **173** was transformed into the 4,5-disubstituted 2-cyclopentenone **174** (Scheme 48). Although the ester function was hydrolyzed during the reaction and the diastereoselectivity was not established, the yield was essentially quantitative, which was the key feature required of the transformation.¹²⁵ The rearrangement of a series of representative furan substrates **175** was examined in water under microwave conditions (210 °C, 15 bar) without added catalyst: providing the substrate had some miscibility with water, reaction times of no more than 15 minutes were required for good conversions into the corresponding 2-cyclopentenones **176** with excellent *anti* diastereoselectivity.¹²⁶



Scheme 48

Furandiols 177 were converted into fused 2-cyclopentenones 178 using a catalytic amount of Lewis acid in aqueous glyme (Scheme 49).¹²⁷ The probable mechanism for this transformation was an acid-induced spiroketal formation, followed by hydrolysis to a 2-ene-1,4-dione. Intramolecular aldol condensation of this intermediate followed by intramolecular conjugate addition and finally dehydration furnished the bicyclic products.



Scheme 49

An aza-Piancatelli rearrangement was established, whereby the treatment of 2-furylcarbinols **179** with an aniline in the presence of dysprosium(III) triflate as a catalyst provided a mild and direct method for the preparation of 4arylamino-2-cyclopentenones **180** (Scheme 50). Only *trans* adducts were obtained and yields were mostly excellent, compromised only occasionally by a competing Friedel–Crafts alkylation of electron-rich anilines.¹²⁸ Recently, aza-Piancatelli rearrangements were described using as little as 0.03 mol% of phosphomolybdic acid as the catalyst.¹²⁹



Scheme 50

In related developments in this area, 2-furaldehyde **181** was treated with two equivalents of a secondary amine in the presence of a lanthanide(III) triflate to provide very high yields of the corresponding 4,5-diamino-2-cyclopentenones **182**, free of the more thermodynamically stable 2,4-isomers (Scheme 51). Here again, a concerted ringclosure step was invoked to explain the exclusively *trans* stereoselectivity.¹³⁰ Furans bearing donor–acceptor cyclopropane substituents **183** have also been used as substrates for the aza-Piancatelli rearrangement. The reaction proceeded with either primary or secondary anilines to provide the functionalized 2-cyclopentenones **184** in good yields. Excellent diastereoselectivity was also observed, except when an electron-poor aryl group was present on the cyclopropane substrate.¹³¹





An unprecedented rearrangement was reported for a polyfunctional tetrahydrofuran **185** which was transformed into cyclopentenone **186** simply upon standing for five days (Scheme 52). A mechanism involving an acidinduced series of intramolecular rearrangements proceeding via an acyclic pentadienal intermediate was proposed.¹³²





The mild amine-catalyzed rearrangement of pyranones **187** provided *trans*-4,5-dioxygenated 2-cyclopentenones **188**, a reaction which was best explained in terms of an electrocyclic ring opening followed by a conrotatory 4π -electron cyclization (Scheme 53).¹³³ A similar ring-contraction process had been observed previously for some C2 and C4 ulopyranosides.¹³⁴



Scheme 53

During the total syntheses of guanacaterenes A and E, a cyanohydrin lactone **189**, obtained as a mixture of four diastereoisomers in several steps from (+)-carvone, was treated with an excess of lithium hexamethyldisilazide to provide a single isomer of a 2-hydroxy-2-cyclopentenone **190** in acceptable yield (Scheme 54). The proposed mechanism involved intramolecular attack of the nitrile-stabilized anion on the lactone to form an epoxyalkoxide, which then rearranged with expulsion of cyanide to give a 1,2-diketocyclopentane which tautomerized to the most stable enone structure.¹³⁵



Scheme 54

A simple acid-mediated transformation of 3-methoxycarbonyl-2-methoxydihydropyrans **191** into the corresponding 3-methoxycarbonyl-2-cyclopentenones **192** was discovered (Scheme 55). The proposed mechanism, supported by an isotopic labelling study, implied the initial elimination of methanol to give a 2*H*-pyran which underwent electrocyclic 6π ring-opening to a dienal ester. Prins cyclization followed by elimination of water and isomerization gave the products in moderate yields.¹³⁶



Scheme 55

A novel aza-variant of the pyran-ring-contraction approach has been established for the reaction of 4,6-dialkylglycals **193** with aromatic amines in the presence of indium bromide, to give *trans*-2,5-dialkyl-4-arylamino-2cyclopentenones **194** in fair to good yields (Scheme 56). A mechanism was proposed, whereby a Ferrier-type reaction produced a cyclic aminal intermediate which underwent indium-mediated ring opening, dehydration, then conrotatory 4π -electron ring closure.¹³⁷



Scheme 56

Reactions of Sugar Lactones

Over the years, ribosides and ribonolactone derivatives have been popular precursors for the preparation of 4,5dioxygenated 2-cyclopentenones in single enantiomeric form. As part of a synthesis of carbon-substituted nucleosides, a Wittig reagent was added to the silvl enol ether 195 to generate an intermediate diketophosphorus ylide **196**, which cyclized to provide the highly substituted 2cyclopentenone 197 in reasonable yield (Scheme 57).¹³⁸ The enol ether 199 was prepared from a D-ribose derivative 198 by treatment with the Tebbe reagent; its treatment with dimethylaluminum chloride gave a good yield of the 2-cyclopentenone **200**, via a Lewis acid induced isopropyl group cleavage followed by an intramolecular aldol reaction. It had been hoped that the Tebbe reagent might induce the rearrangement and thus provide a cascade sequence towards some pentenomycin targets; in the event, dimethylaluminum chloride was the only Lewis acid able to induce the rearrangement, and even this failed with the corresponding compound lacking the hydroxymethyl group at C5.¹³⁹ An interesting tandem process was developed in which allylic substrates **201** were isomerized using photochemically activated iron(0) carbonyl to generate silyl ketal enol ethers **202**, which were treated with fluoride to induce ring opening followed by an intramolecular aldol reaction, to provide the 2-cyclopentenones **203**.¹⁴⁰





Ring Contractions

The photorearrangement of cyclohexa-2,5-dienones to bicyclo[3.1.0]hex-3-en-2-ones is a well-studied transformation, and was used to carry out formal syntheses of (\pm)methyleneomycins A and B (Scheme 58). The masked *p*benzoquinones **204** were irradiated to furnish 2-cyclopentenones **205** or **206** with a carboxylic ester in either the 4or 5-position, respectively. After the usual di- π -methane rearrangement from the excited state, the regioselectivity of the three-membered-ring bond fission of the bicyclic intermediate depends on the presence of a bridgehead substituent.¹⁴¹





