SYNLETT Spotlight 213

This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

Copper(I) Iodide

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Introduction

The first recorded attempt at making an organocopper was Buckton's use of diethylzinc on CuCl in the 1850s.¹ From that time, copper halides and consequently CuI have found widespread use in synthetic organic chemistry. Indeed, copper iodide, also called cuprous iodide, acts as a useful precursor of organocopper compounds which due to their unique chemoselectivity and reactivity occupy a special place in organic synthesis.^{2–4} Furthermore, the effectiveness of CuI as catalyst or co-catalyst for crosscoupling reactions has been widely reported in the literature.^{5,6}

The utilisation of CuI also encompasses a broad range of chemical transformations like construction of heterocycles,^{7–9} iodination reations,¹⁰ click chemistry¹¹ or multicomponent coupling reactions.^{12–14}

CuI is commercially available as an off-white solid but samples with time are often tan due to impurities. A dissolution–precipitation process with water in the presence of NaI or KI is used to purify CuI.^{15,16} The colourless CuI is then stored under argon and protected from light to avoid decomposition.

Abstracts

(A) Preparation of Chiral Tributylstannyl- α -Amino Alcohols: Ring-opening of 2-tributylstannyloxazolidines by organocopper reagents in the presence of BF₃·OEt₂ affords the corresponding tributylstannyl- α -amino alcohols in moderate to excellent yields.¹⁷ This reaction proceeds in a diastereoselective fashion in favour of the *anti* isomer (dr close to 85:15).

(B) Modified Stille Cross-Coupling:

In 1990, Liebeskind et al. highlighted the beneficial effect of catalytic copper iodide on the Stille reaction.¹⁸ 3-Stannyl-cyclobutenediones and stannylcyclobutenedione monoacetals undergo efficient palladium-catalysed cross-coupling reactions with organic iodides.

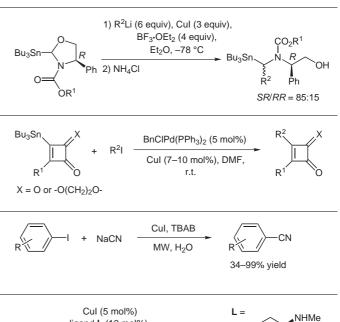
(C) Preparation of Aryl Nitriles:

A rapid cyanation of aryl iodides under microwave activation occurs by using inexpensive NaCN as the cyanide source, and CuI as an additive. The reaction is performed in water, and tetrabutyl-ammonium bromide is used as a phase-transfer catalyst.¹⁹

(D) Iodination Reaction:

In the presence of a catalyst system comprising CuI and a 1,2- or 1,3-diamine ligand, aryl or heteroaryl bromides are converted into the corresponding iodides in excellent yields (93–100%).¹⁰ This method can also be applied to halogen exchange in vinyl bromides.

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Arl

′NHMe

ligand L (10 mol%)

Nal (2 equiv)

dioxane, 110 °C, 22-24 h

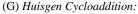
ArBr

(E) Vinvlation Reaction:

CuI, in combination with tetradendate ligand L, represents a remarkable catalyst system for vinylation reactions.²⁰ Under this set of conditions, both azoles and substituted phenols can be coupled with β -bromostyrene. While this protocol is effective for the vinylation of (E)- β -bromostyrene, the Z-isomer leads mainly to byproducts due to its lower reactivity.

(F) Construction of Substituted Pyrroles:

A general and practical method for the construction of 2-monosubstituted and 2,5-disubstituted pyrroles via a CuI-assisted cycloisomerisation of readily available alkynyl imines has been recently reported.7 This approach has also been extended to the synthesis of fused aromatic heterocycles containing a pyrrole ring.



Fixation of CuI on Amberlyst A-21 furnishes an efficient heterogeneous catalytic system for the Huisgen's [3+2] cycloaddition between azides and alkynes. Furthermore, this polymer-supported catalyst can be reused for several cycles without decreased reaction yields in triazoles.21

(H) Synthesis of Propargylamines:

Propargylamines can be synthesised through a three-component coupling reaction of an aldehyde, an alkyne and an amine under ultrasound conditions.14 Addition of CuI, as a catalyst, dramatically increases the yields (up to 98% yield).

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