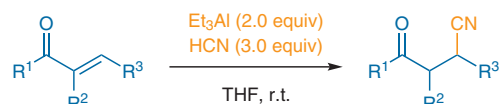
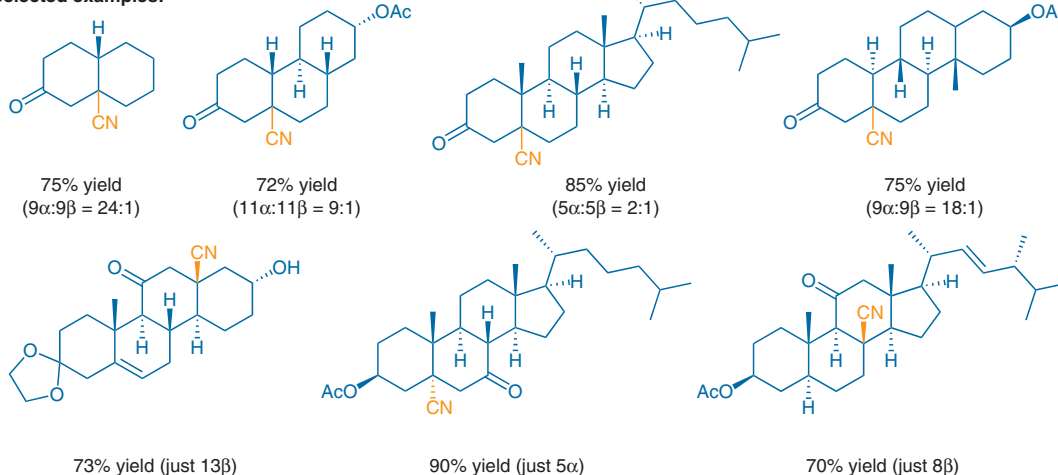


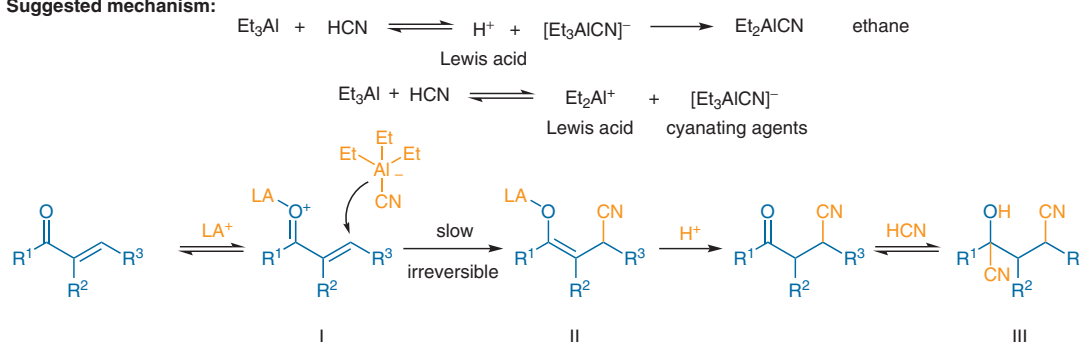
## Nagata Hydrocyanation: Synthesis of $\beta$ -Cyanoketones



### Selected examples:



### Suggested mechanism:



**Significance:** In 1962 Nagata et al. reported a method for the hydrocyanation of demanding  $\alpha,\beta$ -unsaturated ketones for the synthesis of  $\beta$ -cyano-ketones. This reaction employed a mixture of HCN and triethylaluminium in THF at room temperature as a new reagent that performed better than alkali cyanides (commonly used reagents at the time) in terms of yield and diastereoselectivity. This method has become known as Method A for hydrocyanation of unsaturated ketones.

**Comment:** Interesting mechanistic observations were made. It was suggested that polar solvent (THF) leads to a cyanotriethylaluminum, which was deemed to be the active cyanating reagent.  $\text{Et}_2\text{Al}^+$  or a simple proton were proposed as Lewis acids. A slow and irreversible addition of cyanide ion gave **II** as an intermediate, which was quenched with HCN present in situ to give the expected product. Cyanohydrine **III** was also observed. The reaction proceeded with gas evolution and formation of diethylaluminumcyanide. This compound in inert solvent would become known as Method B and is the only commercially available reagent for Nagata hydrocyanation.