Metal-Free Synthesis of Biaryl- and Teraryl-Cored Diarylmethanes by Ring Transformation of 2H-Pyran-2-ones

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Abstract  
An efficient metal-free approach for the synthesis of functionalized biaryl-cored diarylmethanes is described by the ring transformation of 2H-pyran-2-ones using 4-phenylbutan-2-one as carbanion source. Moreover, 2H-pyran-2-ones were reacted with 1,3-diphenylacetone in the presence of base to achieve functionalized teraryl-cored diarylmethanes. All the ring transformation reactions were performed under mild reaction conditions to afford the biaryl- and teraryl-cored reaction products in high yields.

Key words  
biaryl-cored diarylmethanes, teraryl-cored diarylmethanes, 2H-pyran-2-ones, carbanion, ring transformation reactions, 4-phenylbutan-2-one, 1,3-diphenylacetone

Functionalized diarylmethanes are important scaffolds found in various biologically active synthetic and naturally occurring compounds.1 Diarylmethane-cored compounds are known to exhibit various biological activities such as antitumor,2 thyroid hormone and histamine H1-receptor antagonist,3,4 antiviral,5 antiallergic,6 and antidiabetic activities.7 Molecules embedded with diarylmethane units are widely found in various biologically active compounds (Figure 1). Beclobrate (1) has been introduced as potent cholesterol and triglyceride-lowering drug.8 Trimethoprim (2) has been developed as effective antibiotic and used for the treatment of urinary tract infections.9 In addition, naturally occurring compound avrainvilleol (3) is isolated from red algae and found to exhibit antioxidant activity.10 Moreover, the alkaloid papaverine (4), isolated from Papaver somniferum L., exhibits diverse biological activities such as antiplasmodic activity,11 cerebral vasodilator,12 and non-selective phosphodiesterase inhibitory activity.13 In addition, various tetronic acid derived chiral analogues of diarylmethanes have been found to show anti-HIV and anticancer activities.14 Recently diarylmethane derivatives have been used as an important precursors in several dye preparations.15

There are several methods available in literature for the synthesis of diarylmethanes, but most of them are associated with transition-metal-catalyzed coupling reactions such as Pd-catalyzed Suzuki type cross-coupling reactions of benzylic halides with arylboranes.16 In 2015, Yoshikai and co-workers reported the synthesis of diarylmethanes by ortho-CH benzylation of arylimines using Cobalt-Pyphos catalytic system.17 In 2016, Zhang and co-workers described a one-pot synthesis of diarylmethanes from benzyl chlorides by a Ni-catalyzed reductive cross-coupling reactions.18 Furthermore, synthesis of allyldiarylmethanes was accomplished via 1,6-conjugate allylation of p-quinonemethides using B(C6F5)3 as catalyst.19 In 2016, Hemelaere and co-workers developed the synthesis of diarylmethanes by Friedel–Crafts reaction of benzyl fluorides in the presence of TFA.20 Furthermore, the synthesis of diaryl-
methanes was achieved through transition-metal-free cross-coupling reaction of benzylic bromides with aryloboronic acids in the presence of Cs₂CO₃ using the solvent combination BTF/H₂O (10:1).⁹

Most of the existing approaches are associated with some limitations such as the use of toxic transition-metal catalysts and harsh reaction conditions. Despite the availability of several existing approaches, there is still scope to develop a new approach that could overcome the problems associated with them and offers the flexibility of introducing a wide range of functional groups in the diarylmethane architecture.

Herein, we report a metal-free approach for the synthesis of functionalized biaryl-cored diarylmethanes 9 in high yields by the ring transformation of 6-aryl-2H-pyran-2-ones 8 using 4-phenylbutan-2-one (6) as a source of nucleophile. The parent precursors 5 were synthesized by the reaction of methyl 2-cyano-3,3-dimethylsulfanylacrylate with functionalized acetophenones in DMSO at room temperature under alkaline conditions.²² Furthermore, the substrates 5 were reacted with secondary amines in methanol at reflux temperature to synthesize 6-aryl-4-amino-2H-pyran-2-ones 8.²²

The ring transformation of 2H-pyran-2-ones has been used to synthesize various arenes,²³ heteroarenes,²⁴ and fused cyclic systems.²⁵ Recently, we have reported the ultrasound-assisted synthesis of functionalized 2-tetralones via ring transformation of 2H-pyran-2-ones.²⁶

Our approach to prepare functionalized biaryl-cored diarylmethanes 9 was based on the ring transformation of 6-aryl-2H-pyran-2-ones 5 and 8 using 4-phenylbutan-2-one (6) as a carbannion source. Both substrates 5 and 8 have three electrophilic centers at C-2, C-4, and C-6. The presence of the electron-withdrawing substituent at C-3 position of pyran ring and the extended conjugation makes the latter position to be more reactive towards nucleophiles.