Rearrangements of cyclohexa-2,4-dienones may also lead to 2-cyclopentenones. When solutions of selected masked *o*-benzoquinones **207** in chloroform were treated with singlet oxygen followed by thiourea (a reducing agent), the anticipated endoperoxides were unexpectedly accompanied by 4-hydroxy-2-cyclopentenones **208**; these latter products were the only ones obtained when the reactions were run in methanol (Scheme 59). This ring contraction was proposed to occur via initial formation of the perepoxide intermediate, which underwent an unusual [1,2]acyl shift in a highly stereoselective manner.¹⁴²





Ring Expansions

Cyclopropanes and cyclobutanes have considerable ring strain, so rearrangements involving expansion to a fivemembered ring are energetically favored. Some reviews on small-ring chemistry have covered formation of 2-cyclopentenones (amongst other compounds) via these routes.¹⁴³

A popular way of conducting a cyclobutanone ring expansion is to prepare the corresponding spiroepoxide, then rearrange it using a Lewis acid. A suitably placed leaving group is necessary for the extra unsaturation present in 2cyclopentenones (Scheme 60). After some optimization studies with a small series of 2-ethoxycyclobutanones, the chiral derivative 209 was treated with dimethylsulfoxonium methylide to generate a single spiroepoxide which was efficiently ring-expanded in a one-pot procedure using scandium triflate to give the cyclopentanone 210. Treatment with base was necessary to eliminate ethanol, and the resulting 2-cyclopentenone **211** was then used in the syntheses of (+)-carbovir and (+)-aristeromycin.¹⁴⁴ A selection of protected 2-aminocyclobutanones 212 were transformed into the corresponding spiroepoxides 213 using dimethylsulfonium methylide. Each of these was smoothly converted into the corresponding 2-cyclopentenone 214 upon treatment with lithium iodide, via ring expansion and spontaneous β -elimination of the amide fragment. Yields were generally high and the transformation proceeded with no loss of the stereochemical information derived from the cyclobutanone substrate.145

An interesting regioselectivity issue was uncovered during studies on the ring expansion of 2-hydroxycyclobutanone aldol adducts **215**, using acidic ion-exchange resin (Scheme 61). With no other ring substituents, the skeletal rearrangement gave the 3-hydroxy-2-cyclopentenone **216**, which is a tautomer of the cyclic 1,3-diketone. On the other hand, with the corresponding 3,3-dimethylated cyclobutanones, the acid-mediated rearrangement gave the 2-hydroxy-2-cyclopentenone regioisomers **217**.¹⁴⁶ These reaction profiles were interpreted in terms of the relative migratory aptitudes of the acyl and alkyl groups involved.









2,2-Dichlorocyclobutanes react with diazomethane via a regioselective ring-expansion followed by dehydrochlorination to give the 2-chloro-2-cyclopentenone framework. This reactivity was exploited to transform the dichloroketene adduct of 7-methylcycloheptatriene **218** into a single hydroazulenone **219** in 76% yield (Scheme 62).¹⁴⁷ This compound was a key intermediate in the total synthesis of (±)-6-deoxygeigerin and, later, of (±)-geigerin.¹⁴⁸





The unique chemistry of squaric acid and its derivatives has been exploited for selective ring expansions to give 2cyclopentenones (Scheme 63). Double addition of vinyl magnesium bromide to the derivatives **220** proceeded in a 1,2-/1,4-fashion to give linear octatetraene species **221**, which upon selective protonation during work-up underwent cyclization to give the polyfunctional derivatives **222** in good yield in a one-pot operation.¹⁴⁹





A 'squarate ester cascade' protocol has been used to construct angular or linear triguinane sequiterpene skeletons in an elegant fashion (Scheme 64). Sequential addition of two alkenyllithium species to diisopropyl squarate 223 provided, irrespective of the geometry of the dienolate adduct, an intermediate which evolved first via eightmembered-ring closure to dienolate 224 then further via intramolecular aldol closure to give the linear tricyclic structure 225 in 24% yield (after acidic hydrolysis of the enol).¹⁵⁰ On the other hand, the successive addition of an alkenyllithium then an alkynyllithium to 223 proceeded via a helical dienolate which ring-closed via a strained intermediate structure 226; the latter then evolved via intramolecular enolate attack to close the fused five-membered ring of 227, in 68% yield. These ring-assembly protocols were exploited for the total synthesis of sesquiterpenes.



Scheme 64

1-Vinyl-1-silyloxy-2,2-dichlorocyclopropanes **228**, easily prepared from silyloxydienes, were treated with a silver(I) reagent to induce a sequential 2π ring opening and 4π -electron ring closure, the latter step being analogous to the Nazarov cyclization (Scheme 65). This process provided chloro-substituted 2-cyclopentenones **229** and **230** in variable proportions and reasonable yields, albeit with limited control of regio- and diastereoselectivities.¹⁵¹



Scheme 65

The ring expansion of alkynylcyclopropanols **231** to provide 3-substituted 2-cyclopentenones **232** was achieved using a ruthenium catalyst in the presence of indium triflate and a Brønsted acid (Scheme 66). The reaction was successful with alkyl-substituted alkynes, for which an insertion mechanism via a ruthenium metallacycle was sug-

19

gested. With more electron-deficient alkynes, notably acyl derivatives, the reaction evolved differently to give 2-vinylcyclobutanones.¹⁵²



Scheme 66

A ring expansion of highly substituted epoxides bearing acetate and propargylic ester functions **233** was achieved using a platinum catalyst (Scheme 67). Even though *syn/anti* mixtures of the substrates were employed, the product 2-cyclopentenones **234** were obtained as single diastereoisomers. The proposed mechanism involved evolution of a metal–alkyne complex with acetate migration to form a platinum carbene, which reacted with the epoxide to form a pyran **235**. An unprecedented tandem oxa- 6π ring opening followed by platinum-mediated conrotatory 4π -electron ring closure accompanied by acetate transposition was proposed to explain the transformation of the pyran to give the final product **234**.¹⁵³ This reaction is clearly related to the Nazarov and Rautenstrauch cyclization reactions (*vide supra*).





The treatment of the vicinal diol **236** with strong base provided the 2-cyclopentenone **237**, reportedly in good yield (Scheme 68).¹⁵⁴ The transformation is noteworthy as the first (and so far only) apparent example of an anionic oxy retro-ene reaction, giving in this case a diketone enolate which evolved by intramolecular aldol condensation.



