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Introduction

In recent times, several advancements were made in using catalytic amounts of iodine in combination with *tert*-butyl hydroperoxide (TBHP) as co-oxidant. The increased utility is mainly due to its inexpensive, environmentally be-

nign nature, good efficiency and compatibility to work in place of rare or toxic heavy metal oxidants.¹ The I₂/TBHP catalytic system works efficiently for numerous C–C and C–X (X = heteroatom) bond-forming organic transformations under mild reaction conditions to offer the desired products in excellent yields.²

Table 1 Use of Iodine-TBHP

(A) Jiang et al. reported an I₂/TBHP mediated domino oxidative cyclization for the one-pot synthesis of polysubstituted oxazoles from readily available styrenes and benzylamines under mild conditions.³

(B) Zhang et al. demonstrated the synthesis of 2-phenylquinazolines in good to excellent yields via tandem sp³ C–H functionalization of 2-amino benzophenones and benzylic amines.⁴

(C) Manjunath and Prabhu reported a metal-free catalytic route to 2-aminobenzoxazoles by amination of benzoxazoles via C-H bond activation of primary or secondary amines. Further the methodology was demonstrated to synthesize therapeutically active benzoxazoles.⁵

(D) An I₂/TBHP catalyzed first efficient and direct synthesis of tertiary amides from alcohols and dimethylformamide has been developed. This transition-metal-free protocol provides a practical synthetic tool for the construction of *N*,*N*-dimethyl-substituted amides.⁶

I₂ (0.5 equiv) TBHP (6.0 equiv) i-PrOH, r.t. 45 examples 40-95%

+ R-SO₂Na
$$\frac{I_2 (0.1 \text{ equiv})}{AcOH, r.t.}$$
 + R-SO₂Na $\frac{I_2 (0.1 \text{ equiv})}{AcOH, r.t.}$ + R-SO₂Na $\frac{I_3 (0.1 \text{ equiv})}{AcOH, r.t.}$ + R-SO₂Na $\frac{I_3 (0.1 \text{ equiv})}{AcOH, r.t.}$ + OOO'.

25 examples 34-90%

l₂ (0.1 equiv) l₂ (1.5 equiv) TBHP (1.0 equiv) TBHP (1.0 equiv) DMSO. 80 °C DMSO, 80 °C 26 examples 28–88% 7 examples 70–81%

> l₂ (1.2 equiv) TBHP (5.0 equiv) 16 examples DMSO, 80 °C

(E) A metal-free oxidative coupling of methyl ketones and primary or secondary amines to α-keto amides was developed by Wan and his group. Four types of intermediates, α -iodo ketones, α -amino ketones, iminium intermediates and α-hydroxy amines were identified through a series of control experiments. The atom-economic methodology can be scaled-up, tolerates a variety of functional groups, and is operationally simple.⁷

(F) A novel I₂/TBHP catalyzed selective 2-arylsulfonylation of indoles was demonstrated. Various substituted 2-arylsulfonyl indoles were obtained in good to excellent yields in one pot. The direct sulfonylation reaction occurred selectively at C-2 position of the indole ring and only a catalytic amount of iodine acted as an efficient promoter. This method is a novel alternative approach for the synthesis of biologically important hetero diaryl sulfones from sodium sulfinates.8

(G) Recently, we have developed an I2/TBHP mediated synthesis of isatin and iodoisatin from 2'-aminoacetophenone via intramoleular oxidative amidation of the sp³ C-H bond. The reaction proceeds through sequential iodination, Kornblum oxidation, and amidation in one pot. The stoichiometric amount of I₂ plays a significant role in delivering iodoisatin exclusively in high yields.

(H) Ji's research group reported an I₂/TBHP mediated oxidation of commercially available indoles to isatins in moderate to good yields.10

References

- (1) (a) Finkbeiner, P.; Nachtsheim, B. J. Synthesis 2013, 979. (b) Kirihara, M.; Asai, Y.; Ogawa, S.; Noguchi, T.; Hatano, A.; Hirai, Y. Synthesis 2007, 3286.
- (2) (a) Hummel, S.; Kirsch, S. F. Beilstein J. Org. Chem. 2011, 7, 847. (b) Zhao, Q.; Miao, T.; Zhang, X.; Zhou, W.; Wang, L. Org. Biomol. Chem. 2013, 11, 1867.
- (3) Jiang, H.; Huang, H.; Cao, H.; Qi, C. Org. Lett. 2010, 12, 5561.
- (4) Zhang, J.; Zhu, D.; Yu, C.; Wan, C.; Wang, Z. Org. Lett. 2010, 12,
- (5) Manjunath, L.; Prabhu, K. R. J. Org. Chem. 2011, 76, 7938.
- (6) Xu, K.; Hu, Y.; Zhang, S.; Zha, Z.; Wang, Z. Chem. Eur. J. 2012, 18, 9793.
- Wei, W.; Shao, Y.; Hu, H.; Zhang, F.; Zhang, C.; Xu, Y.; Wan, X. J. Org. Chem. 2012, 77, 7157.
- (8) Xiao, F.; Chen, H.; Xie, H.; Chen, S.; Yang, L.; Deng, G.-J. Org. Lett. 2014, 16, 50.
- (9) Ilangovan, A.; Satish, G. J. Org. Chem. 2014, 79, 4984.
- (10) Zi, Y.; Cai, Z.-J.; Wang, S.-Y.; Ji, S.-J. Org. Lett. 2014, 16, 3094.