Initially, our studies were focused on the ring transformation of 3-cyano-4-methylsulfonyl-2H-pyran-2-ones (5a) with 4-phenylbutan-2-one (6). The ring transformation of substrate 5a with 6 was performed in DMF in the presence of KOH at room temperature (Table 1, entry 1). Unfortunately, the reaction suffered from several unwanted side reactions and ring transformed product 7a was obtained in 50% yield. When other substrates 5b and 5c were used in similar reactions and the desired products were observed in slightly improved yields (entries 2 and 3). Probably, the presence of SMe group at C-4 position of lactone ring in substrate 5 makes this position more susceptible for nucleophilic attack, which leads to various undesired side products.

Next, our efforts were directed to limit the reactivity of 2H-pyran-2-ones 5 at C-4 position towards the nucleophile. In order to limit the reactivity at C-4 position, the leaving group SMe in substrates 5 was replaced with tert-amino functionality by treating with cyclic amines and new substrates 6-phenyl-4-amino-2H-pyran-2-ones 8 were synthesized.²² Further, to know the influence of amino functionality in the ring transformation reaction, the substrate 2-oxo-6-phenyl-4-(piperidin-1-yl)-2H-pyran-3-carbonitrile (8a) was treated with the same ketone 6 under similar reaction conditions and a significant improvement was observed in the yield of ring-transformed product 9a (Scheme 1).

Table 1  Ring Transformation of 6-Aryl-4-(methylthio)-2-oxo-2H-pyran-3-carbonitriles 5a–c with 4-Phenylbutan-2-one (6)

<table>
<thead>
<tr>
<th>Entry</th>
<th>Ar</th>
<th>Reaction time (h)</th>
<th>Yield (%) of 7</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Ph</td>
<td>10</td>
<td>50</td>
</tr>
<tr>
<td>2</td>
<td>4-BrC₆H₄</td>
<td>12</td>
<td>52</td>
</tr>
<tr>
<td>3</td>
<td>4-MeOC₆H₄</td>
<td>10</td>
<td>57</td>
</tr>
</tbody>
</table>

Scheme 1

After that our efforts were directed towards the screening of different solvents using 8a as model substrate. Various polar and nonpolar solvents were employed for the ring transformation of model substrate 8a to the corresponding synthesis of diarylmethane derivative 9a using ketone 6 as source of carbannion and the results obtained are summarized in Table 2. Initially, the ring transformation reaction of 8a was performed in polar and aprotic solvents DMF and DMSO, and the reaction product 9a was isolated in 85% and 82% yield, respectively (Table 2, entries 1 and 2). The yield of ring transformed product was slightly lowered when acetonitrile was used as solvent (entry 3). The yield was further reduced up to 69% yield in EtOAc (entry 4). Additionally, the ring transformation reaction was performed in polar and protic solvents but could not proceed (entries 5–7). The ring transformation reaction could proceed in dichloromethane but desired product 9a was obtained in moderate yield (entry 8). The course of reaction was investigated in aprotic and nonpolar solvents such as benzene and toluene. The ring transformed product was formed in toluene but was observed only in traces in benzene (entries 9 and 10).
Finally, the reaction was performed in THF, diethyl ether, and 1,4-dioxane but the reaction did not proceed in any of these solvents (entries 11–13).

**Table 2** Solvent Optimization for the Ring Transformation of 6-Phenyl-2H-pyran-2-one 8a to Diarylmethane Derivative 9a

<table>
<thead>
<tr>
<th>Entry</th>
<th>Solvent</th>
<th>Reaction time (h)</th>
<th>9a Yield (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>DMF</td>
<td>10</td>
<td>85</td>
</tr>
<tr>
<td>2</td>
<td>DMSO</td>
<td>10</td>
<td>82</td>
</tr>
<tr>
<td>3</td>
<td>MeCN</td>
<td>10</td>
<td>79</td>
</tr>
<tr>
<td>4</td>
<td>EtOAc</td>
<td>10</td>
<td>69</td>
</tr>
<tr>
<td>5</td>
<td>AcOH</td>
<td>14</td>
<td>–</td>
</tr>
<tr>
<td>6</td>
<td>MeOH</td>
<td>14</td>
<td>–</td>
</tr>
<tr>
<td>7</td>
<td>EtOH</td>
<td>14</td>
<td>–</td>
</tr>
<tr>
<td>8</td>
<td>CH2Cl2</td>
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<td>58</td>
</tr>
<tr>
<td>9</td>
<td>toluene</td>
<td>12</td>
<td>30</td>
</tr>
<tr>
<td>10</td>
<td>benzene</td>
<td>12</td>
<td>–</td>
</tr>
<tr>
<td>11</td>
<td>Et2O</td>
<td>12</td>
<td>–</td>
</tr>
<tr>
<td>12</td>
<td>THF</td>
<td>14</td>
<td>–</td>
</tr>
<tr>
<td>13</td>
<td>1,4-dioxane</td>
<td>14</td>
<td>–</td>
</tr>
</tbody>
</table>

