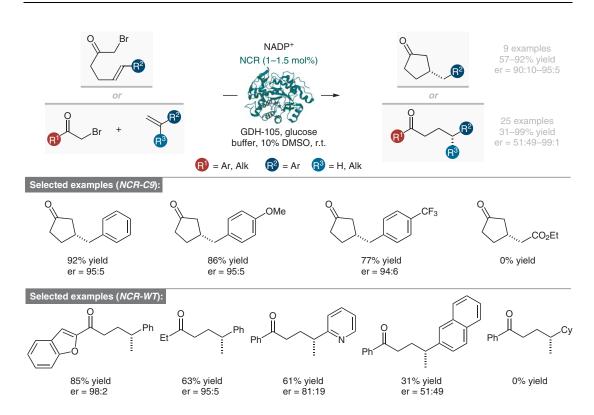
H. FU, H. LAM, M. A. EMMANUEL, J. H. KIM, B. A. SANDOVAL, T. K. HYSTER* (PRINCETON UNIVERSITY AND CORNELL UNIVERSITY, ITHACA, USA) Ground-State Electron Transfer as an Initiation Mechanism for Biocatalytic C-C Bond Forming Reactions *J. Am. Chem. Soc.* **2021**, *143*, 9622–9629, DOI: 10.1021/jacs.1c04334.

Enzymatic Intra- and Intermolecular Hydroalkylations of Alkenes through Ground-State Electron Transfer



Significance: Hyster and co-workers report intraand intermolecular reductive hydroalkylations of aromatic olefins to form cyclopentanones or linear ketones in excellent yields and enantioselectivities. Quadruply mutated or wild-type nicotinamidedependent cyclohexanone reductase (NCR), respectively, serve as efficient biocatalysts. Starting from α-bromo ketones, ground-state electron transfer from a flavinmononucleotide generates a ketyl radical that, through mesolytic C-Br bond cleavage, generates the reactive α -ketonyl radical. Notably, whereas the stereocenter in the cyclization reaction is set in the C-C bond-forming step, the enantiocontrol in intermolecular reactions originates from a stereoselective radical-terminating hydrogenatom transfer.

Comment: Flavin-dependent ene-reductases (EREDs) have been previously applied in photoenzymatic settings (see, for example: K. F. Biegasiewicz et al. Science 2019, 364, 1166). Whereas those reactions rely on the photoexcitation of a charge-transfer complex between enzyme, cofactor, and substrates, the analogous ground-state electron transfer had not previously been utilized as an initiation mechanism in C-C bond-forming reactions. The authors therefore selected α -bromo ketones as substrates due to their relatively high reduction potential, rendering ground-state reactivity kinetically feasible. Although the present method is an impressive example of enantiocontrol over real radical intermediates, the extension to less-stabilized nonaromatic substrates represents a considerable challenge for future research.

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