Scheme 68

Finally, 1,1-disubstituted spiropentanes **238** were converted into 2-cyclopentenones **239** in the presence of carbon monoxide and a rhodium catalyst (Scheme 69). The proposed mechanism involved evolution through a series of rhodacycles of increasing size, so that one of the original spiropentane carbons ended up as a methyl substituent in an exocyclic locus in the final structure.¹⁵⁵ Two examples of cyclic-fused 1,2-disubstituted spiropentanes also evolved in a similar fashion with retention of the ring junction stereochemistry, albeit in slightly lower yield.





5 Miscellaneous Methods

A few reports of 2-cyclopentenone preparation cannot be conveniently classed in any of the above sections and are therefore collected in this final section.

The retro-Diels-Alder strategy for the preparation of substituted 2-cyclopentenones from the tetrahydro-4,7-methano-1-indanone skeleton was established some time ago, but the cracking conditions required to liberate the target structures can be somewhat restrictive. Nevertheless, an attractive feature of this approach is the relative ease of access to the starting materials. The approach therefore continues to enjoy some synthetic applications (Scheme 70). The preparation of the tri-substituted 2-cyclopentenone 241, a key building block for a proposed total synthesis of the diterpene antibiotic guanacasterpene A, was achieved via a flash vacuum pyrolysis (FVP) retro-Diels-Alder reaction of **240** in excellent yield.¹⁵⁶ In the synthesis of selected phytoprostanes for evaluation as PPAR-y ligands, a retro-Diels-Alder reaction was performed on tricyclic adduct 242 under mild conditions employing a Lewis acid and an excess of maleic anhydride to trap the cyclopentadiene by-product. The E-isomer of the 2-cyclopentenone product 243 predominated, regardless of the configuration (Z or E) of the substrate.¹⁵⁷ Similarly, the use of zeolites serving as heterogeneous Brønsted acids facilitated the transformation of a series of tricyclic substrates 244 into 4-alkyl-2-cyclopentenones 245 under mild conditions.158

The tandem [2+2]-cycloaddition–Dieckmann condensation of γ -keto esters **246** with lithium ynolates gave, in the first instance, bicyclic β -lactones **247**, which readily eliminated carbon dioxide upon heating to provide 2,3-disubstituted 2-cyclopentenones **248** in a one-pot process (Scheme 71).¹⁵⁹ An application of this protocol was made





in the final step of a total synthesis of the triquinane natural product cucumin E (250) by transformation of the bicyclic precursor 249.¹⁶⁰





In a study of synthetic routes to prostanoids, an unusual tandem reaction process was discovered when selected aldehydes **251** were transformed into the pyrrolidine enamines and heated with 3-iodo-2-methoxymethoxy-1propene. 4,4-Disubstituted 2-cyclopentenones **252** were formed directly, albeit in moderate yields, in what appeared to be an original domino aza-Claisen–Mannich cyclization, terminating in β -elimination of the secondary amine (Scheme 72). In several cases, the aldehyde substrate was a chiral sugar derivative but the spiro adducts were obtained without significant diastereomeric excesses.¹⁶¹





4-Alkylidene-2-cyclopentenones **254** were prepared from unsaturated 1,4-diketones **253** and selected nitroalkanes in a mild tandem ring-closure–Michael addition–elimination process, with good yields and excellent *E*-stereose-lectivity (Scheme 73).¹⁶² This is a surprisingly simple yet attractive route to the target compounds which can otherwise be difficult to prepare.



Scheme 73

An intriguing cascade reaction combining benzil (255) and two equivalents of an alkynyllithium has been described, in which it was proposed that the initial 1,5-hexadiyn-3,4-olate adduct evolved via an anionic oxy-Cope rearrangement then an intramolecular aldol condensation and a [1,2]-aryl shift, to give 2-phenyl-5-benzoyl-2-cyclopentenones 256 (Scheme 74). The reaction was substrate-dependent, for competing reaction pathways were evidenced, but several examples employing an aryl or alkyl acetylide proceeded in high yield. The 5-acyl-2-cyclopentenone products exhibited the expected keto–enol tautomeric equilibria.¹⁶³

Finally, an unprecedented cyclotrimerization of α -monosubstituted aldehydes induced by dibromotriphenylphosphine has been described (Scheme 75).¹⁶⁴ The proposed mechanism consisted of a series of Lewis acid mediated aldol condensations going via a propenal to give a pentadienal, which isomerized before cyclizing via a Nazarovtype process to give the 2,3,5-trisubstituted 2-cyclopentenone products **257**. On this postulate, mixed condensations of aldehydes and enals were carried out using the same conditions to give the expected products **258**.



Scheme 75

6 Conclusions

As this review shows, an ever-expanding array of synthetic methodologies is available for the synthesis of 2-cyclopentenones. Established reactions continue to be further developed while new approaches continue to emerge. Some developments emphasize the use of catalytic processes or benign reaction conditions, while others prioritize selectivity issues or other practicalities. A significant generality to emerge from this overview is that there is no one reaction which dominates the scene: the 'best' choice of synthesis for any given target will depend on a balance of many factors. Given the enduring importance of the title compounds and the efforts currently geared towards their synthesis, it seems likely that methodologies will continue to develop at a considerable pace.

Acknowledgment

One of us (H.E.) was the beneficiary of a PhD grant (French MESR Allocation) earmarked by Université Paris-Sud for bilateral international collaboration. The authors are grateful to the Ile-de-France region for travel funding (SETCI exchange programme).