After determining the optimal solvent, our aim was to screen the different bases for the same ring transformation reaction. Several bases were employed for the ring transformation of 8a to the desired product 9a and results are listed in Table 3. Initially, the reaction was carried out with KOH in DMF and ring transformation product was obtained in 85% yield (Table 3, entry 1). Similarly, the reaction was tested with NaHCO3 under the same reaction condition and the product 9a was isolated in 75% yield (entry 2). Additionally, K2CO3 and Cs2CO3 were also used as base for same reaction and desired product 9a was achieved in 65% and 60% yield, respectively (entries 3 and 4). The course of reaction was quite similar with LiOH and the reaction product 9a was isolated in 63% yield (entry 5).

After the optimization studies, the presence of KOH as base and DMF as solvent at room temperature for 10 hours was found as the best reaction conditions for the ring transformation of 6-phenyl-2H-pyran-2-one 8a to 2-benzyl-3-methyl-5-(piperidin-1-yl)[1H-biphenyl]-4-carbonitrile (9a).

After achieving the best reaction conditions, a series of biaryl-cored diarylmethanes 9a–k were synthesized in 78–94% yields by the reaction of various 2H-pyran-2-ones 8a–k with 4-phenyl-butan-2-one (6) in DMF in the presence of KOH for 10–14 hours at room temperature (Table 4, entries 1–11). The ring transformation reaction proceeded well with both electron-withdrawing and electron-donating groups on the aromatic ring of 6-aryl-2H-pyran-2-ones 8. Notably, the ring transformation products 9f–h were ob-
In order to generalize this approach, the same synthetic protocol was applied to construct teraryl-cored diarylmethanes 11. To achieve the synthesis of teraryl-cored diarylmethanes 11, the similar substrates 6-aryl-2H-pyran-2-ones 8 were treated with 1,3-diphenylacetone (10) in DMF in the presence of KOH at room temperature and teraryl-cored diarylmethanes 11 were obtained 74–95% yields (Table 5, entries 1–11). Various functional groups were successfully tolerated during these ring transformations.

It was observed that the course of reaction was quite similar with both electron-donating and -withdrawing cored substrates but ring transformation products were obtained in slightly higher yields in the case of substrates having electron-donating functionalities (Table 5, entries 6–8). Additionally, the reaction was found to be slightly slower when substrates 8j and 8l having methyl group at C-5 position were used and reaction products were obtained in 74% and 79% yield, respectively (entries 10 and 11). All the synthesized compounds were characterized by spectroscopic analysis.

Finally, the reaction of substrate 5a was performed with 1,3-diphenylacetone (10) under similar reaction conditions. Expectedly, the reaction suffered from some undesired side reactions probably due to the presence of good leaving SMe at C-4 position in substrate 5a, which makes this position more vulnerable towards the nucleophile. The ring transformation product 12a was isolated in 62% yield (Scheme 2). Unfortunately, we could not isolate any side product from the reaction mixture. The isolated compound 12a was characterized as 3′-benzyl-5′-(methylthio)-[1,1′:2′,1″-terphenyl]-4′-carbonitrile by its spectroscopic analysis.

The possible mechanism for the ring transformation of 2H-pyran-2-ones 8 with 4-phenylbutan-2-one (6) to diarylmethanes 9 is described in Scheme 3. Initially, the formation of bicyclic intermediate 13 takes place by the nucleophilic attack of anion generated from ketone 6 to the C-6 position of 2H-pyran-2-ones 8, followed by intramolecular cyclization involving the carbonyl functionality of 6 and C-3 of the pyranone ring. Furthermore, the bicyclic intermediate 13 transforms to final product 9 on decarboxylation followed by dehydration.

In conclusion, we have developed a facile metal-free synthetic methodology for the synthesis of functionalized biaryl-cored diarylmethanes 9 through carbanion-induced ring transformation of 6-aryl-2H-pyran-2-ones 8 in good yields. In addition, the same approach was employed to achieve the synthesis of teraryl-cored diarylmethanes 11. Our methodology for the synthesis of functionalized diarylmethanes is simple, economical and does not require any toxic transition metal. Further investigations about this ring transformation approach are currently in progress.

