Asymmetric Synthesis of Chiral 1,3-Dimethyl Units Through a Double Michael Reaction of Nitromethane and Crotonaldehyde Catalyzed by Diphenylprolinol Silyl Ether

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Abstract An efficient synthetic route to install chiral 1,3-dimethyl units through a double Michael reaction of crotonaldehyde and nitromethane catalyzed by diphenylprolinol silyl ether is developed. Either 1,3-syn- or 1,3-anti-dimethyl units are obtained selectively depending on the enantiomer of the diphenylprolinol silyl ether catalyst used. The side chain of pneumocandin B₉ is synthesized enantioselectively by using the present method as a key step.

Key words organocatalyst, Michael reaction, asymmetric synthesis, diastereoselective reaction, diphenylprolinol silyl ether

The 1,3-dimethyl unit is found in many natural products, including siphonarienal,¹ ionomycin,² scyphostatin,³ and borrelidin (Figure 1),⁴ and the stereoselective synthesis of chiral 1,3-dimethyl units is considered an important synthetic topic.⁵ There are many methods available for the diastereo- and enantioselective synthesis of anti- and syn-1,3-dimethyl units. The iterative Michael reaction of methyl groups under reagent control is a widely employed method,⁶ and iterative allylic substitution and alklylation of chiral enolates is also used.⁷ Negishi’s Zr-catalyzed carboalumination (ZACA) reaction is a powerful method for the preparation of 1,3-dimethyl units,⁸ and Aggarwal recently reported an assembly-line synthesis that proceeds through iterative homologation of boronic esters with chiral lithiated benzoate esters and chloromethylithium.⁹ Some of the methods use asymmetric catalytic reactions.¹⁰,¹¹ In spite of these elegant methods, a procedure which is suitable for the large-scale preparation of 1,3-dimethyl units is needed.

We have already reported the asymmetric Michael reaction of an α,β-unsaturated aldehyde with nitromethane catalyzed by diphenylprolinol silyl ether¹² as an effective organocatalyst (Scheme 1).³ The sequential use of this Michael reaction would afford either the syn- or anti-1,3-dimethyl unit stereoselectively (Scheme 2). The Michael reaction of nitromethane and crotonaldehyde catalyzed by (S)-diphenylprolinol silyl ether (S)-1a,¹² followed by acetalization would provide 2. A second Michael reaction of the generated nitroalkane 2 and crotonaldehyde, catalyzed by either (S)- or (R)-diphenylprolinol silyl ether, would then afford the desired anti- or syn-1,3-dimethyl unit, respectively. The realization of this scenario is described herein.

Figure 1 Natural products with a 1,3-dimethyl unit

Scheme 1 An asymmetric Michael reaction catalyzed by a diphenylprolinol silyl ether
The first Michael reaction of crotonaldehyde and nitromethane was carried out using 5 mol% of (S)-diphenylprolinol diphenylmethylsilyl ether (S)-1a\(^1\) as the catalyst in MeOH in the presence of 10 equivalents of H\(_2\)O to afford the Michael product, which was treated with HC(OMe)\(_3\)) and a catalytic amount of TsOH in the same vessel to provide nitroacetal 2 in 94% yield and 90% ee (Scheme 3). This reaction required four days to reach completion when THF was employed as the solvent, as described previously,\(^1\) but was complete within two days in MeOH and proceeded with excellent enantioselectivity.

**Scheme 3** The initial Michael reaction of crotonaldehyde and nitromethane

The second Michael reaction of 2 and crotonaldehyde was then investigated using diphenylprolinol trimethylsilyl ether (S)-1b as the catalyst (Table 1). Alcohol 3 was obtained in 43% yield as a diastereomeric mixture after treatment of the Michael product with NaBH\(_4\) (entry 1). To improve this yield, four days to reach completion when THF was employed as the solvent, as described previously,\(^1\) but was complete within two days in MeOH and proceeded with excellent enantioselectivity.