Scheme 74

© Georg Thieme Verlag Stuttgart · New York

References

- (a) Gibson, S. E.; Lewis, S. E.; Mainolfi, N. J. Organomet. Chem. 2004, 689, 3873. (b) Roche, S. P.; Aitken, D. J. Eur. J. Org. Chem. 2010, 5339. (c) Mehta, G.; Srikrishna, A. Chem. Rev. 1997, 97, 671. (d) Hudlicky, T.; Price, J. D. Chem. Rev. 1989, 89, 1467. (e) Trost, B. M. Chem. Soc. Rev. 1982, 11, 141. (f) Marsden, S. P. Science of Synthesis; Vol. 26; Cossy, J., Ed.; Thieme: Stuttgart, 2004, 1045.
- (2) Ellison, R. A. Synthesis 1973, 397.
- (3) Coote, S. C.; O'Brien, P.; Whitwood, A. C. Org. Biomol. Chem. 2008, 6, 4299.
- (4) Basavaiah, D.; Reddy, B. S.; Badsara, S. S. Chem. Rev. 2010, 110, 5447.
- (5) Declerck, V.; Martinez, J.; Lamaty, F. Chem. Rev. 2009, 109, 1.
- (6) Koech, P. K.; Krische, M. J. J. Am. Chem. Soc. 2004, 126, 5350.
- (7) Krafft, M. E.; Cran, J. W. Synlett 2005, 1263.
- (8) (a) Smith, A. B. III; Branca, S. J.; Guaciaro, M. A.; Wovkulich, P. M.; Korn, A. Org. Synth. 1983, 61, 65.
 (b) Kim, K. M.; Chung, K. H.; Kim, J. N.; Ryu, E. K. Synthesis 1993, 283. (c) Adam, W.; Saha-Moeller, C. R.; Zhao, C.-G. Org. React. 2002, 61, 219.
- (9) (a) Negishi, E.-i. *Tetrahedron* 2000, *56*, 10197. (b) Miller, M. W.; Johnson, C. R. *J. Org. Chem.* 1997, *62*, 1582.
 (c) Dyker, G.; Markwitz, H.; Henkel, G. *Eur. J. Org. Chem.* 2001, 2415. (d) Molander, G. A.; Ham, J.; Seapy, D. G. *Tetrahedron* 2007, *63*, 768.
- (10) Liu, K.-M.; Chau, C.-M.; Sha, C.-K. Chem. Commun. 2008, 91.
- (11) Cho, D. J.; Wu, C. J.; S, S.; Han, W.-S.; Kang, S. O.; Lee, B. Y. Organometallics 2006, 25, 2133.
- (12) (a) Morisaki, Y.; Imoto, H.; Hirano, K.; Hayashi, T.; Chujo, Y. J. Org. Chem. 2011, 76, 1795. (b) Thaler, T.; Guo, L.-N.; Steib, A. K.; Raducan, M.; Karaghiosoff, K.; Mayer, P.; Knochel, P. Org. Lett. 2011, 13, 3182. (c) Wei, C.-H.; Mannathan, S.; Cheng, C.-H. J. Am. Chem. Soc. 2011, 133, 6942. (d) Piovesana, S.; Scarpino Schietroma, D. M.; Tulli, L. G.; Monaco, M. R.; Bella, M. Chem. Commun. 2010, 46, 5160. (e) Sibi, M. P.; Manyem, S. Tetrahedron 2000, 56, 8033.
- (13) (a) Perdicchia, D.; Jørgensen, K. A. J. Org. Chem. 2007, 72, 3565. (b) Yang, L.; Xu, L.-W.; Zhou, W.; Li, L.; Xia, C.-G. *Tetrahedron Lett.* 2006, 47, 7723. (c) Rana, N. K.; Selvakumar, S.; Singh, V. K. J. Org. Chem. 2010, 75, 2029.
- (14) (a) Krishna, T. R.; Jayaraman, N. *Tetrahedron* 2004, 60, 10325. (b) Li, J.-H.; Wang, D.-P.; Xie, Y.-X. *Tetrahedron Lett.* 2005, 46, 4941.
- (15) Guo, H.-C.; Ma, J.-A. Angew. Chem. Int. Ed. 2006, 45, 354.
- (16) Gerdil, R.; Liu, H.; Bernardinelli, G. *Helv. Chim. Acta* **1999**, 82, 418.
- (17) (a) Kelly, J. M.; Leeper, F. J. *Tetrahedron Lett.* 2012, *53*, 819. (b) Tolstikov, G. A.; Miftakhov, M. S.; Danilova, N. A.; Vel'der, Ya. L.; Spirikhin, L. V. *Synthesis* 1989, 625.
- (18) (a) Demir, A. S.; Enders, D. *Tetrahedron Lett.* **1989**, *30*, 1705. (b) Mukaiyama, T.; Ohsumi, T. *Chem. Lett.* **1983**, 875. (c) Lakhvich, F. A.; Lis, L. G.; Pap, A. A.; Rubinov, D. B.; Zheldakova, T. A.; Akhrem, A. A. *Zh. Org. Khim.* **1988**, *24*, 518.
- (19) (a) Schulé, A.; Liang, H.; Vors, J.-P.; Ciufolini, M. A. J. Org. Chem. 2009, 74, 1588. (b) Altenbach, R. J.; Brune, M. E.; Buckner, S. A.; Coghlan, M. J.; Daza, A. V.; Fabiyi, A.; Gopalakrishnan, M.; Henry, R. F.; Khilevich, A.; Kort, M. E.; Milicic, I.; Scott, V. E.; Smith, J. C.; Whiteaker, K. L.; Carroll, W. A. J. Med. Chem. 2006, 49, 6869.
- (20) (a) Kobayashi, Y.; Murugesh, M. G.; Nakano, M.; Takahisa, E.; Usmani, S. B.; Ainai, T. J. Org. Chem. 2002, 67, 7110.

(b) Weaving, R.; Roulland, E.; Monneret, C.; Florent, J.-C. *Tetrahedron Lett.* **2003**, *44*, 2579. (c) Shi, M.; Zhang, W. *Tetrahedron* **2005**, *61*, 11887.