Melting points were measured with REMI DDMS 2545 melting point apparatus. IR spectra were recorded with a Thermo Scientific Nicolet Nexus 470FT-IR spectrophotometer and band positions are reported in reciprocal centimeters. Samples were subjected to ATR mode to record the IR data. 1H NMR and 13C NMR spectra were recorded on a Bruker AV-400 spectrometer using the solvents indicated at 400 and
100 MHz, respectively. Mass spectra (m/z) were recorded under the conditions of electron ionization (EI). All reactions were monitored by TLC that was performed on pre-coated sheets of silica gel 60 and column chromatography was performed with Al₂O₃ (neutral, 95%) (Avra synthesis Pvt. Ltd.). Hexane and EtOAc were used as eluting solvents and bought from Avra Synthesis Pvt. Ltd. DMF was bought from Avra Synthesis Pvt. Ltd and was used without further purification.

Diarylmethanes 7a–c; General Procedure

A mixture of 3-cyano-4-methylsulfanyl-2H-pyran-2-one 5 (1.0 mmol, 1.0 equiv), 4-phenylbutan-2-one (6; 0.18 mL, 1.2 mmol, 1.2 equiv), and powdered KOH (84 mg, 1.5 mmol, 1.5 equiv) in DMF (5 mL) was stirred at r.t. for 10–12 h. The course of reaction was monitored by TLC. On completion of the reaction, few ice pieces were added to the reaction mixture and neutralized with aq 2 M HCl. The mixture was extracted with EtOAc (3 × 10 mL), the combined organic layers were dried (anhyd Na₂SO₄), filtered, and concentrated under vacuum. The crude residue was purified by neutral alumina column chromatography using EtOAc–hexane (1:49) as an eluent and isolated products were characterized as diarylmethanes 7 by their spectroscopic analysis (Table 1).

2-Benzyl-3-methyl-5-(methylthio)[1,1′-biphenyl]-4-carbonitrile (7a)

White solid; yield: 164 mg (0.5 mmol, 50%); mp 148–150 °C; R_{f} = 0.5 (EtOAc–hexane 1:49).

IR (ATR): 2212 cm⁻¹ (C≡N).

¹H NMR (400 MHz, CDCl₃): δ = 2.33 (s, 3 H, CH₃), 2.46 (s, 3 H, SCH₃), 3.88 (s, 2 H, CH₂), 6.81 (d, J = 7.2 Hz, 2 H, ArH), 6.98 (s, 1 H, ArH), 7.05–7.20 (m, 5 H, ArH), 7.22–7.28 (m, 3 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 15.9, 18.9, 35.8, 112.1, 116.6, 125.4, 126.1, 127.7, 127.9, 128.3, 128.5, 128.6, 133.8, 139.6, 140.6, 141.2, 143.3, 147.8.

GC-MS: m/z = 330 [M + 1]⁺.

Anal. Calcd for C₂₂H₁₉NS: C, 80.20; H, 5.81; N, 4.25; S, 9.73. Found: C, 79.75; H, 5.83; N, 4.20; S, 9.06.

2-Benzyl-4′-bromo-3-methyl-5-(methylthio)[1,1′-biphenyl]-4-carbonitrile (7b)

White solid; yield: 212 mg (0.52 mmol, 52%); mp 150–152 °C; R_{f} = 0.5 (EtOAc–hexane 1:49).

IR (ATR): 2214 cm⁻¹ (C≡N).

¹H NMR (400 MHz, CDCl₃): δ = 2.34 (s, 3 H, CH₃), 2.46 (s, 3 H, SCH₃), 3.86 (s, 2 H, CH₂), 6.79 (d, J = 6.8 Hz, 2 H, ArH), 6.94 (s, 1 H, ArH), 7.07–7.20 (m, 3 H, ArH), 7.38 (td, J₁ = 8.8 Hz, J₂ = 2.0 Hz, 2 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 15.8, 18.9, 35.7, 112.4, 116.5, 122.2, 125.1, 126.3, 127.6, 128.7, 130.2, 131.5, 133.6, 139.3, 139.4, 141.5, 143.4, 146.5.


Anal. Calcd for C₂₂H₁₈BrNS: C, 64.71; H, 4.44; N, 3.43; S, 7.85. Found: C, 64.69; H, 4.50; N, 3.39; S, 7.81.

2-Benzyl-4′-methoxy-3-methyl-5-(methylthio)[1,1′-biphenyl]-4-carbonitrile (7c)

White solid; yield: 204 mg (0.57 mmol, 57%); mp 144–146 °C; R_{f} = 0.5 (EtOAc–hexane 1:49).

