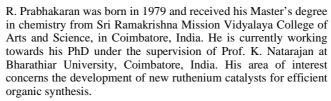
SYNLETT Spotlight 98

This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

Shvo's Diruthenium Complex [(η⁵-C₄Ph₄COHOCC₄Ph₄-η⁵) (μ-H)(CO)₄Ru₂]

Compiled by R. Prabhakaran



E-mail: rpnchemist@rediffmail.com



Introduction

After robust diruthenium complex 1 was reported by Shvo,¹ it found enormous utility in organic synthesis as a versatile catalyst, and it is now called Shvo complex. The main reason for its efficient catalytic activity is that it dissociates into 2 and 3 in the reaction mixture under thermal conditions. The 16-electron species 2 and 18-electron complex 3 play vital roles in the catalytic cycle.² Recently, a modified procedure for the synthesis of Shvo com-

plex was reported by Bäckvall and co-workers.³ Shvo complex serves as an effective catalyst in many organic transformations, including the reduction of aldehydes and ketones to alcohols,⁴ bimolecular disproportionation reaction of aldehydes to esters,⁵ isomerization of allylic alcohols⁶ and oxidation of alcohols.⁷ In addition, Shvo complex as a racemization catalyst in dynamic kinetic resolution (DKR) is well precedented.⁸

Preparation

Shvo complex can be synthesized from $Ru(CO)_3(\eta^4-Ph_4C_4CO)$, which is made from $Ru_3(CO)_{12}$ and tetraphenylcyclopentadienone.³

Abstracts

(A) Shvo complex catalyzed the in situ racemization of secondary alcohols, which then underwent enzyme-catalyzed DKR, in the preparation of enantiomerically pure acetates.³ The same catalyst has also been used for the racemization of amines and then applied to the synthesis of enantiomerically pure amines.⁹

SPOTLIGHT 2049

(B) The combination of enzyme-catalyzed DKR and an aldol reaction provided an access to β -hydroxy ester derivatives with high enantiomeric purity in a one-pot procedure using Shvo complex as catalyst for racemization. ¹⁰

(C) Oxidation of 5-unsaturated 3β -hydroxy steroids to the corresponding 4-en-3-one derivatives can be achieved by acetone at reflux in the presence of a Shvo catalyst. The reaction proceeds via a ruthenium-catalyzed dehydrogenation and subsequent hydrogen transfer to acetone with concomitant double bond migration.¹¹

(D) Shvo complex is able to catalyze the transfer dehydrogenation of amines to imines under mild conditions using 2,6-dimethoxy-benzoquinone as oxidant.¹²

(E) Complex 1 serves as a catalyst precursor for the hydrogenation of alkynes. ¹³

(F) Efficient aerobic oxidation of alcohols was developed via biomimetic catalytic system. Complex 1 dehydrogenates the alcohol and the hydrogens abstracted are transferred to an electron-rich quinone. The hydroquinone thus formed is reoxidized by air with the aid of Co-Salen complex.^{2a}

References

- Shvo, Y.; Czarkie, D.; Rahamim, Y. J. Am. Chem. Soc. 1986, 108, 7400.
- (2) (a) Csjernvik, G.; Éll, A. H.; Fadini, L.; Pugin, B.; Bäckvall, J.-E. J. Org. Chem. 2002, 67, 1657. (b) Casey, C. P.; Singer, S. W.; Powell, D. R.; Hayashi, R. K.; Kavana, M. J. Am. Chem. Soc. 2001, 123, 1090.
- (3) Persson, B. A.; Larsson, A. L. E.; Ray, M. L.; Bäckvall, J.-E. J. Am. Chem. Soc. 1999, 121, 1645.
- (4) Menashe, N.; Salant, E.; Shvo, Y. J. Organomet. Chem. 1996, 514, 97.
- (5) Menashe, N.; Shvo, Y. Organometallics 1991, 10, 3885.
- (6) Bäckvall, J.-E.; Andreasson, U. Tetrahedron Lett. 1993, 34, 5459.
- (7) Jung, H. M.; Choi, J. H.; Lee, S. O.; Kim, Y. H.; Park, J. H.; Park, J. Organometallics 2002, 21, 5674.

- (8) (a) Pàmies, O.; Bäckvall, J.-E. Chem. Rev. 2003, 103, 3247.
 (b) Kim, M.-J.; Ahn, Y.; Park, J. Curr. Opin. Biotechnol. 2002, 13, 578. (c) Huerta, F. F.; Minidis, A. B. E.; Bäckvall, J.-E. Chem. Soc. Rev. 2001, 30, 321.
- (9) Pàmies, O.; Éll, A. H.; Samec, J. S. M.; Hermanns, N.; Bäckvall, J.-E. *Tetrahedron Lett.* **2002**, *43*, 4699.
- (10) Huerta, F. F.; Bäckvall, J.-E. Org. Lett. 2001, 3, 1209.
- (11) Almeida, M. L. S.; Koovsk, P.; Bäckvall, J.-E. J. Org. Chem. 1996, 61, 6587.
- (12) Éll, A. H.; Samec, J. S. M.; Brasse, C.; Bäckvall, J.-E. *Chem. Commun.* **2002**, 1144.
- (13) Shvo, Y.; Goldberg, I.; Czerkie, D.; Reshef, D.; Stein, Z. Organometallics 1997, 16, 133.