- (21) (a) Mizuno, M.; Inoue, H.; Naito, T.; Zhou, L.; Nishiyama, H. *Chem. Eur. J.* 2009, *15*, 8985. (b) Das, J.; Le Cavelier, F.; Rouden, J.; Blanchet, J. *Eur. J. Org. Chem.* 2011, 6628.
- (22) (a) Pete, J.-P. In *CRC Handbook of Organic Photochemistry* and Photobiology, 2nd ed.; Horspool, W.; Lenci, F., Eds.; CRC Press: Boca Raton, 2003, Chap. 71, 1–14. (b) Le Liepvre, M.; Ollivier, J.; Aitken, D. J. *Eur. J. Org. Chem.* 2009, 5953. (c) Margaretha, P. In *Synthetic Organic Photochemistry*; Griesbeck, A. G.; Mattay, J., Eds.; Marcel Dekker: New York, 2005, Chap. 8, 211–237.
- (23) (a) Xu, H.; Wolf, C. Angew. Chem. Int. Ed. 2011, 50, 12249.
 (b) Canales, E.; Corey, E. J. Org. Lett. 2008, 10, 3271.
 (c) Northrup, A. B.; MacMillan, D. W. C. J. Am. Chem. Soc. 2002, 124, 2458.
- (24) (a) Henry, S. S.; Brady, M. D.; Laird, D. L. T.; Ruble, J. C.; Varie, D. L.; Monn, J. A. *Org. Lett.* 2012, *14*, 2662.
 (b) Garcia Ruano, J. L.; Alonso, M.; Cruz, D.; Fraile, A.; Martin, M. R.; Peromingo, M. T.; Tito, A.; Yuste, F. *Tetrahedron* 2008, *64*, 10546.
- (25) Menjo, Y.; Hamajima, A.; Sasaki, N.; Hamada, Y. Org. Lett. 2011, 13, 5744.
- (26) Lee, A.; Reisinger, C. M.; List, B. Adv. Synth. Catal. 2012, 354, 1701.
- (27) (a) Schore, N. E. Org. React. 1991, 40, 1. (b) Geis, O.; Schmalz, H.-G. Synthesis 1998, 911. (c) Brummond, K. M.; Kent, J. L. Tetrahedron 2000, 56, 3263. (d) Laschat, S.; Becheanu, A.; Bell, T.; Baro, A. Synlett 2005, 2547. (e) Lee, H.-W.; Kwong, F.-Y. Eur. J. Org. Chem. 2010, 789. (f) Blanco-Urgoiti, J.; Añorbe, L.; Pérez-Serrano, L.; Domínguez, G.; Pérez-Castells, J. Chem. Soc. Rev. 2004, 33, 32.
- (28) Gibson, S. E.; Mainolfi, N. Angew. Chem. Int. Ed. 2005, 44, 3022.
- (29) Shibata, T. Adv. Synth. Catal. 2006, 348, 2328.
- (30) *The Pauson-Khand Reaction: Scope Variations and Applications*; Torres, R. R., Ed.; Wiley: Chichester, **2012**.
- (31) For leading references, see: Wu, N.; Deng, L.; Liu, L.; Liu, Q.; Li, C.; Yang, Z. Chem. Asian J. 2013, 8, 65.
- (32) Morimoto, T.; Kakiuchi, K. Angew. Chem. Int. Ed. 2004, 43, 5580.
- (33) Hong, B. C.; Dange, N. S.; Yen, P.-J.; Lee, G.-H.; Liao, J.-H. Org. Lett. 2012, 14, 5346.
- (34) Su, S.; Rodriguez, E. A.; Baran, P. S. J. Am. Chem. Soc. 2011, 133, 13922.
- (35) Tap, A.; Jouanneau, M.; Galvani, G.; Sorin, G.; Lannou, M.-I.; Ferezou, J.-P.; Ardisson, J. Org. Biomol. Chem. 2012, 10, 8140.
- (36) Turlington, M.; Du, Y.; Oqtrum, S. G.; Santosh, V.; Wren, K.; Lin, T.; Sabat, M.; Pu, L. J. Am. Chem. Soc. 2011, 133, 11780.
- (37) Morisaki, Y.; Kondo, T.; Mitsudo, T.-A. Org. Lett. 2000, 2, 949.
- (38) Huang, Q.; Hua, R. Chem. Eur. J. 2009, 13, 3817.
- (39) Ohashi, M.; Taniguchi, T.; Ogoshi, S. J. Am. Chem. Soc. 2011, 133, 14900.
- Barluenga, J.; Barrio, P.; Riesgo, L.; López, L. A.; Tomás, M. J. Am. Chem. Soc. 2007, 129, 14422.
- (41) Barluenga, J.; Álvarez-Fernández, A.; Suárez-Sobrino, Á. L.; Tomás, M. Angew. Chem. Int. Ed. 2012, 51, 183.
- (42) Qi, X.; Ready, J. M. Angew. Chem. Int. Ed. 2008, 47, 7068.
- (43) Miller, J. A.; Pugh, A. W.; Ullah, G. M.; Welsh, G. M. *Tetrahedron Lett.* **2001**, *42*, 955.
- (44) Tius, M. A. Acc. Chem. Res. 2003, 36, 284.