IR (ATR): 2214 cm⁻¹ (C≡N).

¹H NMR (400 MHz, CDCl₃): δ = 2.32 (s, 3 H, CH₃), 2.46 (s, 3 H, SCH₃), 3.73 (s, 3 H, OCH₃), 3.90 (s, 2 H, CH₂), 6.78 (td, J₁ = 8.8 Hz, J₂ = 2.0 Hz, 2 H, ArH), 6.82 (d, J = 7.2 Hz, 2 H, ArH), 6.98 (s, 1 H, ArH), 7.03 (td, J₁ = 8.8 Hz, J₂ = 2.0 Hz, 2 H, ArH), 7.08–7.20 (m, 3 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 15.9, 18.9, 35.8, 55.3, 111.8, 113.7, 116.7, 125.6, 126.1, 127.7, 127.9, 128.3, 128.5, 128.6, 133.8, 139.6, 140.6, 141.2, 143.3, 147.5, 159.3.

GC-MS: m/z = 360 [M + 1]+.

Anal. Calcd for C₂₃H₂₁NOS: C, 76.85; H, 5.84; N, 3.90; S, 8.92. Found: C, 76.10; H, 5.84; N, 3.84; S, 8.58.

Scheme 3 Proposed mechanism for the synthesis of diarylmethanes 9 by the ring transformation of 2H-pyran-2-ones 8 with 4-phenylbutan-2-one (6)
Biaryl-Cored Diarylmethanes 9a–k; General Procedure

A mixture of 2H-pyran-2-one 8 (1.0 mmol, 1.0 equiv), 4-phenylbutan-2-one 6 (0.18 mL, 1.2 mmol, 1.2 equiv), and powdered KOH (84 mg, 1.5 mmol, 1.5 equiv) in DMF (5 mL) was stirred at r.t. for 10–14 h. The course of reaction was monitored by TLC. On completion of the reaction, few ice pieces were added to the reaction mixture and neutralized with q 2 M HCl. The reaction mixture was extracted with EtOAc (3 × 10 mL), the combined organic layers were dried (anhdy Na2SO4), filtered, and concentrated under vacuum. The crude residue was purified by neutral alumina column chromatography using EtOAc–hexane (1:49) as an eluent and isolated products were characterized as biaryl-cored diarylmethanes 9 by their spectroscopic analysis (Table 4).

2-Benzyl-3-methyl-5-(piperidin-1-yl)[1,1′-biphenyl]-4-carboxitrile (9a)

White solid; yield: 311 mg, 0.85 mmol (85%); mp 114–116 °C; Rf = 0.5 (EtOAc–hexane 1:49).

IR (ATR): 2215 cm⁻¹ (C=O).

1H NMR (400 MHz, CDCl3): δ = 1.46–1.54 (m, 2 H, CH2), 1.66–1.75 (m, 4 H, 2 × CH2), 2.30 (s, 3 H, CH3), 3.06 (t, J = 5.2 Hz, 4 H, 2 × NCH2), 3.84 (s, 2 H, CH2), 6.70 (s, 1 H, ArH), 6.82 (d, J = 7.6 Hz, 2 H, ArH), 7.03–7.18 (m, 5 H, ArH), 7.20–7.25 (m, 3 H, ArH).

13C NMR (100 MHz, CDCl3): δ = 18.9, 24.1, 26.2, 35.6, 53.5, 107.7, 117.9, 118.2, 126.0, 127.7, 128.4, 128.6, 129.4, 129.9, 133.7, 139.8, 140.0, 143.5, 146.9, 155.7.


Anal. Calc. for C26H25N2: C, 81.65; H, 6.98; N, 6.79. Found: C, 81.62; H, 6.87; N, 6.84.

2-Benzyl-4′-bromo-3-methyl-5-(piperidin-1-yl)[1,1′-biphenyl]-4-carboxitrile (9d)

White solid; yield: 364 mg (0.82 mmol, 82%); mp 116–118 °C; Rf = 0.5 (EtOAc–hexane 1:49).

IR (ATR): 2217 cm⁻¹ (C=O).

1H NMR (400 MHz, CDCl3): δ = 1.47–1.56 (m, 2 H, CH2), 1.66–1.73 (m, 4 H, 2 × CH2), 2.30 (s, 3 H, CH3), 3.06 (t, J = 4.8 Hz, 4 H, 2 × NCH2), 3.81 (s, 2 H, CH2), 6.65 (s, 1 H, ArH), 6.80 (d, J = 7.6 Hz, 2 H, ArH), 6.95 (d, J = 8.0 Hz, 2 H, ArH), 7.03–7.19 (m, 3 H, ArH), 7.34 (d, J = 8.0 Hz, 2 H, ArH).