**Table 1** The Effect of Solvent, Additive and Addition Time on the Asymmetric Michael Reaction of Nitroalkane 2 and Crotonaldehyde

<table>
<thead>
<tr>
<th>Entry</th>
<th>Solvent</th>
<th>H(_2)O (equiv)(^b)</th>
<th>Time (h)</th>
<th>Yield (%)(^c)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>MeOH</td>
<td>10</td>
<td>7.5</td>
<td>43</td>
</tr>
<tr>
<td>2</td>
<td>MeOH</td>
<td>0</td>
<td>7.5</td>
<td>&lt;5</td>
</tr>
<tr>
<td>3</td>
<td>MeOH</td>
<td>10</td>
<td>2</td>
<td>&lt;5</td>
</tr>
<tr>
<td>4</td>
<td>THF</td>
<td>10</td>
<td>28</td>
<td>&lt;5</td>
</tr>
<tr>
<td>5</td>
<td>THF</td>
<td>0</td>
<td>28</td>
<td>&lt;5</td>
</tr>
<tr>
<td>6</td>
<td>neat</td>
<td>10</td>
<td>28</td>
<td>&lt;5</td>
</tr>
<tr>
<td>7</td>
<td>neat</td>
<td>0</td>
<td>28</td>
<td>&lt;5</td>
</tr>
<tr>
<td>8</td>
<td>MeOH</td>
<td>10</td>
<td>11</td>
<td>&lt;5</td>
</tr>
<tr>
<td>9</td>
<td>MeOH</td>
<td>10</td>
<td>11</td>
<td>62</td>
</tr>
</tbody>
</table>

\(^a\) Unless noted otherwise, the reaction was performed by employing 2 (0.6 mmol), crotonaldehyde (1.2 mmol), and (S)-1b (0.12 mmol) in solvent (1.2 mL) with H\(_2\)O (6.0 mmol) (or without H\(_2\)O) at room temperature for the indicated time.

\(^b\) Amount of water.

\(^c\) Yield of purified product.

\(^d\) Benzoic acid (20 mol%) was added.

\(^e\) A MeOH solution of crotonaldehyde was added over 10 h.

\(^f\) A THF solution of crotonaldehyde was added over 10 h.
The product, which contains three chiral centers, was obtained as a mixture of several diastereomers. Denitration was then investigated. Alcohol 3 was converted into its benzyol ester 4. After optimization of the denitration conditions, it was found that the reaction of 4 with n-Bu3SnH proceeded at 150 °C to afford alcohol 5 in 68% yield with 2.2:1 diastereoselectivity (Scheme 4a).18 To increase the diastereoselectivity, we further optimized the second Michael reaction using an organocatalyst with a different silyl substituent. An improved result was obtained when diphenylmethyisilyl ether (S)-1a12 was employed instead of trimethylsilyl ether (S)-1b to provide, after denitration, the product 6 with 3.7:1 diastereoselectivity (Scheme 4b). As we found that protection of the hydroxy moiety was not necessary during our investigation of the denitration, we converted 3 into alcohol 6 according to the method shown in Scheme 4b. Although the diastereoselectivity was moderate, excellent enantioselectivity was obtained (97% ee). It is noteworthy that the enantioselectivity increased from 90% to 97% (vide infra).

Next, the generality of the asymmetric double Michael reaction was investigated (Table 2). Although the anti-1,3-dimethyl substituent was obtained with moderate diastereoselectivity, excellent enantioselectivity was generated (entry 1). Both the 1,3-syn-dimethyl isomer and the 1,3-syn-methyl ethyl isomer were obtained with excellent diastereoselectivities and enantioselectivities (entries 2 and 3). In the second Michael reaction, cinnamaldehyde was also a suitable Michael acceptor, affording the syn- and anti-isomers with excellent stereoselectivity (entries 4 and 5). Aryl-substituted propenals could also be successfully employed. Notably, both an electron-deficient aryl, such as that with a p-trifluoromethylphenyl substituent, and an electron-rich aryl, such as that with a p-methoxyphenyl substituent, were suitable substrates (entries 6 and 7). Table 2 indicates that the diastereoselectivities are moderate to good and that they depend on the substituents. However, the enantioselectivities of the final products are found to be excellent (>95% ee) for both 1,3-anti- and 1,3-syn-isomers. It should be noted that the enantioselectivity increased in all the cases, although that of the first Michael product 2 was 90%.