23

- (45) Hu, H.; Smith, D.; Cramer, R. E.; Tius, M. A. J. Am. Chem. Soc. 1999, 121, 9895.
- (46) Schultz-Fademrecht, C.; Tius, M. A.; Grimme, S.;
 Wibbeling, B.; Hoppe, D. *Angew. Chem. Int. Ed.* 2002, *41*, 1532.
- (47) Jacobsen, C. B.; Jensen, K. L.; Udmark, J.; Jørgensen, K. A. Org. Lett. 2011, 13, 4790.
- (48) Feist, H.; Langer, P. Synthesis 2008, 3877.
- (49) Langer, P.; Kracke, B. Synlett 2001, 1790.
- (50) Shinohara, Y.; Kurata, T.; Kitano, H.; Matsumoto, K.; Takahashi, I.; Hosoi, S.; Ota, T.; Hatanaka, M. Synlett 2002, 1245.
- (51) (a) Cai, X.-h.; Xie, B. *Res. J. Chem. Sci.* 2011, *1*, 120.
 (b) Marjani, K.; Mohsen, M.; Arazi, A.; Ashouri, A.; Bourghani, S.; Rajabi, M. *Monatsh. Chem.* 2009, *140*, 1331.
 (c) Marjani, K.; Mousavi, M.; Ashouri, A.; Arazi, O.; Bourghani, S.; Asgari, M. *J. Chem. Res.* 2008, 398.
 (d) Majani, K.; Asgari, M.; Ashouri, A.; Mahdavinia, G. H.; Ahangar, H. A. *Chin. Chem. Lett.* 2009, *20*, 401.
- (52) Holtz, E.; Köhler, V.; Appel, B.; Langer, P. Eur. J. Org. Chem. 2005, 532.
- (53) Jose, A.; Lakshmi, K. C. S.; Suresh, E.; Nair, V. Org. Lett. 2013, 15, 1858.
- (54) Gagnier, S. V.; Larock, R. C. J. Am. Chem. Soc. 2003, 125, 4804.
- (55) Li, H.; Liu, L.; Zhao, F.; Wang, C.; Wang, C.; Song, Q.; Zhang, W.-X.; Xi, Z. J. Org. Chem. 2012, 77, 4793.
- (56) Laroche, C.; Bertus, P.; Szymoniak, J. Chem. Commun. 2005, 3030.
- (57) Rigby, J. H.; Wang, Z. Org. Lett. 2003, 5, 263.
- (58) Davie, C. P.; Danheiser, R. L. Angew. Chem. Int. Ed. 2005, 44, 5867.
- (59) Moser, W. H.; Feltes, L. A.; Sun, L.; Giese, M.; Farrell, R. W. J. Org. Chem. 2006, 71, 6542.
- (60) Giese, M.; Moser, W. H. Org. Lett. 2008, 10, 4215.
- (61) Li, Z.; Moser, W. H.; Deng, R.; Sun, L. J. Org. Chem. 2007, 72, 10254.
- (62) Kurahashi, T.; Wu, Y.-T.; Meindl, K.; Rühl, S.; de Meijere, A. Synlett 2005, 805.
- (63) Kurteva, V. B.; Afonso, C. A. M. Chem. Rev. 2009, 109, 6809.
- (64) Nazarov, I. N.; Zaretskaya, I. I. *Izv. Akad. Nauk. SSSR, Ser. Khim.* **1944**, 65.
- (65) (a) Rueping, M.; Ieawsuwan, W.; Antonchick, A. P.; Nachtsheim, B. J. *Angew. Chem. Int. Ed.* 2007, *46*, 2097.
 (b) Basak, A. K.; Shimada, N.; Bow, W. F.; Vicic, D. A.; Tius, M. A. *J. Am. Chem. Soc.* 2010, *132*, 8266. (c) Bow, W. E.; Basak, A. K.; Jolit, A.; Vicic, D. A.; Tius, M. A. Org. Lett. 2010, *12*, 440. (d) Rueping, M.; Ieawsuwan, W. Chem. Commun. 2010, 11450.
- (66) (a) Bee, C.; Leclerc, E.; Tius, M. A. Org. Lett. 2003, 5, 4927.
 (b) Aggarwal, V. K.; Belfield, A. J. Org. Lett. 2003, 5, 5075.
 (c) Liang, G.; Trauner, D. J. Am. Chem. Soc. 2004, 126, 9544. (d) Janka, M.; He, W.; Haedicke, I. E.; Fronczek, F. R.; Frontier, A. J.; Eisenberg, R. J. Am. Chem. Soc. 2006, 128, 5312. (e) Cao, P.; Deng, C.; Zhou, Y.-Y.; Sun, X.-L.; Zheng, J.-C.; Xie, Z.; Tang, Y. Angew. Chem. Int. Ed. 2010, 49, 4463.
- (67) (a) Pellissier, H. *Tetrahedron* 2005, *61*, 6479. (b) Frontier, A. J.; Collinson, C. *Tetrahedron* 2005, *61*, 7577. (c) Tius, M. A. *Eur. J. Org. Chem.* 2005, 2193. (d) Habermas, K. L.; Denmark, S. E. *Org. React.* 1994, *45*, 1. (e) Denmark, S. E. In *Comprehensive Organic Synthesis*; Vol. 5; Paquette, L. A., Ed.; Pergamon: Oxford, 1991, 751–784. (f) Santelli-Rouvier, C.; Santelli, M. *Synthesis* 1983, 429.
- (68) Vaida, T.; Eisenberg, R.; Frontier, A. J. ChemCatChem 2011, 3, 1531.

- (69) Shimada, N.; Stewart, C.; Tius, M. A. Tetrahedron 2011, 67, 5851.
- (70) Spencer, W. T. III; Vaidya, T.; Frontier, A. J. Eur. J. Org. Chem. 2013, 3621.
- (71) Grant, T. N.; Rieder, C. J.; West, F. G. Chem. Commun. 2009, 5676.
- (72) Yaji, K.; Shindo, M. Tetrahedron 2010, 66, 9808.
- (73) Williams, D. R.; Robinson, L. A.; Nevill, C. R.; Reddy, J. P. Angew. Chem. Int. Ed. 2007, 46, 915.
- (74) Magnus, P.; Freund, W. A.; Moorhead, E. J.; Rainey, T. J. Am. Chem. Soc. 2012, 134, 6140.
- (75) Kerr, D. J.; Flynn, B. L. Org. Lett. 2012, 14, 1740.
- (76) Lebœuf, D.; Wright, C. M.; Frontier, A. J. Chem. Eur. J. 2013, 19, 4835.
- (77) Kerr, D. J.; Miletic, M.; Chaplin, J. H.; White, J. M.; Flynn, B. L. Org. Lett. 2012, 14, 1732.
- (78) Flynn, B. L.; Manchala, N.; Krenske, E. H. J. Am. Chem. Soc. 2013, 135, 9156.
- (79) Cao, P.; Sun, X.-L.; Zhu, B.-H.; Shen, Q.; Xie, Z.; Tang, Y. Org. Lett. 2009, 11, 3048.
- (80) Li, W.-D. Z.; Duo, W.-G.; Zhuang, C.-H. Org. Lett. 2011, 13, 3538.
- (81) Spencer, W. T. III; Levin, M. D.; Frontier, A. J. Org. Lett. 2011, 13, 414.
- (82) Rautenstrauch, V. J. Org. Chem. 1984, 49, 950.
- (83) Nakagawa, D.; Miyashita, M.; Tanino, K. *Tetrahedron Lett.* 2010, *51*, 2771.
- (84) Shi, X.; Gorin, D. J.; Toste, F. D. J. Am. Chem. Soc. 2005, 127, 5802.
- (85) Kato, K.; Kobayashi, T.; Fujimami, T.; Motodate, S.; Kusakabe, T.; Mochida, T.; Akita, H. Synlett 2008, 1081.
- (86) Zhang, L.; Wang, S. J. Am. Chem. Soc. 2006, 128, 1442.
- (87) An, S. E.; Jeong, J.; Baskar, B.; Lee, J.; Seo, J.; Rhee, Y. H. *Chem. Eur. J.* **2009**, *15*, 11837.
- (88) Oh, C. H.; Karmakar, S. J. Org. Chem. 2009, 74, 370.
- (89) Tang, J.-M.; Liu, T.-A.; Liu, R.-S. J. Org. Chem. 2008, 73, 8479.
- (90) (a) Prunet, J. Eur. J. Org. Chem. 2011, 3634. (b) Metathesis in Natural Product Synthesis; Cossy, J.; Arseniyadis, S.; Meyer, C., Eds.; Wiley-VCH: Weinheim, 2010.
 (c) Monfette, S.; Fogg, D. E. Chem. Rev. 2009, 109, 3783.
 (d) Majumdar, K. C.; Muhuri, S.; Islam, R.; Chattopadhyay, B. Heterocycles 2009, 78, 1109. (e) Chattopadhyay, S. K.; Karmakar, S.; Biswas, T.; Majumdar, K. C.; Rahaman, H.; Roy, B. Tetrahedron 2007, 63, 3919. (f) Villar, H.; Frings, M.; Bolm, C. Chem. Soc. Rev. 2007, 36, 55. (g) Brown, R. C. D.; Satcharoen, V. Heterocycles 2006, 70, 705.
 (h) Schmidt, B.; Hermanns, J. Curr. Org. Chem. 2006, 10, 1363. (i) Deiters, A.; Martin, S. F. Chem. Rev. 2004, 104, 2199. (j) Han, S. Y.; Chang, S. In Handbook of Metathesis; Vol. 2; Grubbs, R. H., Ed.; Wiley-VCH: Weinheim, 2010, Chap. 2, 5–127.
- (91) Funel, J.-A.; Prunet, J. J. Org. Chem. 2004, 69, 4555.
- (92) Dübon, P.; Schelwies, M.; Helmchen, G. Chem. Eur. J. 2008, 14, 6722.
- (93) Miller, A. K.; Hughes, C. C.; Kennedy-Smith, J. J.; Gradl, S. N.; Trauner, D. J. Am. Chem. Soc. 2006, 128, 17057.
- (94) (a) Davis, F. A.; Wu, Y. *Org. Lett.* 2004, *6*, 1269.
 (b) Bullock, K.; Chong, P.; Davies, R.; Elitzin, V.; Hatcher, M.; Jackson, M.; Liu, B.; Patterson, D.; Powers, J.; Salmons, M.; Tabet, E.; Toczko, M. *Top. Catal.* 2012, *55*, 446.
- (95) Harmata, M.; Wacharasindhu, S. Org. Lett. 2005, 7, 2563.
- (96) Pfeiffer, M. W. B.; Phillips, A. J. J. Am. Chem. Soc. 2005, 127, 5334.
- (97) Toueg, J.; Prunet, J. Synlett 2006, 2807.
- (98) Diver, S. T.; Guissert, A. J. Chem. Rev. 2004, 104, 1317.