13C NMR (100 MHz, CDCl3): δ = 18.9, 24.1, 26.2, 35.6, 53.5, 107.7, 117.9, 118.1, 121.9, 126.1, 127.7, 128.6, 129.4, 130.2, 131.3, 134.0, 140.2, 143.6, 146.8, 155.7.


Anal. Calc. for C26H23BrN2: C, 67.11; H, 5.66; N, 6.29. Found: C, 67.03; H, 5.66; N, 6.11.

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**2-Benzyl-4′-methoxy-3-methyl-5-(piperidin-1-yl)-1,1′-biphenyl]-4-carbonitrile (9g)**

White solid; yield: 356 mg (0.90 mmol, 90%); mp 172–174 °C; Rf = 0.5 (EtOAc–hexane 1:4).

IR (ATR): 2213 cm⁻¹ (C=N).

1H NMR (400 MHz, CDCl₃): δ = 1.45–1.58 (m, 2 H, CH₂), 1.66–1.75 (m, 4 H, 2 × CH₂), 2.28 (s, 3 H, CH₃), 3.05 (t, J = 5.2 Hz, 4 H, 2 × NCH₂), 3.71 (s, 3 H, OCH₃), 3.86 (s, 2 H, CH₂), 6.70 (s, 1 H, ArH), 6.75 (d, J = 8.0 Hz, 2 H, ArH), 6.83 (d, J = 7.2 Hz, 2 H, ArH), 7.02 (d, J = 8.0 Hz, 2 H, ArH), 7.04–7.10 (m, 1 H, ArH), 7.14 (t, J = 7.2 Hz, 2 H, ArH).

13C NMR (100 MHz, CDCl₃): δ = 18.9, 24.2, 26.3, 35.7, 53.6, 55.3, 107.2, 113.6, 118.2, 118.5, 125.9, 127.8, 129.7, 133.9, 140.1, 140.2, 148.9, 151.5, 151.8.

GC-MS: m/z = 458 [M + 1]⁺.


**3-Benzyl-2-methyl-6-(piperidin-1-yl)-4-(thiophen-2-yl)benzonitrile (9k)**

White solid; yield: 308 mg (0.83 mmol (83%); mp 115–117 °C; Rf = 0.5 (EtOAc–hexane 1:4).

IR (ATR): 2215 cm⁻¹ (C=N).

1H NMR (400 MHz, CDCl₃): δ = 1.49–1.58 (m, 2 H, CH₂), 1.67–1.76 (m, 4 H, 2 × CH₂), 2.30 (s, 3 H, CH₃), 3.07 (t, J = 5.2 Hz, 4 H, 2 × NCH₂), 3.87 (s, 2 H, CH₂), 6.79 (s, 1 H, ArH), 6.82 (d, J = 6.8 Hz, 2 H, ArH), 7.07 (d, J = 6.8 Hz, 1 H, ArH), 7.13–7.24 (m, 5 H, ArH).

13C NMR (100 MHz, CDCl₃): δ = 19.0, 35.7, 49.6, 51.9, 55.3, 107.3, 113.7, 116.4, 118.5, 120.1, 126.0, 127.8, 128.5, 129.7, 132.9, 130.8, 133.5, 140.2, 143.7, 148.1, 151.2, 154.1, 159.2.

GC-MS: m/z = 474 [M + 1]⁺.

Anal. Calcd for C₂₃H₂₄N₂S: C, 77.38; H, 6.49; N, 7.52; S, 8.61. Found: C, 77.68; H, 6.44; N, 7.43; S, 8.50.

**3-Benzyl-2-methyl-6-(piperidin-1-yl)-4-(thiophen-2-yl)benzonitrile (9k)**

White solid; yield: 356 mg (0.90 mmol, 90%); mp 140–142 °C; Rf = 0.5 (EtOAc–hexane 1:4).

IR (ATR): 2215 cm⁻¹ (C=N).

1H NMR (400 MHz, CDCl₃): δ = 1.89 (s, 3 H, CH₃), 2.30 (s, 3 H, CH₃), 3.18–3.55 (m, 8 H, 4 × NCH₂), 3.60 (s, 2 H, CH₂), 6.72 (d, J = 7.2 Hz, 2 H, ArH), 6.79 (d, J = 7.2 Hz, 1 H, ArH), 6.85–6.94 (m, 4 H, ArH), 7.01–7.15 (m, 3 H, ArH), 7.16–7.24 (m, 5 H, ArH).