The double Michael product could also be transformed into the 1,3-disubstituted-2-oxo derivative through a Nef reaction.20 When anti-7 and syn-7 were treated with NaOMe and dimethyldioxirane (DMDO),1,3-anti- and 1,3-syn-dimethylketones (anti-8 and syn-8), respectively, were obtained in good yields, albeit with a slight decrease of the diastereoselectivity and enantioselectivity (Scheme 5).

Scheme 5 Transformation of Michael products 7 into 1,3-disubstituted-2-oxo derivatives syn-8 and anti-8

Although the enantiomeric excess of the first Michael product was 90%, the double Michael product was formed with an excellent enantioselectivity that was much higher than that of the first Michael reaction. The origin of this enhanced enantioselectivity can be explained as follows (Scheme 6). In the first Michael reaction, 2 and ent-2 were generated in a 95:5 ratio, in which 2 was formed predominantly rather than ent-2. When 2 reacted with crotonaldehyde catalyzed by (S)-1a, in which the (R)-isomer of the newly generated methyl group would be predominantly generated,11 anti-3 was formed predominantly, while the generation of (S)-isomers such as syn-3 and anti-ent-3 would be minor. As ent-2 is generated in a small amount in the first reaction and the generation of anti-ent-3 is also a minor reaction path in the second Michael reaction, the amount of anti-ent-3 would be very little. If the stereoselec-
The ee in the final product 3 is much higher than that of the first Michael product 2.

The present method was applied to the asymmetric synthesis of the side chain of pneumocandin B0 (9) (Figure 2).21 Pneumocandin B0 was isolated from the fermentation broth of the fungus Glarea lozoyensis by Merck & Co. Its fungal-specific mode of action is inhibition of the biosynthesis of β-(1,3)-D-glucan, which is an essential cell wall component of many pathogenic fungi. The stereoselective synthesis of the (10R,12S)-dimethylmyristoyl side chain 10 of this compound through the use of Enders' RAMP method and diastereoselective alkylation of the chiral enolate has previously been reported.21c

**Table 2  The Two-Pot Synthesis of 1,3-Disubstituted Alkanols**

<table>
<thead>
<tr>
<th>Entry</th>
<th>Product</th>
<th>Cat.</th>
<th>Michael reaction yield (%)</th>
<th>Denitration yield (%)</th>
<th>anti/syn</th>
<th>ee (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>S</td>
<td>60</td>
<td>nd</td>
<td>49</td>
<td>3.7:1</td>
<td>97</td>
</tr>
<tr>
<td>2</td>
<td>R</td>
<td>63</td>
<td>nd</td>
<td>51</td>
<td>1:10</td>
<td>98</td>
</tr>
<tr>
<td>3*</td>
<td>R</td>
<td>60</td>
<td>nd</td>
<td>44</td>
<td>1:20</td>
<td>97</td>
</tr>
<tr>
<td>4</td>
<td>S</td>
<td>80</td>
<td>63:28:7:2</td>
<td>54</td>
<td>13:1</td>
<td>98</td>
</tr>
<tr>
<td>5</td>
<td>R</td>
<td>78</td>
<td>59:30:6:5</td>
<td>48</td>
<td>1:15</td>
<td>96</td>
</tr>
<tr>
<td>6</td>
<td>S</td>
<td>91</td>
<td>53:42:5:0</td>
<td>62</td>
<td>&gt;20:1</td>
<td>&gt;99</td>
</tr>
<tr>
<td>7</td>
<td>S</td>
<td>80</td>
<td>62:26:9:3</td>
<td>65</td>
<td>5.9:1</td>
<td>99</td>
</tr>
</tbody>
</table>