This document was downloaded for personal use only. Unauthorized distribution is strictly prohibited.

- (99) Ray, D.; Paul, S.; Brahma, S.; Ray, J. K. Tetrahedron Lett. 2007, 48, 8005.
- (100) Ray, D.; Ray, J. K. Org. Lett. 2007, 9, 191.
- (101) Ray, D.; Mal, S. K.; Ray, J. K. Synlett 2005, 2135.
- (102) Morita, A.; Kuwahara, S. Org. Lett. 2006, 8, 1613.
- (103) Tanaka, K.; Fu, G. C. J. Am. Chem. Soc. 2001, 123, 11492.
- (104) (a) Tanaka, K.; Fu, G. C. J. Am. Chem. Soc. 2001, 124, 10296. (b) Tanaka, K.; Fu, G. C. J. Am. Chem. Soc. 2003, 125, 9078.
- (105) Miller, A. K.; Banghart, M. R.; Beaudry, C. M.; Suh, J. M.; Trauner, D. *Tetrahedron* **2003**, *59*, 8919.
- (106) Kuroda, C.; Honda, S.; Nagura, Y.; Koshio, H.; Shibue, T.; Takeshita, T. *Tetrahedron* **2004**, *60*, 319.
- (107) Lo, C.-Y.; Lin, C.-C.; Cheng, H.-M.; Liu, R.-S. Org. Lett. **2006**, *8*, 3135.
- (108) Srikrishna, A.; Kumar, P. P. Tetrahedron 2000, 56, 8189.
- (109) Pearson, A. J.; Kim, E. H.; Sun, H. *Tetrahedron* **2010**, *66*, 4943.
- (110) Mehta, G.; Shinde, H. M. Tetrahedron Lett. 2007, 48, 8297.
- (111) Andrei, M.; Undheim, K. *Tetrahedron: Asymmetry* **2004**, *15*, 53.
- (112) Samanta, S.; Yasmin, N.; Kundu, D.; Ray, J. K. Tetrahedron Lett. 2010, 51, 4132.
- (113) Kitamura, M.; Chiba, S.; Narasaka, K. *Chem. Lett.* **2004**, *33*, 942.
- (114) Al-Tel, T. H.; Semreen, M. H.; Voelter, W. Org. Biomol. Chem. 2010, 8, 5375.
- (115) Christoffers, J.; Werner, T.; Frey, W.; Baro, A. Chem. Eur. J. **2004**, 10, 1042.
- (116) Li, W.-D. Z.; Wang, Y.-Q. Org. Lett. 2003, 5, 2931.
- (117) Cho, Y. S.; Carcache, D. A.; Tian, Y.; Li, Y.-M.; Danishefsky, S. J. J. Am. Chem. Soc. 2004, 126, 14358.
 (119) Van D. Scilling, C. D. L. C., Charles and Computer Sciences and Computer Scie
- (118) Yan, B.; Spilling, C. D. J. Org. Chem. 2008, 73, 5385.
- (119) Xu, Y.; McLaughlin, M.; Chen, C.-y.; Reamer, R. A.; Dormer, P. G.; Davies, I. W. J. Org. Chem. 2009, 74, 5100.
 (120) Deng, G.; Xu, B.; Wang, J. Tetrahedron 2005, 61, 10811.
- (121) Zhuo, X.; Xiang, K.; Zhang, F.-M.; Tu, Y.-Q. J. Org. Chem. 2011, 76, 6918.
- (122) Piancatelli, G.; D'Auria, M.; D'Onofrio, F. Synthesis 1994, 867.
- (123) Faza, O. N.; López, C. S.; Álvarez, R.; de Lera, Á. R. Chem. Eur. J. 2004, 10, 4324.
- (124) (a) Csákÿ, A. G.; Contreras, C.; Mba, M.; Plumet, J. Synlett
 2002, 1451. (b) Csákÿ, A. G.; Mba, M.; Plumet, J. Synlett
 2003, 2092.
- (125) Harikrishna, M.; Mohan, H. R.; Dubey, P. K.; Subbaraju, G. V. Synth. Commun. 2009, 39, 2763.
- (126) Ulbrich, K.; Kreitmeier, P.; Reiser, O. Synlett 2010, 2037.
- (127) Yin, B.-L.; Wu, Y.-L.; Lai, J.-Q. Eur. J. Org. Chem. 2009, 2695.
- (128) Veits, G. K.; Wenz, D. R.; Read de Alaniz, J. Angew. Chem. Int. Ed. **2010**, 49, 9484.
- (129) Reddy, B. V. S.; Narasimhulu, G.; Lakshumma, P. S.; Reddy, Y. V.; Yadav, J. S. *Tetrahedron Lett.* **2012**, *53*, 1776.
- (130) Li, S.-W.; Batey, R. A. Chem. Commun. 2007, 3759.
- (131) Wenz, D. R.; Read de Alaniz, J. Org. Lett. 2013, 15, 3250.
- (132) Jung, M. E.; Yoo, D. J. Org. Chem. 2007, 72, 8565.
- (133) (a) Nunes, J. P. M.; Veiros, L. F.; Vaz, P. D.; Afonso, C. A. M.; Caddick, S. *Tetrahedron* 2011, 67, 2779. (b) Caddick, S.; Cheung, S.; Doyle, V. E.; Frost, L. M.; Soscia, M. G.; Delisser, V. M.; Williams, M. R. V.; Etheridge, Z. E. Khan S.; Hitchcock, P. B.; Pairaudeau, G.; Vile, S. *Tetrahedron* 2001, 57, 6295.