13C NMR (100 MHz, CDCl₃): δ = 16.9, 18.5, 36.6, 50.5, 50.6, 111.7, 116.6, 118.4, 125.8, 127.3, 127.8, 128.2, 128.3, 129.1, 133.5, 134.3, 139.8, 140.1, 140.2, 148.9, 151.5, 151.8.

GC-MS: m/z = 458 [M + 1]⁺.

3'-Benzy1-5'-(4-phenylpiperazin-1-yl)[1,1'2',1''-terphenyl]-4'-carbonitrile (11b)

White solid; yield: 414 mg (0.82 mmol, 82%); mp 167–169 °C; Rf = 0.5 (EtOAc–hexane:1:4)

IR (ATR): 2215 cm⁻¹ (C=O).

1H NMR (400 MHz, CDCl₃): δ = 3.32–3.16 (m, 8 H, 4 × NCH₂), 4.96 (s, 2 H, CH₂), 6.72–6.84 (m, 5 H, ArH), 6.88–6.96 (m, 5 H, ArH), 6.99–7.07 (m, 9 H, ArH), 7.20 (t, J = 7.6 Hz, 2 H, ArH).

13C NMR (100 MHz, CDCl₃): δ = 38.5, 49.6, 52.0, 107.3, 116.5, 117.9, 118.9, 120.1, 126.1, 126.9, 127.1, 127.7, 127.8, 128.2, 128.5, 129.2, 129.4, 130.8, 135.9, 137.9, 139.3, 140.8, 144.7, 147.4, 151.2, 155.3.


Anal. Calcd for C₃₉H₃₅N₃O: C, 82.96; H, 6.21; N, 7.84. Found: C, 82.86; H, 6.25; N, 7.73.

3'-Benzy1-4-methyl-5'-(piperidin-1-yl)[1,1'2',1''-terphenyl]-4'-carbonitrile (11f)

White solid; yield: 419 mg (0.95 mmol, 95%); mp 156–158 °C; Rf = 0.5 (EtOAc–hexane:1:4).

IR (ATR): 2210 cm⁻¹ (C=O).

1H NMR (400 MHz, CDCl₃): δ = 1.47–1.57 (m, 2 H, CH₂), 1.68–1.77 (m, 4 H, 2 × CH₂), 2.15 (s, 3 H, CH₃), 3.12 (t, J = 5.2 Hz, 4 H, 2 × NCH₂), 4.04 (s, 2 H, CH₂), 6.76 (d, J = 8.4 Hz, 4 H, ArH), 6.94–7.10 (m, 6 H, ArH).

13C NMR (100 MHz, CDCl₃): δ = 20.1, 24.1, 26.2, 28.8, 34.5, 53.6, 55.1, 106.8, 112.8, 119.0, 125.9, 126.7, 127.7, 128.1, 128.4, 128.5, 129.3, 130.9, 135.0, 136.6, 138.1, 138.4, 139.5, 144.3, 147.1, 156.8.

GC-MS: m/z = 443 [M + 1]+.

Anal. Calcd for C₃₂H₃₁N₂: C, 86.84; H, 6.33. Found: C, 86.77; H, 6.82; N, 6.24.

3'-Benzy1-4-methoxy-5'-(piperidin-1-yl)[1,1'2',1''-terphenyl]-4'-carbonitrile (11g)

White solid; yield: 421 mg (0.95 mmol, 92%); mp 95–97 °C; Rf = 0.5 (EtOAc–hexane:1:4).

IR (ATR): 2212 cm⁻¹ (C=O).

1H NMR (400 MHz, CDCl₃): δ = 1.48–1.59 (m, 2 H, CH₂), 1.68–1.77 (m, 4 H, 2 × CH₂), 3.12 (t, J = 5.2 Hz, 4 H, 2 × NCH₂), 3.63 (s, 3 H, OCH₃), 4.04 (s, 2 H, CH₂), 6.57 (d, J = 8.4 Hz, 4 H, ArH), 6.76 (t, J = 8.4 Hz, 4 H, ArH), 6.82–6.88 (m, 3 H, ArH), 6.98–7.10 (m, 6 H, ArH).

13C NMR (100 MHz, CDCl₃): δ = 24.2, 26.2, 38.5, 53.6, 55.1, 106.8, 113.1, 118.3, 119.0, 125.9, 126.8, 127.8, 128.1, 128.5, 130.6, 130.9, 133.4, 135.0, 138.4, 139.5, 144.3, 146.7, 156.8, 158.5.

GC-MS: m/z = 459 [M + 1]+.

Anal. Calcd for C₃₉H₃₅N₃O: C, 83.81; H, 6.59; N, 6.11. Found: C, 83.08; H, 6.70; N, 6.07.