*First step (Michael reaction): Unless noted otherwise, the reactions were performed by employing 2 (0.6 mmol), α,β-unsaturated aldehyde (1.2 mmol), (S)-1a or (R)-1a (0.12 mmol), and H2O (6.0 mmol) in MeOH (1.2 mL) at room temperature via slow addition of the aldehyde over 20 h and further stirring of the reaction mixture for 1 h. Second step (denitration reaction): Unless noted otherwise, the reactions were performed by employing the Michael adduct (0.4 mmol), n-Bu3SnH (2.0 mmol), AIBN (0.32 mmol), and 1,3,5-trimethoxybenzene (14.0 mmol) at 250 °C for 5 min.

*b Yield of purified product.

c dr = diastereomer ratio in the Michael reaction determined by 1H NMR spectroscopy; nd = not determined.

d Diastereomer ratio and enantiomeric excess were determined by HPLC analysis on a chiral column.

e Slow addition over 40 h during the Michael reaction.
Our synthesis of the side chain 10 started with the Michael reaction of nitromethane and crotonaldehyde catalyzed by diphenylprolinol silyl ether (S)-1a. Subsequent acetalization provided 2 in 94% yield with 90% ee. The second Michael reaction with crotonaldehyde proceeded in the presence of (R)-1a, followed by treatment with NaBH₄ to afford alcohol syn-3 in 63% yield. The enantioselectivity of syn-3 is 98%, which was determined after denitration (see Table 2, entry 2). Alcohol syn-3 was converted into haloalkane 11 in 69% yield by reaction with PPh₃ and I₂. Dehalogenation and denitration occurred in the same pot by treatment with n-Bu₃SnH and AIBN at 150 °C to afford acetal 12 in 73% yield. Treatment of acetal 12 with aqueous HCl gave aldehyde 13, which was used in the next step without purification. The Julia–Kocienski reaction with 14 proceeded smoothly to afford (E)-alkene 15 in 56% yield over two steps. Hydrogenation followed by hydrolysis using aqueous NaOH afforded the side chain of pneumocandin B₀ 10 in 72% yield over two steps (Scheme 7). The physical properties of synthetic 10 were identical in all respects to the reported data.¹²d

**Scheme 6** The reason for the higher ee of the second Michael product

**Scheme 7** Synthesis of the side chain of pneumocandin B₀
In conclusion, we have developed an efficient method for the synthesis of chiral 1,3-dimethyl units through a double Michael reaction of an aldehyde and nitroalkane catalyzed by a diphenylprolinol silyl ether. There are several noteworthy features of this reaction. Either 1,3-syn- or 1,3-anti-dimethyl units can be selectively synthesized depending on the appropriate choice of enantiomer of the diphenylprolinol silyl ether catalyst. The excellent optical purity of the double Michael product was much higher than that of the first Michael reaction because of the ‘meso-trick’. In addition to the 1,3-dimethyl unit, both 1,3-methyl alkyl and 1,3-methyl aryl units can be prepared. Finally, the side chain of pneumocandin B₉ was enantioselectively synthesized by using the present method as a key step.

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**Supporting Information**

Supporting information for this article is available online at https://doi.org/10.1055/s-0037-1611644.

**References and Notes**

(16) In the reactions of entries 2–8 in Table 1, nitroalkane 2 was recovered in good yield (>90%).
(17) The diastereoselectivity and enantioselectivity of 3 (Table 1, entry 9) were not determined. The dr after denitration is 2:2:1, see Scheme 4.
(19) The enantioselectivity of compound 5 is determined according to the scheme below.
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