- (134) (a) Zou, W.; Shao, H.; Wu, S.-H. Carbohydr. Res. 2004, 339, 2475. (b) Zou, W.; Wang, Z.; Lacroix, E.; Wu, S.-H.; Jennings, H. J. Carbohydr. Res. 2001, 334, 223.
- (135) Shipe, W.; Sorensen, E. J. J. Am. Chem. Soc. 2006, 128, 7025.
- (136) Matoušová, E.; Růžička, A.; Kuneš, J.; Králová, J.; Pour, M. *Chem. Commun.* **2011**, *47*, 9390.
- (137) Li, F.; Ding, C.; Wang, M.; Yao, Q.; Zhang, A. J. Org. Chem. 2011, 76, 2820.
- (138) Pryde, D. C.; Middleton, D. S.; Stephenson, P. T.; Wainwright, P.; Maddaford, A.; Zhang, X.; Leese, D.; Glen, R.; Hart, J.; Forrest, N.; Guyot, T. *Tetrahedron Lett.* 2011, 52, 6415.
- (139) Kumar, B. S.; Mishra, G. P.; Rao, B. V. Tetrahedron Lett. 2013, 54, 2845.
- (140) Petrignet, J.; Prathap, I.; Chandrasekhar, S.; Yadav, J. S.; Grée, R. Angew. Chem. Int. Ed. 2007, 46, 6297.
- (141) Hong, F.-T.; Lee, K.-S.; Liao, C.-C. J. Chin. Chem. Soc. 2000, 47, 77.
- (142) Kao, T.-C.; Chuang, G. J.; Liao, C.-C. Angew. Chem. Int. Ed. **2008**, 47, 7325.
- (143) (a) Leemans, E.; D'hooghe, M.; De Kimpe, N. *Chem. Rev.* **2011**, *111*, 3268. (b) Mack, D. J.; Njardarson, J. T. *ACS Catal.* **2013**, *3*, 272.
- (144) Brown, B.; Hegedus, L. S. J. Org. Chem. 2000, 65, 1865.
- (145) Mahuteau-Betzer, F.; Ghosez, L. *Tetrahedron* **2002**, *58*,
- 6991. 146) Gao E V : Burnell D I *L Own Cham* 2006 71 256
- (146) Gao, F. Y.; Burnell, D. J. J. Org. Chem. **2006**, 71, 356. (147) Coquerel, Y.; Greene, A. E.; Deprés, J.-P. Org. Lett. **2003**, 5,
- 4453. (148) Carrat S - Daprás I. D. Angou, Cham. Int. Ed. 2007, 44
- (148) Carret, S.; Deprés, J.-P. Angew. Chem. Int. Ed. 2007, 46, 6870.
- (149) Varea, T.; Alcalde, A.; de Castillo, C. L.; de Arellano, C. R.; Cossio, F. P.; Asensio, G. J. Org. Chem. 2012, 77, 6327.
- (150) (a) Geng, F.; Liu, J.; Paquette, L. A. Org. Lett. 2002, 4, 71.
 (b) Paquette, L. A.; Geng, F. Org. Lett. 2002, 4, 4547.
- (151) Grant, T. N.; West, F. G. J. Am. Chem. Soc. 2006, 128, 9348.
- (152) Trost, B. M.; Xia, J.; Maulide, N. J. Am. Chem. Soc. 2008, 130, 17258.
- (153) Pujanauski, B. G.; Prasad, B. A. B.; Sarpong, R. J. Am. *Chem. Soc.* **2006**, *128*, 6787.
- (154) Jung, M. E.; Davidov, P. Org. Lett. 2001, 3, 3025.
- (155) Matsuda, T.; Tsuboi, T.; Murakami, M. J. Am. Chem. Soc. 2007, 129, 12596.
- (156) Mehta, G.; Umarye, J. D. *Org. Lett.* **2002**, *4*, 1063.
- (157) Iqbal, M.; Duffy, P.; Evans, P.; Cloughley, G.; Allan, B.; Lledó, A.; Verdaguer, X.; Riera, A. Org. Biomol. Chem. 2008, 6, 4649.
- (158) Demuynck, A. L.W.; Levecque, P.; Kidane, A.; Gammon,
 D. W.; Sickle, E.; Jacobs, P. A.; De Vos, D. E.; Sels, B. F.
 Adv. Synth. Catal. 2010, 352, 3419.
- (159) Shindo, M.; Sato, Y.; Shishido, K. J. Org. Chem. 2001, 66, 7818.
- (160) Shindo, M.; Sato, Y.; Shishido, K. Tetrahedron Lett. 2002, 43, 5039.
- (161) (a) Kuhn, C.; Skaltsounis, L.; Monneret, C.; Florent, J.-C. *Eur. J. Org. Chem.* **2003**, 2585. (b) Kuhn, C.; Roulland, E.; Madelmont, J.-C.; Monneret, C.; Florent, J.-C. *Org. Biomol. Chem.* **2004**, *2*, 2028.
- (162) Ballini, R.; Boscia, G.; Fiorini, D.; Gil, M. V.; Petrini, M. Org. Lett. 2001, 3, 1265.
- (163) Pal, R.; Clark, R. J.; Manoharan, M.; Alabugin, I. V. J. Org. Chem. 2010, 75, 8689.
- (164) Heck, M.-P.; Matt, C.; Wagner, A.; Toupet, L.; Mioskowski, C. Eur. J. Org. Chem. 2010, 966.