3'-Benzy1-4-methoxy-5'-(4-phenylpiperazin-1-yl)[1,1'2',1''-terphenyl]-4'-carbonitrile (11h)

White solid; yield: 497 mg (0.93 mmol, 93%); mp 120–122 °C; Rf = 0.5 (EtOAc–hexane:1:4).

IR (ATR): 2209 cm⁻¹ (C=O).

1H NMR (400 MHz, CDCl₃): δ = 3.32–3.38 (m, 8 H, 4 × NCH₂), 3.64 (s, 3 H, OCH₃), 4.07 (s, 2 H, CH₂), 6.58 (d, J = 8.8 Hz, 2 H, ArH), 6.77 (t, J = 7.6 Hz, 4 H, ArH), 6.80–6.94 (m, 6 H, ArH), 6.98–7.11 (m, 6 H, ArH), 7.21 (t, J = 8.8 Hz, 2 H, ArH).

13C NMR (100 MHz, CDCl₃): δ = 38.5, 49.6, 52.0, 55.2, 106.9, 113.2, 116.5, 118.1, 118.9, 120.2, 126.0, 126.9, 127.9, 128.1, 128.5, 129.2, 130.6, 130.8, 133.2, 135.9, 138.2, 139.4, 144.7, 147.0, 151.2, 155.3, 158.6.

GC-MS: m/z = 536 [M + 1]+.

Anal. Calcd for C₃₅H₃₃N₃O: C, 82.96; H, 6.21; N, 7.84. Found: C, 82.86; H, 6.25; N, 7.73.

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2-Benzyl-6-(naphthalen-2-yl)-4-(piperidin-1-yl)[1,1'-biphenyl]-3-carbonitrile (11i)

White solid; yield: 430 mg (0.90 mmol, 90%); mp 182–184 °C; Rf = 0.5 (EtOAc–hexane 1:4).

IR (ATR): 2214 cm⁻¹ (C≡N).

1H NMR (400 MHz, CDCl₃): δ = 1.63–1.71 (m, 2 H, CH₂), 1.82–1.90 (m, 4 H, 2 × CH₂), 3.28 (t, J = 5.2 Hz, 4 H, 2 × NCH₂), 4.22 (s, 2 H, CH₂), 6.90–6.96 (m, 4 H, ArH), 7.08–7.13 (m, 5 H, ArH), 7.14–7.23 (m, 3 H, ArH), 7.42–7.50 (m, 2 H, ArH), 7.56 (d, J = 8.4 Hz, 1 H, ArH), 7.64 (s, 1 H, ArH), 7.71–7.78 (m, 2 H, ArH).

13C NMR (100 MHz, CDCl₃): δ = 144.5, 147.0, 156.9, 128.1, 128.4, 128.5, 130.9, 132.1, 132.9, 135.1, 138.2, 138.8, 139.5, 119.3, 125.9, 126.1, 126.2, 126.8, 126.9, 127.4, 127.6, 127.8, 127.9, 130.3, 133.9, 138.6, 139.5, 139.9, 139.9, 141.4, 148.2, 151.8, 152.8.

GC-MS: m/z = 479 [M + 1]⁺.


3'-Benzyloxy-5'-[(4-phenylpiperazin-1-yl)[1,1'-terphenyl]-4-carbonitrile (11j)

White solid; yield: 384 mg (0.74 mmol, 74%); mp 178–180 °C; Rf = 0.7 (EtOAc–hexane 1:49).

IR (ATR): 2217 cm⁻¹ (C≡N).

1H NMR (400 MHz, CDCl₃): δ = 1.99 (s, 3 H, CH₃), 3.22–3.60 (m, 8 H, 4 × NCH₂), 3.98 (s, 2 H, CH₂), 6.64 (dd, J₁ = 7.2 Hz, J₂ = 1.6 Hz, 2 H, ArH), 6.78 (dt, J = 7.2 Hz, J = 1.6 Hz, 5 H, ArH), 6.89–6.96 (m, 5 H, ArH), 6.95–7.09 (m, 6 H, ArH), 7.20 (t, J = 8.4 Hz, 2 H, ArH).

13C NMR (100 MHz, CDCl₃): δ = 17.2, 38.2, 50.5, 50.7, 111.4, 116.6, 118.5, 120.0, 126.0, 126.6, 127.4, 127.7, 128.1, 128.5, 129.1, 129.3, 130.3, 133.9, 138.6, 139.5, 139.9, 141.4, 148.2, 151.8, 152.8.

GC-MS: m/z = 520 [M + 1]⁺.


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Supporting Information

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